

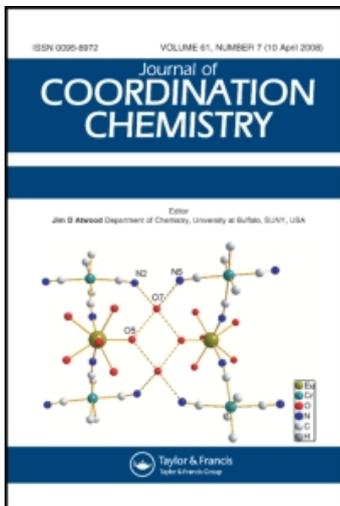
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Journal of Coordination Chemistry

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713455674>

COPPER(II) COORDINATION COMPOUNDS: CLASSIFICATION AND ANALYSIS OF CRYSTALLOGRAPHIC AND STRUCTURAL DATA II. MONONUCLEAR-, HEXA-, HEPTA-AND OCTACOORDINATE COMPOUNDS

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To cite this Article Melník, Milan , Kabešová, Mária , Macášková, Ľubov and Holloway, Clive E.(1998) 'COPPER(II) COORDINATION COMPOUNDS: CLASSIFICATION AND ANALYSIS OF CRYSTALLOGRAPHIC AND STRUCTURAL DATA II. MONONUCLEAR-, HEXA-, HEPTA-AND OCTACOORDINATE COMPOUNDS', Journal of Coordination Chemistry, 45: 1, 31 – 145

To link to this Article: DOI: 10.1080/00958979808027143

URL: <http://dx.doi.org/10.1080/00958979808027143>

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Review

COPPER(II) COORDINATION COMPOUNDS: CLASSIFICATION AND ANALYSIS OF CRYSTALLOGRAPHIC AND STRUCTURAL DATA II. MONONUCLEAR-, HEXA-, HEPTA- AND OCTACOORDINATE COMPOUNDS

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(Received 7 August 1996; In final form 3 March 1997)

This review summarizes the data for over five hundred and fifty six-, seven- and eight-coordinate Cu(II) coordination compounds. The six-coordinate derivatives are mostly distorted tetragonal (elongation along a single C_4 axis). The seven-coordinate derivatives are only pentagonal-bipyramidal with different degrees of distortion and the eight-coordinate complexes are dodecahedral. The most common ligands are O- and N-donors. There is a trend for the Cu–L distance to increase with covalent radius of the coordinated atom and also increasing coordination number. In general, Cu(II) compounds “prefer” a monoclinic class and are blue or green. Several relationships were found and discussed. The data are compared and discussed with those of four- and five-coordinate Cu(II) compounds.

Keywords: Review; Cu(II); crystallography; structures

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ABBREVIATIONS

aad	8-azaadenine
abpt	3,5-bis(pyridin-2-yl)-4-amino-1,2,4-triazole
abst	2-(aminobenzensulfonamide)-5-ethyl-1,3,4-thiadiazole
ac	acetate
ac- α -ala	N-acetyl- α -alaninate
acatr	3-acetylamino-1,2,4-triazole
acgl	N-acetylglycine
ac-DL-pgly	N-acetyl-DL-phenylglycinate
acsal	acetylsalicylate
act-DL	N-acetyl-DL-tryptophanate
acth	2-[(5-amino-4-carboxamidinium)[1,2,3]triazole]
actr	5-amino-4-carboxamide[1-3]triazole
acts	5-acetamido-1,3,4-thiadiazole-2-sulfonamide
adip	adipate
aenol	2[(3-aminopropyl)amino]ethanolate
aepy	2-(2-aminoethyl)pyridine
ahb	2-amino-1-heptybenzimidazole
ahmi	2-azohexamethyleneimine
ainc	amidoisonicotinate
6-am	6-aminohexanoic acid
amb	2-amino-1-methylbenzimidazole
amcyclam	3,10-diammonio-3,10-dimethyl-1,4,8,11-tetra- azacyclotetradecane
amdza	1,3bis(2-aminoethyl)5-ammonio 5-methyl 1,3-diazacyclo- hexane
ami	11-aminoundecane
amnur	6-amino-1,3-dimethyl-5-nitrosouracil
amOac	aminoxyacetate

amp	DL- α -amino- β -methylaminopropionic acid
ampr	2-amino-2-methyl-1-propanole
ampy	2-aminoethylpyridine
4-amsal	4-aminosalicylate
9-aneN ₂ S	1-thia-4,7-diazacyclononane
9-aneN ₂ Sac	dimethyl 1-thia-4,7-diaza-4,7-cyclononanediacetate
10-aneN ₂ S ₂	1,7-bis(5-methylimidazol-4-yl)-2,6-dithiaheptane
9-aneN ₃	1,4,7-triazacyclononane
12-aneN ₄	N,N'-dimethyl-2,11-diaza[3.3](2,6)pyridinophane
12-aneN ₄ ac	12-membered tetraazamacrocyclic-N,N',N'',N'''-tetraacetic acid
12-aneO ₄	1,4,7,10-tetraoxacyclododecane
14-aneS ₄	1,4,8,11-tetrathiacyclotetradecane
14-aneS ₂ N ₂	6-methyl-6-nitro-1,11-dithia-4,8-diazacyclotetradecane
15-aneN ₄	1,4,8,12-tetraazacyclopentadecane
15-aneS ₄	1,4,8,12-tetrathiacyclopentadecane
16-aneS ₄	1,5,9,13-tetrathiacyclohexadecane
17-aneO ₃ N ₂	pyridynyl-derivat macrocycle
17-aneO ₃ N ₂	3,22-dioxa-11,14,28-triazatetra-cyclo[22.3.1.0 ^{4,9} 0 ^{16,21}]octacosal(28),4,6,8,16,18,20,24,26 nonaene
18-aneO ₄ N ₂	1.7.10.16 tetraoxa-4.13-diaza-cyclotetradecane
18-aneN ₂ S ₂	3,3,7,7,11,11,15,15-octamethyl-1,9-dithia-5,13-diazacyclohexadecane
18-aneS ₄ N ₂	1,4,10,13-tetrathia-7,16-diazacyclooctadecane
apy	antipyrine
asp	L-asparaginate
athal	N-(2-aminoethyl)thiophen-2-aldimine
atr	atratone
azcm	catena- μ -(4-aminoimidazole-5-carboxamidoxime)
bcmgly	N,N-bis(carboxymethyl)glycinate
bet	betaine
bheg	N,N-bis(2-hydroxyethyl)glycinate
bhdhx	1,6-bis(4(5)-imidazolyl)-2,5-dithiahexane
bim	benzimidazole
bimm	benzimidazol-2-yl-methanol
bimpa	N,N-bis(benzimidazol-2-yl-methyl)isopropylamine
bip	2,2'-bipiperidine
bipyam	2,2'-N,N'-bispyridylamine

biu	biuret
1,3-bn	1,3-diaminobutane
bpc	N-2-pyridinylcarbonyl-2-pyridinecarboximidate
bpdz	1,7-bis(2-pyridyl)-2,6-diazaheptane
bpm	2,2'-bipyrimidine
bpy	2,2'-bipyridine
bpykh	2,2'-N,N'-bipyridyl ketone hydrate
bpyO ₂	2,2'-bipyridine-N,N'-dioxide
btac	(benzylthio)acetate
bz	benzoate
bzibza	N,N-bis(2-benzimidazolylmethyl)benzylamine
bzimox	1,6-bis(benzimidazol-2-yl)-2,5-dioxahexane
bzph	2-aminobenzophenone
bzpy	2-benzoylpyridine
bz-pyrpy	2,6-bis{(2S)-2-[(benzyloxy)methyl]pyrrolidin-yl}methyl}pyridine
bzs	benzenesulphonate
c	cubic
camphs	D-camphor-10-sulphonate
cas	calix[4]arene sulphonate
cbq	1-carbamoylguanidinate
C ₅ (CO ₂ Me) ₅	pentakis(methoxycarbonyl)cyclopentadiene
ceten	N,N'-bis(β-carbamoylethyl)trimethylenediamine
C ₂ H ₆ O ₂	1,2-ethanediol
C ₂ H ₁₀ N ₂	ethylenediamonium
C ₆ H ₅ O ₇	citrate
C ₇ H ₁₈ N ₂	N ¹ -isopropyl-2-methyl-1,2-propanediamine
C ₈ H ₈ O ₅	7-oxabicyclo[2.2.1]heptane-2,3-dicarboxylate
C ₈ H ₁₁ NO ₃	2-methyl-oxy-4,5-di(oxyethyl)pyrrolidine
C ₈ H ₁₆ N ₄ O ₂	N,N'-bis(β-carbamoylethyl)ethylenediamine
C ₂₀ H ₂₈ N ₄ S	quinquedentate macrocyclic ligand
C ₂₂ H ₃₄ N ₆	3,6,12,15-tetramethyl-18{[N(2,6)-pyridino,N-1.2.1] 2-coronand-6}
Cl ₂ aq	8-amino-5,7-dichloroquinoline
2-Clbquam	2-chlorobiquanide
4-Clbzs	4-chlorobenzenesulphonate
2,4,6-Cl ₃ ph	2,4,6-trichlorophenolate
4-ClphOac	4-chlorophenoxyacetate
Clqo	4-chloro-1-benzoquinone 2-oximate
cnact	chloronitroacetamidate

1-CN _{gud}	1-cyanoguanidine
croc	croconate
15-crown-5	macrocyclic polyethers
c-sar ₂	cyclosarcosylsarcosine
cyclam	1,4,8,11-tetraazacyclotetradecane
cyclamac	1,4,8,11-tetraazacyclotetradecane-N,N',N'',N'''-tetraacetate
cyclox	3-methyl-3-nitro-1,5,9,13-tetraazacyclohexadecane
cyt	cytosine
dacac	1,4,10,13-tetraoxa-7,16-diazacyclooctadecane-7-acetate
daco	1,5-diazacyclooctane
daes	1,7-diaza-4-thiaheptane
dafone	4,5-diazafluoren-9-one
damp _r	diamine-2,3-propionique(N ² O)acid
dapsc	2,6-diacetylpyridine bis semicarbazone
dcam	dicyandiamide
debd	1,6-bis(3,5-dimethylpyrazol-1-yl)-2,5-dimethyl-2,5-diaza-hexane
degly	diethanolglycine
dempz	3,5-diethyl-4-methylpyrazole
dhmp	2,3-dihydroxy-2-methylpropanoate
dien	diethylenetriamine
dim	diimidazole
dmact	N,N-dimethyl-acetamide
dmco	2,5-dimethyloxazolidine-4-carboxylate
dmen	N,N-dimethyl-1,2-ethanediamine
dmf	dimethylformamide
dmoc	2,5-dimetyloxazolidine-4-carboxylate
dmpr	1,3-diamino-2-propanole
2,6-dnph	2,6-dinitrophenolate
dsc _o	3-hydroxy-1,5-dithiacyclooctane
dptp	1-(3,5-dimethyl-1-pyrazolyl)-3-thiapentane
dtpa	trihydrogen diethylenetriaminepentaacetate
dtun	1,11-diamino-3,6,9-trithiaundecane
dund	6-hydroxy-4,8-diazaundecanediamine
dzco	1,4,10,13-tetraoxa-7,16-diazacyclooctadecane-7,16-N,N'-diacetate
eac	ethylenediamine-tetraacetate
ebtb	N,N,N',N'-tetrakis(2'-benzimidazolymethyl)1,2-ethane diamine
ecgly	ethylenebis(N-methylcarbamyglycinate)

ecgm	N,N'-1,2-ethanediylbis[N-(carboxymethyl)glycyl-L-methionine]ethylesterate
eiphim	3,3'-ethylenedi-iminobis(o-phenyleneimino-methylidene)bis[pentane-2,dionate(2-)]
en	ethylenediamine
enphgly	ethylenebis(o-hydroxyphenyl)glycine
esbim	2-(ethylthiomethyl)benzimidazole
esop	2-(ethylthio)-4-oxopteridine
etam	ethanolamine
etapy	2-(2-diethylammonioethyl)pyridine
5,5'-Et ₂ barb	5,5'-diethylbarbiturate
Eten	N-ethylenediamine
EtenOH	N-(2-hydroxyethyl)ethylenediamine
Et ₂ en	N,N'-diethylenediamine
ethc	ethyl-1,5,9,13-tetraazabicyclo[11.2.2]hepta-decane-7-carboxylate
1-Etim	1-ethylimidazole
EtMecyclam	5,7,12,14-tetraethyl-7,14-dimethyl-1,4,8,11-tetraazacyclotetradecane
EtOac	ethoxyacetate
Etsac	ethylthioacetate
Et ₂ sp	diethylthiophosphate
evda	D,L-ethylvalinate-N,N-diacetate
F ₆ acac	1,1,1,6,6,6-hexafluoro-2,4-pentanedionate(hexafluoroacetylacetonate)
fm	formiate
fmpn	4-formyl-2-methoxyphenolate
4-FphOac	4-fluorophenoxyacetate
fpsem	2-formylpyridine thiosemicarbazone
F ₅ Sph	pentafluorothiophenolate
ghgly	glycyl-L-histidylglycinate
ggly	glycylglycinate
gly	glycine
hbbzim	2- α -hydroxybenzylbenzimidazole
HB(3,5-Me ₂ pz) ₃	hydrotris(3,5-dimethyl-1-pyrazolyl)borate
HB(pz) ₃	hydrotris-(pyrazol-1-yl)borate
hd	3,4-hexanedione dioximate
hdmpz	1-(2-hydroxyethyl)-3,5-dimethylpyrazole
heac	dihydrogen ethylenediaminetetraacetate

hemq	9-[(2-hydroxyethoxy)methyl]quanine
himp	2-hydroxyethyliminopyruvate
hist	histidine
hmnc	N-(hydroxymethyl)nicotinamid
hpca	N-(hydroxymethyl)-3-pyridinecarboxamide
Hpth	hydrogen o-phthalate
hip	hippurate
Hpydca	hydrogen-pyridine-2,6-dicarboxylate
ibpca	N-(2-(4-imidazolyl)ethyl)-2-methyl-5-bromopyrimidine-4-carboxamide
idac	iminodiacetate
3-Ihpur	3-iodohippurate
im	imidazole
imdth	1,6-bis(5-methyl-4-imidazolyl)-2,5-dithiahexane
5'-imp	inosine-5'-monophosphate
inc	isonicotinamide
inco	isonicotinate N-oxide
ipcp	tetraisopropylmethylenediphosphonate
iprthacq	isopropylthioacetate isoquinoline
ipm-1,2-pn	N ¹ -isopropyl-2-methyl-1,2-propanediamine
iqu	isoquinoline
lac	lactate
lc	lignocaine
lys	L-lysine
m	monoclinic
mact	N-methylacetamide
mal	malonate
mbpy	2,6-bis(1'-methylbenzimidazol-2'-yl)pyridine
mbpym	1,3-dimethyl-2,4-dioxo-6-hydroxy-5-nitroso-1,2,3,4-tetrahydropyrimidine
mceam	β -methylmercaptoethylamine
mcgly	N,N'-methylethylene bis(N-methylcarbamyglycinate)
mdac	iminodiacetamide
9-Meade	9-methyladenine
Me ₂ amet	2-dimethylaminoethanole
4-Mebzcs	4-methylbenzenecyanamidosulfonate
2-Mebzim	2-methylbenzimidazole
Mecyclam	2,5,5,7,9,12,14-octamethyl-1,4,8,11-tetra-azacyclotetradecane

Me ₂ CO	acetone
4,4'5,5'-Me ₄ dim	4,4'5,5'-tetramethyldiimidazole
9-Mehpx	9-methylhypoxanthin
Meial	10-methylisoalloxazine
Meim	N-methylimidazole
Meims	bis(1-methylimidazol-2-yl)sulfide
Meimthb	4-(5-methyl-4-imidazolyl)-1-amino-3-thiabutane
Meimthp	1,3-bis(5-methyl-4-imidazolyl)-2-thiapropane
MeNO ₂	nitromethane
mep	mepirizole
Me ₂ pam	3,3'-(2,2-dimethylpropylene)diiminodipropionamide
Me-picam	methyl-2-picoylamine
9-Meop	9-methyl-6-oxypurine
2Me-1,2-pn	2-methyl-1,2-propanediamine
Me ₂ -1,3-pn	2,2-dimethylpropane-1,3-diamine
Me ₄ psim	2-(2-pyridyl)-4,4,5,5-tetramethyl-4,5-dihydro-1H-imidazolyl-1-oxy
4-MepyNO	4-methylpyridine 1-oxide
Me-pyrpy	2,6-bis{[(2S)-2-(methoxymethyl)pyrrolidin-1-yl]methyl}pyridine
4-Mepz	4-methyl-1H-pyrazole
8-Mesthp	8-methylthiotheophyllinate
9-Mequa	9-methylquanine
3,5-Me ₂ guapz	3,5-dimethyl-1-guanylpyrazole
5,7-Me ₂ tpym	5,7-dimethyl-[1,2,4]triazolo[1,5-a]pyrimidine
metri	methyltribenzo[b,f,j][1,5,9]triazacyclododecine
mic	miconazole
mimdth	1,6-bis(5-methyl-4-imidazolyl)-2,5-dithiahexane
mncyclam	C-meso-5,5,7,12,12,14-hexamethyl-1,4,8,11-tetraazacyclotetradecane or C-meso-5,5,7,12,12,14-hexamethyl-1,4,8,11-tetraazacyclotetradecane
mntcp	10-methyl-10-nitro-1,4,8,12-tetraazacyclopentadecane
mnim	2-methyl-5-nitroimidazole
mocpz	3,5-dimethoxycarbonylpyrazole
monp	2-methoxy-4-nitrophenolate
mor	morpholine
mpo	2-(methylaminomethyl)pyridine 1-oxide
mtpo	5-methyl[1,2,4]triazolo[1,5-a]pyrimidin-7-olate

mug	mugineic acid
nbta	1,1,1-trifluoro-4-(4'-nitrophenyl)-buta-2,4-dionate
nc	nicotinamide
ndap	N,N'-bis(2-pyridylmethylene)-1,3-diamino-2-methyl-2-nitropropane
nn	1-nitroso-2-naphtole
6-NO ₂ cyclam	6-(p-nitrobenzyl)-1,4,8,11-tetraazacyclo-tetradecane-1,4,8,11-tetraacetate
4-NO ₂ pyNO	4-nitropyridine N-oxide
npac	nitrilodi(β -propionamide)monoacetate
nphyr	1,8-naphthyridine
nphyrO	1,8-naphthyridine-N-monoxide
N(py) ₃	2,2',2''-tripyridylamine
odac	oxydiacetate
Opy	2,2'-(1,3,4-oxadiazole-2,5-diyl)dipyridine
OH-1,3-pn	2-hydroxy-1,3-propanediamine
2-OH-2-Mepr	2-hydroxy-2-methyl-propionate
OH-pyry	2,6-bis{[(2S)-2-(hydroxymethyl)pyrrolidin-1-yl]methyl}pyridine
ompa	octamethylpyrophosphoramide
opai	1-oxa-1,6,8,12,15-penta-aza-tricyclo[13,3,1.1 ^{8,12}]icosane
opca	N-oxymethylamide-pyridine-3-carbon acid (nikodine)
2-Opy	2-pyridone
or	orthorhombic
orn	σ -N-hydro-L-ornithinate
ottd	1-oxa-4,8,12-triazacyclotetradecane-4,12-diacetate
ox	oxalate
papt	1,4-di(2'-pyridyl)aminophthalazine
pbpc	1-oxo-1,2-diphenyl-3,3,5-tricarbbutoxy-1,2-2-diphosphacyclopentanone
pcam	bis(2-pyridylcarbonyl)amide
pcb	1-pyrroline-2-carboxylate (L-ornithine)
pcp	octamethylenediphosphonic diamine
pdto	1,8-diamino-3,6-dithiaoctane
peme	1,1-di(2-pyridyl)ethylmethyl ether
Ph	phenyl

phen	1,10-phenanthroline
phenO	1,10-phenanthroline-N-monoxide
phepy	2,6-bis[1-(phenylimino)ethyl]pyridine
Phnit	2-phenyl-4,4,5,5-tetramethylimidazoline-1-oxyl-3-oxide
Phimthp	1,3-bis(5-phenyl-2-imidazolyl)-2-thiopropene
phOac	phenoxyacetate
Ph ₂ P(O)CH ₂ P(O)Ph ₂	tetraphenylmethylenediphosphonium dioxide
picl	4-amino-3,5,6-trichloropyridine-2-carboxylate
picNO	picolin-N-oxide
pipe	2-piperidinecarboxylate
2-pira	2-pyridylaniline
pith	pyridyl, imine, thioether donor
pmca	bis(2-pyrimidylcarbonyl)aminatate
pmtrz	2,4-bis(n-propylamino)-6-methoxy-1,3,5-triazine
1,2-pn	1,2-propanediamine
1,3-pn	1,3-diaminopropane
poet	bis[(diphenylphosphinyl)methyl]ethyl phosphinate
pppa	N-(2-pyridyl)-3-phenyl-2-propeneamide
pr	propionate
preim	4-(n-propylmercaptomethyl)imidazole
prepy	2-(n-propylmercaptomethyl)pyridine
prim	2-pyrimidinone
prol	prolinate
prometone	2,4-bis(isopropylamino)-6-methoxy-1,3,5-triazine
prtac	isopropylthioacetate
psac	phenylsulphinylacetate
psalh	N-picolinylidene-N'-salicylhydrazine
pslh	N-picolinylidene-N'-salicyloylhydrazinae
ptcbx	pterin-6-carboxylate
py	pyridine
pyact	pyridine-2-acetamide
pyam	pyrimidin-2-amine
pybox	pyridine-2-carboxamide
2-pyc	pyridine-2-carboxylate
pydca	pyridine-2,6-dicarboxylate
pydcaH	pyridine-2,6-dicarboxylic acid
pydoc	1,8-bis(2'-pyridyl)-3,6-diazaoctane

pym	2-pyridylmethanol
pyf	2-pyridylformamide
pyha	7-(2'-pyridyl)-3,6-diazaheptylamine
pyNO	pyridine 1-oxide
pyos	1,9-bis(2-pyridyl)-5-oxa-2,8-dithianonane
pypr	pyridinopropionate
pytrz	2,4,6-tris(2-pyridyl)-1,3,5-triazine
purp	purpurate
pz	pyrazole
pzc	pyrazinecarboxylate
pzdc	pyrazine-2,3-dicarboxamide
(pz) ₂ (py)CH	bis(pyrazol-1-yl)(pyridin-2-yl)methane
5'-qmp	guanosine 5'-monophosphate
qu	quinoline
qusp	1,5-bis(quinoline-8'-yl)-1,5-dithiapentane
rh	rhombohedral
sach	saccharin, (o-sulphobenzoimide)
sal	salicylate
salen	N-salicylidene-N'-methylenediamine
sbh	Schiff-base ligand
sbzh	thiobenzoylhydrazine
sc	semicarbazide
sim	simatone
SO ₃ CF ₃	trifluoromethanesulphonate
suc	succinate
tach	cis,cis-1,3,5-triaminocyclohexane
taci	1,3,5-triamino-1,3,5-trideoxy-cis-inositol
tcne	O-tricyanoethyleneoate
tcnq	7,7',8,8'-tetracyanoquinodimethane
tdac	thiodiacetate
tdc	1,2,3-triazole-4,5-dicarboxylate
tdpom	trisphenylphosphitomethane
terpy	2,2':6',2''-terpyridine
tfac	thenoyltrifluoroacetate
tg	tetragonal
tgly	N-tosyl-glycinate
thc	L-1,3-thiazolidine-4-carboxylate
thp	3,4,5,6-tetrahydropicolinate (L-lysine)

thr	threonine
tim	2,3,9,10-tetramethyl-1,4,8,11-tetraaza-1,3,8,10-cyclo-tetradecatetraene
tmic	2,2,5,5-tetramethyl-3-imidazoline-1-oxo-4-carboxylate
tmpCN	3-cyano-2,2,5,5-tetramethylpyrrolinyl-1-oxy
tmtc	tetrakis(5-methylbenzo [b,f,j,n][1,5,9,13]tetraacyclohexadecine
tnph	2,4,6-trinitrophenolate
tocop	2,2'-(1,4,10,13-tetraoxa-7,16-diaza-7,16-cyclooctadecylene)-dipropionate
tol	toluene
tols	toluene-4-sulphonate
tpy	tetrapyridine
tr	triclinic
trg	trigonal
trien	triethylenetetramine
trioc	1,3,6,9,11,14-hexaazatricyclo[12.2.1.1 ^{6,9}]octadecane
trt	D-tartrate
tval	N-tosylvalinate
ttbd	4,4,4-trifluoro-1-(2-thienyl)butanedione-1,3
ttbt	1-(2-thienyl)-4,4,4-trifluoro-1,3-butanedionate
ttn	1,4,7-trithiacyclononane
tu	thiourea
1-vim	1-vinylimidazole

INTRODUCTION

The structural chemistry for over one thousand copper compounds has been reviewed recently.¹ Several relationships were found and discussed. The overwhelming majority of X-ray studies of transition metal compounds are of copper compounds, and of these Cu(II) is the most common. Comprehensive data on mononuclear tetra- and pentacoordinate Cu(II) compounds² as well as on mixed-valence Cu(I)-Cu(II) compounds³ have also appeared.

The systems discussed in this paper are mononuclear six-, seven- and eight-coordinate Cu(II) compounds. Within each coordination number, the compounds are listed in order of increasing covalent radius of the principle coordinating ligand atom and increasing complexity of the coordination sphere.

HEXACOORDINATE Cu(II) COMPOUNDS

Structural data for mononuclear six-coordinate Cu(II) compounds are summarized in Table I. There are five hundred and fifty examples.

Because of its electronic configuration, the Cu(II) atom is an ideal candidate for deformation studies of its coordination polyhedron. First of all, it has only one outermost electron which simplifies theoretical considerations. Secondly, if the ligand atoms around the central Cu(II) atom are identical, the coordination polyhedron is subject to the first-order Jahn–Teller effect.

There are three principal forms of distortion of the octahedron. One is tetragonal, elongation or contraction along a single C_4 axis; the resulting symmetry is only D_{4h} . Another is rhombic; changes in the lengths of two of the C_4 axes so that no two are equal; the symmetry is then only D_{2h} . The third is a trigonal distortion, elongation or contraction along one of the C_3 axes so that the symmetry is reduced to D_{3d} . All of the limiting geometries occur in the chemistry of Cu(II) with various degrees of distortion. The tetragonal distortion is most common and involves an elongation of one C_4 axis and, in the limit, two *trans* ligands are lost completely, leaving a square, four-coordinate complex.

The explanation for the axial elongation in *trans*- CuX_4Y_2 type polyhedra is still controversial. Because the highest possible site symmetry around the central Cu(II) atom would be D_{4h} , a first-order Jahn–Teller effect is ruled out. Assuming site symmetry of D_{4h} , the second-order Jahn–Teller deformation takes place only in the CuX_4 plane. On the other hand, it is well known that $d-s$ mixing leads to effects of the same order of magnitude as the first-order ligand field energies. Indeed, this concept has been used to explain axial elongation in coordination polyhedra of the type CuX_4Y_2 . Interestingly, $d-s$ mixing predicts that the most stabilizing distortion in CuX_4Y_2 type coordination polyhedra can be produced by an asymmetric axial elongation. This means that one axial site is weakened more than its antipode. This ultimately leads to five-coordination.

There are some examples^{21–25} which contain a complex cation of composition, $[Cu(pyNO)_6]^{2+}$, where six unidentate pyridine N-oxide molecules form an almost regular octahedron about the Cu(II) atoms.

Finally, there are several examples^{344–358} which contain a complex cation of composition $[CuL_2X]^+$ ($L =$ bipyridine or 1,10-phenanthroline; and $X = NO_2, NO_3$ or $RCOO$) with a fluxional CuN_4O_2 structure, which was described as a *pseudo, cis*-distorted, octahedral Cu(II) stereochemistry and consequently these structural variations can be related to the dynamic *pseudo*-Jahn–Teller effect.

TABLE I Structural data for six-coordinated Cu(II) compounds^a

Compound	Crystal. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
$K_2[Cu(H_2O)_6](SO_4)_2$ (not given)	m P2 ₁ /a 2	9.165(2) 12.224(7) 6.336(2)	103.359(8)	CuO ₆	O ^b 1.936(2) × 2 2.044(2) × 2 2.297(2) × 2	90.0(1, 1.9)	4
$Rb_2[Cu(H_2O)_6](SO_4)_2$ (not given)	m P2 ₁ /a	9.26(2) 12.35(7)	105.3(1)	CuO ₆	O 1.957(3) × 2 2.031(3) × 2 2.307(3) × 2	90.0(3, 0)	5
$Rb_2[Cu(H_2O)_6](SO_4)_2$ (at 293 K)	2	6.22(5)					
$Rb_2[Cu(H_2O)_6](SO_4)_2$ (not given)	m P2 ₁ /a	9.28(1) 12.15(4)	105.4(1)	CuO ₆	O 1.978(8) × 2 2.000(7) × 2 2.317(10) × 2	90.0(3, 2.0)	5
$Tl_2[Cu(H_2O)_6](SO_4)_2$ (not given)	2 P2 ₁ /a	6.23(1)					
$Tl_2[Cu(H_2O)_6](SO_4)_2$ (not given)	m P2 ₁ /a 2	9.27(2) 12.39(7) 6.22(1)	105.6(1)	CuO ₆	O 1.957(5) × 2 2.017(4) × 2 2.317(5) × 2	90.0(2, 4)	6
$K_3[Cu(H_2O)_6](SO_4)_2$ (not given)	not given	not given		CuO ₆	O 1.943(2) × 2 2.069(2) × 2 2.278(2) × 2	not given	7
$Cs_2[Cu(H_2O)_6](SO_4)_2$ (not given)	not given	not given		CuO ₆	O 1.966(5) × 2 2.004(4) × 2 2.315(5) × 2	not given	8
$(NH_4)_2[Cu(H_2O)_6](SO_4)_2$ (not given)	m P2 ₁ /a 2	9.267(2) 12.445(15) 6.298(5)	106.9(6)	CuO ₆	O 1.969(5) × 2 2.095(5) × 2 2.219(5) × 2	90.0(1, 1.4)	9
$(NH_4)_2[Cu(H_2O)_6](SO_4)_2$ (not given)	m P2 ₁ /a 2	9.2105(14) 12.3795(17) 6.3016(13)	106.112(18)	CuO ₆	O 1.9660(11) × 2 2.0725(14) × 2 2.2300(14) × 2	90.0(1, 1.1)	10
$Sr_2[Cu(H_2O)_6]$ (blue)	m P2 ₁ /c 2	5.786(1) 6.154(1) 9.744(2)	124.15(1)	CuO ₆	not given		11
$Ba_2[Cu(H_2O)_6]$ (blue)	m P2 ₁ /c 2	6.030(2) 6.440(2) 10.115(2)	124.03(1)	CuO ₆	O _{eq} O _{ax} 1.965(4, 7) 2.805(4) × 2	90.0(1, 1.1) 90.0(1, 3.1)	11

$[\text{Cu}(\text{H}_2\text{O})_6]\text{SiF}_6$ (not given) (at 240-290 K)	rh R-3 12	18.138(2) 9.787(5)	CuO_6	not given	12
$[\text{Cu}(\text{H}_2\text{O})_6]\text{SiF}_6$ (not given) (at 296-302 K)	rh R-3 3	9.344(2) 9.695(5)	CuO_6	not given	12
$(\text{C}_5\text{H}_{10}\text{N}_2)_3[\text{Cu}(\text{H}_2\text{O})_6](\text{P}_4\text{O}_{12})_2 \cdot 8\text{H}_2\text{O}$ (not given)	m $\text{P2}_1/\text{n}$ 2	13.162(8) 13.301(8) 12.308(8)	CuO_6	O_{eq} O_{ax}	13
$[\text{Cu}(\text{H}_2\text{O})_6] \cdot 2(\text{PhO})_2\text{PO}_2 \cdot (\text{gly})_2$ (not given)	tr P-1 1	17.019(6) 6.520(3) 8.107(3)	CuO_6	O_{eq} O_{ax}	14
$[\text{Cu}(\text{H}_2\text{O})_6](\text{ClO}_4)_2 \cdot 2(\text{c-sar})_2$ (not given)	m $\text{P2}_1/\text{c}$ 4	13.879(7) 14.504(7) 13.083(8)	CuO_6	O_{eq} O_{ax}	15
$[\text{Cu}(\text{H}_2\text{O})_6](\text{bzs})_2^{\text{c}}$ (pale blue)	tr P-1 2	22.51(2) 6.26(1) 6.96(1)	CuO_6	O_{eq} O_{ax}	16
$[\text{Cu}(\text{H}_2\text{O})_6](\text{tols})_2$ (not given)	m $\text{P2}_1/\text{c}$ 2	5.85(1) 25.71(2) 7.35(1)	CuO_6	O_{eq} O_{ax}	16
$[\text{Cu}(\text{H}_2\text{O})_6](\text{camphs})_2$ (not given)	m P2 ₁ 2	17.17(3) 7.05(2) 11.64(2)	CuO_6	O_{eq} O_{ax}	16
$[\text{Cu}(\text{H}_2\text{O})_6](4\text{-Clbzs})_2$ (colourless)	m $\text{P2}_1/\text{c}$ 2	5.7910(5) 25.614(3) 7.3360(9)	CuO_6	O_{eq} O_{ax}	16
$[\text{Cu}(\text{H}_2\text{O})_6][\text{Cu}(\text{C}_8\text{H}_8\text{O}_5)_2]^{\text{c}}$ (pale blue)	m $\text{P2}_1/\text{c}$ 2	9.631(2) 10.725(2) 10.710(2)	CuO_6	$\text{H}_2\text{O}_{\text{eq}}$ $\text{H}_2\text{O}_{\text{ax}}$	18
			CuO_6	O_{eq} O_{ax}	not given

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
$[\text{Cu}(\text{H}_2\text{O})_6][\text{Cu}(\text{C}_8\text{H}_8\text{O}_5)_2]^c$ (green)	tr P-1 1	6.580(1) 8.527(1) 9.859(2)	100.04(2) 101.67(2) 94.75(2)	CuO_6	not given		19
$[\text{Cu}(\text{2-Op})_6][\text{ClO}_4]_2$ (not given)	m P2 ₁ /c 2	9.453(2) 10.492(3) 17.794(3)	95.57(2)	CuO_6 CuO_6	0 2.20×6 1.937(2, 15) 2.546(2) $\times 2$	not given 95.9(1, 4.0) 86.81(7)	20
$[\text{Cu}(\text{pyNO})_6][\text{ClO}_4]_2$ (not given)	rh R-3 1	9.605(1)	81.10(1)	CuO_6	0 $2.076(2) \times 6$	0.0 90.29(7)	21
$[\text{Cu}(\text{pyNO})_6][\text{ClO}_4]_2$ (not given)	rh R-3 1	9.620(2)	81.21(2)	CuO_6	0 $2.086(2) \times 6$	0.0 90.0(1.0)	22
$[\text{Cu}(\text{pyNO})_6][\text{BF}_4]_2$ (not given)	rh R-3 1	9.621(2)	81.46(2)	CuO_6	0 $2.088(1) \times 6$	0.0 90.0(1.0)	22
$[\text{Cu}(\text{pyNO})_6][\text{ClO}_4]_2 \cdot 0.15\text{dmf}$ (not given) (at 20 K)	rh R-3 3	12.260(1) 18.798(2)		CuO_6	0 $2.074(1) \times 6$	0.0 90.0(1.0)	23
$[\text{Cu}(\text{pyNO})_6][\text{NO}_3]_2$ (not given)	rh R-3 1	9.504(3)	83.41(5)	CuO_6	0 $2.097(2) \times 6$	0.0 90.0(4.5)	24
$[\text{Cu}(\text{pyNO})_6][\text{NO}_3]_2$ (not given)	rh R-3 3	12.623(5) 18.190(6)		CuO_6	0 $2.091(4) \times 6$	0.0 90.0(1.3)	25
$[\text{Cu}(\text{pyNO})_6][\text{NO}_3]_2 \cdot 2\text{H}_2\text{O}$ (not given)	m P2 ₁ /c 2	9.556(2) 9.924(2) 18.658(3)	100.43(3)	CuO_6	$1.962(2, 8)$ $2.478(2) \times 2$	$88.9(1)$ $88.0(1, 4.4)$	26

[Cu(4-MepyNO) ₆](ClO ₄) ₂ (not given)	m P2 ₁ /c 2	9.523(5) 10.707(8) 20.512(16)	92.47(6)	CuO ₆	O _{eq} O _{ax}	1.987(0.22) 2.385(11) × 2	O _{eq} , O _{eq} O _{eq} , O _{ax}	90.3(4) 92.1(4, 7)	27
Cu(dmact) ₄ (ClO ₄) ₂ (not given)	m P2 ₁ /n 2	13.080(5) 10.937(5) 9.605(5)	109.95(4)	CuO ₆	O _{eq} O ₃ ClO _{ax}	1.926(5, 16) 2.933(7) × 2	O _{eq} , O _{eq} O _{eq} , O _{ax}	92.3(5) 84.8(7, 4.0)	28
Cu(H ₂ O) ₂ (tsval) ₂ (MeOH) ₂ (not given)	m P2 ₁ /n 2	4.996(2) 19.304(7) 17.124(6)	97.93(1)	CuO ₆	H ₂ O _{eq} tsvalO _{eq} MeHO _{ax}	1.989(3) × 2 1.954(4) × 2 2.492(4) × 2	O _{eq} , O _{eq} O _{eq} , O _{eq} O _{eq} , O _{ax}	89.1(1) 92.3(1, 1.9)	29
[Cu(ahmi) ₆][Cu(ahmi)Cl ₃] ₂ ^e (not given)	m P2 ₁ /c 2	10.757(2) 19.787(3) 14.528(2)	101.26(1)	CuO ₆	O	1.951(4) × 2 1.962(4) × 2 2.514(4) × 2	O, O	90.0(2, 3.8)	30
Cu(H ₂ O) ₄ (MeSO ₃) ₂ (not given)	m P2 ₁ /c 2	9.3064(9) 9.632(2) 7.3076(8)	122.18(1)	CuO ₆	H ₂ O _{eq} MeSO _{ax}	1.958(2, 11) 2.378(2) × 2	O _{eq} , O _{eq} O _{eq} , O _{ax}	89.9(1, 1) 90.1(1)	31
Cu(H ₂ O) ₄ (tols) (light green)	m P2 ₁ /b 2	15.789(4) 11.911(4) 5.086(4)	108.13(20)	CuO ₆	O _{eq} H ₂ O _{eq} H ₂ O _{ax}	2.020(4) × 2 1.973(4) × 2 2.347(4) × 2	O _{eq} , O _{eq} O _{eq} , O _{eq} O _{eq} , O _{ax}	88.6(2, 1.2) 85.9(2)	32
[Cu(H ₂ O) ₄ (3-Ihpur) ₂]:H ₂ O (not given)	m B2/b 4	45.665(5) 8.156(5) 6.828(5)	99.6(2)	CuO ₆	purO _{ax} H ₂ O _{eq} H ₂ O _{ax}	1.88(2) × 2 1.99(2) × 2 2.30(2) × 2	not given	not given	33
Cu(H ₂ O) ₃ (SO ₄)(ami) (green)	tr P-1 2	7.745(2) 9.553(4) 20.343(5)	80.91(3) 79.28(2) 87.58(3)	CuO ₆	H ₂ O _{eq} amiO _{ax} amiO _{ax} O ₃ SO _{ax}	1.973(4, 35) 1.916(4) 2.785(3) 2.238(3)	O _{eq} , O _{eq} O _{eq} , O _{ax} O _{eq} , O _{ax} O _{ax} , O _{ax}	90.4(2, 4.1) 165.5(2, 8.3) 90.1(2, 37.1) 147.4(1)	34
Cu(MeOac) ₂ (H ₂ O) ₂ (not given)	m P2 ₁ /n 2	6.92(2) 7.24(2) 10.10(3)	96.7(1)	CuO ₆	O _{eq} H ₂ O _{ax}	2.04(1, 11) 2.14(1) × 2	not given	not given	35
Cu(picNO) ₂ (H ₂ O) ₂ (not given)	tr P-1 1	6.634(11) 7.302(14) 7.968(1)	80.5(1) 89.4(1) 61.4(1)	CuO ₆	O _{eq} H ₂ O _{ax}	1.935(4, 13) 2.488(5) × 2	O, O	90.19(1) ^d 90.0(2, 2.9)	36

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	<i>a</i> [Å] <i>b</i> [Å] <i>c</i> [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
Cu(4-MeOC ₆ H ₄ OCH ₂ CO ₂) ₂ (H ₂ O) ₂ (blue)	tr P-1 1	14.90(2) 6.56(1) 5.31(1)	104.4(2) 75.4(2) 94.3(2)	CuO ₆	1.945(2) × 2 1.955(2) × 2 2.432(3) × 2	O,O 74.8(1) ^e 93.5(1, 1.5)	37
Cu(C ₆ H ₃ O ₃ S) ₂ (H ₂ O) ₂ (not given)	m P2 ₁ /c 2	8.425(2) 11.532(4) 7.348(9)	92.93(2)	CuO ₆	2.029(2) × 2 1.955(2) × 2 2.290(2) × 2	O,O 76.52(5) ^e 90.1(8, 8)	38
Cu(4-ClphOac) ₂ (H ₂ O) ₂ (pale blue)	tr P-1 1	7.065(3) 13.273(11) 5.114(2)	93.72(5) 103.64(4) 81.63(5)	CuO ₆	1.960(4) × 2 1.956(4) × 2 2.406(4) × 2	O _{eq} ,O _{eq} O _{eq} ,O _{ax} 74.6(3) ^e	39
Cu(dhmp) ₂ (H ₂ O) ₂ (not given)	m C2/c 4	5.688(6) 20.79(3) 10.569(16)	94.55(6)	CuO ₆	1.932(8, 42) 2.542(12) × 2	O _{eq} ,O _{eq} O _{eq} ,O _{ax} 83.5(3) ^e 92.0(4, 1.5)	40
[Cu(acgly) ₂ (H ₂ O) ₂].2H ₂ O (not given)	m C2/c 4	21.13(1) 5.04(1) 17.59(1)	124.7(1)	CuO ₆	1.947(3) × 2 1.955(3) × 2 2.614(3) × 2	O _{eq} ,O _{eq} O _{eq} ,O _{ax} O _{eq} ,O _{ax} 90.6(1) 55.5(1) ^e 88.1(1)	41
[Cu(acgl) ₂ (H ₂ O) ₂].2H ₂ O (blue)	m Cc 4	21.145(6) 5.046(2) 18.248(5)	127.55(4)	CuO ₆	1.944(2) × 2 1.952(2) × 2 2.619(2) × 2	O _{eq} ,O _{eq} O _{eq} ,O _{ax} O _{eq} ,O _{ax} 90.0(1, 6) 55.4(1) 90.0(1, 1, 6) 124.6(1)	42
[Cu(mal) ₂ (H ₃ O) ₂].2H ₂ O ^e (light blue)	tr P-1 2	7.630 10.327 5.285	103.14 96.06 108.54	CuO ₆	1.97(-, 1) 2.48 × 2	O _{eq} ,O _{eq} O _{eq} ,O _{ax} 90.0(-, 1, 2) ^d 90.2(-, 2, 5)	43
Cu(mal) ₂ (H ₂ O) ₂ (not given)	m P2 ₁ /c 2	8.573(1) 7.423(1) 10.315(2)	102.72(1)	CuO ₆	1.98(-, 3) 2.47 × 2 1.937(2, 25) 2.528(2) × 2	O _{eq} ,O _{eq} O _{eq} ,O _{ax} O _{eq} ,O _{eq} O _{eq} ,O _{ax} 87.1 ^d 91.5(-, 1, 5) 83.17(7) ^e 96.83(7)	44

Cu(EtOac) ₂ (H ₂ O) ₂ (blue)	tr P-1	6.60(1) 7.21(1) 7.58(1)	91.6(3) 91.6(3) 117.3(4)	H ₂ O _{eq} O _{ax} O _{ax}	1.990(7) × 2 1.970(7) × 2 2.380(7) × 2	O _{eq} , O _{eq} O _{eq} , O _{ax} O _{ax}	90.9(3) 75.2(3) ^e 89.8(3)	45
[Cu(4-FphOac) ₂ (H ₂ O) ₂]- (4-FphOac) ₂ ·2H ₂ O (dark green)	tr P-1	14.808(2) 9.832(2) 6.847(2)	87.77(2) 98.41(2) 112.33(2)	H ₂ O _{eq} O _{ax} O _{ax}	1.940(3) × 2 1.942(2) × 2 2.471(2) × 2	O _{eq} , O _{eq} O _{eq} , O _{ax} O _{ax}	86.6(1) 73.0(1) ^e 89.4(1)	46
Cu(ac-β-ala) ₂ (H ₂ O) ₂ (blue)	m P2 ₁ /n 2	5.0135(2) 8.415(1) 17.952(2)	91.186(3)	H ₂ O _{eq} O _{eq} O _{ax}	1.960(2) × 2 1.958(2) × 2 2.580(2) × 2	O _{eq} , O _{eq} O _{eq} O _{ax}	89.7(1)	47a
Cu(inco) ₂ (H ₂ O) ₂ (royal blue)	m C2/c 4	14.136(7) 9.791(3) 10.768(3)	110.91(3)	H ₂ O _{eq} O _{ax} O _{ax}	1.930(2) × 2 1.934(4) × 2 2.691(4) × 2	O _{eq} , O _{eq} O _{eq} , O _{ax} O _{ax}	90.0(1, 1.2) 54.7(1) ^f 90.0(1, 2.9) 125.2(1)	47b
Cu(Hpth) ₂ (H ₂ O) ₂ (blue)	m P2 ₁ /c 2	8.31(2) 14.62(2) 7.20(2)	112.2(2)	H ₂ O _{eq} O _{eq} O _{ax}	1.930(8) × 2 1.967(8) × 2 2.677(5) × 2	O _{eq} , O _{eq} O _{eq} , O _{ax} O _{ax}	91.4(3, 0) 54.4(2) ^g 88.7(2, -)	48
trans[Cu(nbta) ₂ (H ₂ O) ₂](pyNO) ₂ (green)	m P2 ₁ /a 2	15.318(5) 9.036(3) 13.253(5)	114.16(1)	O _{eq} H ₂ O _{ax}	1.924(1, 1) 2.594(2) × 2	O _{eq} , O _{eq} O _{eq} , O _{ax}	92.6(0) ^d 93.7(1, 6)	49
Cu(phOac) ₂ (H ₂ O) ₂ ^e (green)	m P2 ₁ /c 6	16.24(4) 7.28(3) 23.99(6)	110.4(3)	H ₂ O _{eq} O _{eq} O _{ax} H ₂ O _{eq} O _{eq} O _{ax}	1.99(-, 1) 1.95(-, 1) 2.47(-, 3) 1.97 × 2 1.94 × 2 2.50 × 2	O _{eq} , O _{eq} O _{eq} , O _{ax} O _{ax} O _{eq} , O _{ax} O _{eq} O _{ax}	89.3(-, 1.2) 74.7(-, 4) ^e 97.0(-, 13.1) 73.6 ^e 90.4(-, 1.4)	50
Cu(MeOac) ₂ (H ₂ O) ₂ (blue)	m P2 ₁ /m 2	6.92(2) 7.26(2) 10.10(3)	96.7(3)	O _{eq} H ₂ O _{ax}	1.93 × 2 2.13 × 2 2.16 × 2	O ₂ O	79.9 ^e 90.7 92.2	50
Cu(2-OH-2-Mepr) ₂ (H ₂ O) ₂ (blue)	m C2/m 2	10.25(5) 5.80(3) 11.24(5)	105.9(5)	O _{eq} H ₂ O _{ax}	1.95(-, 6) 2.56 × 2	O _{eq} , O _{eq} O _{eq} , O _{ax}	82.7 ^e 90.0(-, 0)	50
Cu(NO ₃) ₂ (apy) ₂ (not given)	m P2 ₁ /c 4	13.05(2) 17.50(3) 15.82(3)	136.0(5)	apyO _{eq} (NO ₃) ₂ O _{eq} (NO ₃) ₂ O _{ax}	1.917(16, 20) 1.979(16, 3) 2.481(21, 3)	O _{eq} , O _{eq}	90.3(1, 0, 2, 2) 171.5(2, 0, 2, 2)	51

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å]			α [°]			Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
		a	b	c	α	β	γ				
Cu(NO ₃) ₂ (apy) ₂ (not given)	tr	14.92(3)			117.2(5)			CuO ₆	O _{eq} O _{eq}	90.5(9.9) 168.8(1.6, 1.2)	52
	P-1 2	10.20(2) 9.60(2)			103.8(5) 75.9(5)						
Cu(C ₈ H ₁₁ NO ₃) ₂ (NO ₃) ₂ (light green)	tr	10.483(1)			73.92(2)			CuO ₆	O _{eq} O _{eq} O _{eq} O _{ax}	91.2(1) ^d 87.7(1, 5.8)	53
	P-1 1	6.9278(5) 8.198(1)			67.62(2) 97.81(2)						
Cu(ompa) ₂ (ClO ₄) ₂ (light blue)	m	10.797(1)			109.9(1)			CuO ₆	O _{eq} O ₃ ClO _{ax}	90.0(1, 3) ^d 87.4(1, 2.1)	54
	P2 ₁ /c 2	8.865(1) 20.047(1)									
Cu(C ₅ (CO ₂ Me) ₃) ₂ (MeOH) ₂ (blue green)	m	8.205(5)			108.09(4)			CuO ₆	O _{eq} MeHO _{ap}	93.3(1, 3.4)	55
	P2 ₁ /c 2	20.204(9) 11.975(5)									
[Cu(poet) ₂ (EtOH) ₂](ClO ₄) ₂ (pale blue)	tr	13.688(7)			110.43(4)			CuO ₆	O _{eq} O _{ax}	90.0(1) ^d 90.0(2, 3.8)	56
	P-1 1	14.424(10) 9.865(2)			90.13(2) 115.54(4)						
[Cu(NO ₃) ₂ (Ph ₃ PO) ₂] ₂ ·to (green blue)	m	12.933(3)			100.68(3)			CuO ₆	O _{eq} O _{eq}	56.2(2) ^f 87.6(2, 1.5)	57
	P2 ₁ /c 2	15.532(3) 10.285(3)									
[Cu(pypr) ₄](ClO ₄) ₂ (blue)	tr	7.3760(8)			102.05(1)			CuO ₆	O _{eq} O _{eq} O _{eq} O _{ax}	93.7(1) 50.9(1) ^f 90.7(1)	58
	P-1 1	8.558(1) 15.402(2)			98.55(1) 99.74(1)						
Cu(bpyO ₂ O)(NO ₃) ₂ (H ₂ O) (green)	m	6.635(2)			95.54(1)			CuO ₆	O _{eq} O _{eq}	89.8(1) ^f 90.1(1, 3.0)	59
	P2 ₁ /c 4	7.188(2) 29.038(3)									
[Cu(C ₂ H ₆ O ₂) ₃]SO ₄ (pale blue)	m	10.166(1)			115.666(6)			CuO ₆	O _{eq} O _{ax}	81.2(3) ^e 93.0(4, 5.6) 171.4(3, 5.9)	60
	P2 ₁ /c 4	9.013(1) 15.365(1)									

[Cu(ompa) ₃](ClO ₄) ₂ (not given)	trg P-3cl 2	12.855(1) 90.5(1) 18.260(3)	CuO ₆	O	2.065(2) × 6	O,O	89.5(1, 1.1) ^d	61
[Cu(ipcp) ₃](ClO ₄) ₂ (not given)	m C2/c 8	29.126(6) 23.399(4) 21.696(4)	CuO ₆	O	2.089(2, 22)	O,O	90.5(1.8) ^d 89.9(1, 6.1)	62
[Cu(C ₁₄ H ₁₉ N ₂)(F ₆ acac) ₃] (yellow green)	m P2 ₁ /n 4	12.878(7) 20.998(1) 13.620(7)	CuO ₆	O	2.012(9, 31) 2.179(9, 23)	O,O	87.6(4, 2.9) ^d	63
[Cu(pcp) ₃](ClO ₄) ₂ (colourless)	or Pna2 ₁ 4	19.196(4) 12.131(2) 22.868(4)	CuO ₆	O	2.086(3, 63)	O,O	90.0(1, 3.6) ^d	64
[Cu{Ph ₂ P(O)CH ₂ P(O)Ph ₂ } ₃] [Cu ₂ Cl ₆ ·Me ₂ CO (not given)	m P2 ₁ /c 4	13.144(1) 19.701(3) 30.109(4)	CuO ₆	O	2.022(4, 41) 2.177(4, 79)	O,O	90.7(2, 1.1) ^d 89.8(2, 4.0) 177.4(2, 1.8)	65
Cu(F ₆ acac) ₂ (Phmit) ₂ (blue)	m P2 ₁ /c 4	12.358(6) 15.428(7) 22.61(1)	CuO ₆	O _{eq} nitO _{ax}	1.936(5, 7) 2.378(5, 16)	O _{eq} , O _{eq}	92.3(1, 3) ^d 89.2(1, 2.8) 178.6(2, 8) 90.0(1, 8.3) 172.4(1)	66
(NH ₄) ₄ [Cu(C ₆ H ₅ O ₇) ₂] (pale blue)	m P2 ₁ /c 2	8.755(3) 13.185(4) 9.364(2)	CuO ₆	O	1.969(3) × 2 1.977(3) × 2 2.341(3) × 2	O,O	73.8(1) ^f 85.1(1) ^d 90.3(1) ^g	67
[Cu(topom) ₂](ClO ₄) ₂ ·2H ₂ O (colourless)	tr P-1 1	11.997(3) 12.264(2) 13.159(2)	CuO ₆	O	1.977(2) × 2 2.128(2) × 2 2.152(2) × 2	O,O	88.3(1, 5) ^d 90.9(1, 1.4)	68
[Cu(im) ₆](NO ₃) ₂ (bright blue)	m P2 ₁ /c 2	8.65(1) 17.64(1) 8.38(1)	CuN ₆	N _{eq} N _{ax}	2.031(2, 19) 2.593(3) × 2	N _{eq} , N _{eq} N _{eq} , N _{ax}	91.7(1) 91.8(1, 4)	69
K ₂ Ca[Cu(NO ₂) ₆] (greenish black)	or Fmmm 4	10.743(3) 10.325(2) 10.318(2)	CuN ₆	N _{eq} N _{ax}	2.051(1, 1) 2.313(1) × 2	not given	not given	70
K ₂ Pb[Cu(NO ₂) ₆] (green black) (at 276 K)	or Fmmm 4	10.741(1) 10.734(1) 10.538(1)	CuN ₆	N	2.058(9) × 2 2.153(14) × 2 2.166(13) × 2	not given	not given	71

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
(at 193 K)							
$\text{Rb}_2\text{Pb}[\text{Cu}(\text{NO}_2)_6]^{\text{c}}$ (green black)	or Fmmm 4	10.8296(7) 10.8196(7) 10.6113(7)		CuN_6	$2.071(14) \times 2$ $2.151(16) \times 2$ $2.133(31) \times 2$ $2.036(4) \times 2$ $2.176(5) \times 2$ $2.166(5) \times 2$ $2.063(4) \times 2$ $2.169(5) \times 2$ $2.179(5) \times 2$	not given	71
$\text{Ti}_2\text{Pb}[\text{Cu}(\text{NO}_2)_6]$ (green black)	c Fm3 4	10.7344(5)		CuN_6	$2.118(6) \times 6$	not given	73
$\text{K}_2\text{Ba}[\text{Cu}(\text{NO}_2)_6]$ (not given)	or Fmmm 4	11.219(2) 10.728(2) 10.685(1)		CuN_6	$2.038(2) \times 2$ $2.048(2) \times 2$ $2.311(2) \times 2$	not given	74
$\text{K}_2\text{Sr}[\text{Cu}(\text{NO}_2)_6]$ (green)	or Fmmm 4	10.4301(7) 10.4621(7) 10.9368(9)		CuN_6	$2.029(2) \times 2$ $2.041(2) \times 2$ $2.310(2) \times 2$	not given	75
$[\text{Cu}(4\text{-Mepy})_4(\text{NCS})_2] \cdot 0.67(4\text{-Mepy}) \cdot 0.33\text{H}_2\text{O}$ (green)	trg R-3 9	27.356(7) 11.303(9)		CuN_6	$1.975(7) \times 2$ $2.060(5) \times 2$ $2.503(6) \times 2$	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{ax}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{ax}}$	76
$\text{Cu}(\text{abst})_2(\text{NH}_3)_2$ (dark green)	tr P-1 1	8.693(7) 9.980(14) 9.980(14)	79.86(5) 118.77(7) 74.34(5)	CuN_6	$2.019(6) \times 2$ $2.060(4) \times 2$ $2.479(6) \times 2$	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{ax}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{ax}}$	77
$\text{Cu}(\text{phen})_2(\text{NCS})_2$ (green)	or Pbcn 4	13.029(21) 9.900(15) 17.256(23)		CuN_6	$2.013(10) \times 2$ $2.195(9) \times 2$ $2.125(10) \times 2$	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{ax}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{ax}}$	78
$\text{Cu}(\text{phen})_2(\text{NCS})_2$ (green)	or Pbcn 4	13.250(36) 10.020(6) 17.432(10)		CuN_6	$2.014(14) \times 2$ $2.150(14) \times 2$ $2.255(14) \times 2$	N, N N, N $167.0(6), 2.3$	79

$\text{Cu}(\text{phen})_2(\text{NCSe})_2$ (green)	or Pbcn 4	13.213(7) 10.077(3) 17.725	CuN_6	SeCN phenN	$2.042(13) \times 2$ $2.127(11) \times 2$ $2.183(11) \times 2$	N_iN	77.2(4) ^f 92.6(5, 6.6) 167.0(5, 9)	79
$\text{Cu}(\text{aepy})_2(\text{NCS})_2$ (blue)	m $\text{P}2_{1/c}$ 2	8.398(5) 14.751(10) 7.834(4)	CuN_6	N_{eq} SCN_{ax}	$2.042(2, 20)$ $2.593(3) \times 2$	$\text{N}_{\text{eq}}\text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}\text{N}_{\text{ax}}$	86.9(1) ^d 82.6(1, 8)	80
$\text{Cu}(\text{Me}_3\text{en})_2(\text{NCS})_2$ (not given)	m $\text{P}2_{1/c}$ 2	6.566(1) 14.741(3) 8.582(3)	CuN_6	N_{eq} SCN_{ax}	$2.063(6, 1)$ $2.517(7) \times 2$	$\text{N}_{\text{eq}}\text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}\text{N}_{\text{ax}}$	85.0(2) ^f 85.6(2, 1.5)	81
$\text{Cu}(\text{1,3-pn})_2(\text{NO}_2)_2$ (violet)	m $\text{P}2_{1/c}$ 2	6.088(4) 16.534(7) 6.943(5)	CuN_6	N_{eq} $\text{O}_2\text{N}_{\text{ax}}$	$2.038(6, 10)$ $2.655(8) \times 2$	$\text{N}_{\text{eq}}\text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}\text{N}_{\text{ax}}$	86.8(2) ^d 85.8(2, 3.0)	82a
$\text{Cu}(\text{1,3-pn})_2(\text{acts})_2$ (not given)	m $\text{P}2_{1/n}$ 2	9.336(3) 10.232(1) 14.138(3)	CuN_6	pn N_{eq} acts N_{ax}	$2.046(2, 2)$ $2.457(2) \times 2$	$\text{N}_{\text{eq}}\text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}\text{N}_{\text{ax}}$	87.8(1, 0) ^d 90.4(1, 4)	82b
$\text{Cu}(\text{abp})_2(\text{tcnq})_2$ (black)	tr P-1 1	9.194(1) 9.761(2) 12.235(2)	CuN_6	N_{eq} tcnq N_{ax}	$2.017(5, 30)$ $2.442(5) \times 2$	$\text{N}_{\text{eq}}\text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}\text{N}_{\text{ax}}$ $\text{N}_{\text{ax}}\text{N}_{\text{ax}}$	80.5(2) ^f 99.5(2) 180	83
$[\text{Cu}_2(\text{N}(\text{py})_3)_2(\text{MeCN})_2] \cdot$ $(\text{SO}_3\text{CF}_3)_2$ (dark green)	m $\text{P}2_{1/n}$ 2	11.825(2) 14.774(3) 11.516(2)	CuN_6	N_{eq} MeCN N_{ax}	$2.020(4, 2)$ $2.526(8) \times 2$	$\text{N}_{\text{eq}}\text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}\text{N}_{\text{ax}}$	86.3(2) ^f 86.0(2, 1.0)	84
$[\text{Cu}(\text{en})_3]\text{SO}_4$ (purple)	trg $\text{P}3_1c$ 2	8.966(1) 9.597(1)	CuN_6	N	$2.150(2) \times 6$	N_iN	80.9 ^f 93.5(-, 7)	85
$[\text{Cu}(\text{en})_3]\text{SO}_4$ (blue) (at 120 K)	tr P-1 2	8.814(3) 8.896(3) 9.588(3)	CuN_6	N_{eq}	$2.055(1, 5)$ $2.215(1, 15)$	$\text{N}_{\text{eq}}\text{N}_{\text{eq}}$ $\text{N}_{\text{ax}}\text{N}_{\text{ax}}$	81.3(5, 1.5) 91.60(6, 7.0) 171.5(7, 1.7)	86
$[\text{Cu}(\text{en})_3]\text{Cl}_2 \cdot 0.75\text{en}^{\text{f}}$ (blue)	m C2/c 16	33.774(6) 8.756(2) 23.972(5)	CuN_6	N	$2.14(2, 9)$ $2.21(1, 3)$	N_iN	82.6(6) ^f 79.6(7, 2) ^f 171.06(6)	86
			CuN_6	N	$2.04(3, 13)$ $2.41(3, 8)$	N_iN	76.9(11, -) ^f 91.6(9, 6.0)	
			CuN_6	N	$2.10(4, 14)$ $2.27(4, 3)$	N_iN	90.0(9, 7.0) 164.5(15)	

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(1,2-pn) ₃][Br ₂ ·2H ₂ O (not given)]	m P2 ₁ 2	10.71 8.93 10.88	107.2	CuN ₆	2.09 × 4 2.31 × 2	not given	87
[Cu(bpy) ₃](ClO ₄) ₂ (dark blue)	tr P-1 2	12.673(17) 18.440(21) 7.937(7)	90.37(14) 120.56(13) 98.80(11)	CuN ₆	2.031(8,5) 2.226(7) 2.450(7)	N _{eq} ,N _{eq} N _{eq} ,N _{ax} N _{ax} ,N _{ax}	88
[Cu(phen) ₃](ClO ₄) ₂ (dark blue)	m C2/c 8	35.98(5) 15.97(3) 12.20(3)	101.7(3)	CuN ₆	2.037(14,34) 2.328(11,7)	N _{eq} ,N _{eq} N _{eq} ,N _{ax}	89
[Cu(phen) ₃](ClO ₄) ₂ (not given)	m C2/c 8	35.98(5) 15.97(3) 12.20(3)	101.7(3)	CuN ₆	not given	not given	90
[Cu(phen) ₃](ClO ₄) ₂ (not given)	m P2 ₁ /c 4	9.41(2) 30.11(6) 13.08(3)	111.7(2)	CuN ₆	not given	not given	90
[Cu(phen) ₃](tcnq) ₂ ·0.5MeOH (dark brown)	tr P-1 2	21.945(5) 11.106(3) 10.186(3)	86.50(5) 84.41(6) 83.21(4)	CuN ₆	2.05(1,1) 2.12(1,1) 2.23(1,4)	N,N	91
[Cu(mep) ₃](PF ₆) ₂ (green)	m Cc: C2/c 4	19.950(4) 20.421(5) 12.066(3)	115.90(4)	CuN ₆	2.040(9,57) 2.432(4)	N _{eq} ,N _{eq} N _{eq} ,N _{ax}	92
[Cu(dien) ₂](NO ₃) ₂ (royal blue)	or P2 ₁ 2 ₁ 2 ₁ 4	9.274(16) 12.192(26) 15.029(36)		CuN ₆	2.219(23,49) 2.010(11,8)	N _{eq} ,N _{eq} N _{eq} ,N _{ax} N _{ax} ,N _{ax}	93

[Cu(dien) ₂](NO ₃) ₂ (not given)	or P2 ₁ ² ,2 ₁ 4	9.306(2) 12.248(3) 15.047(3)	CuN ₆	N _{eq} N _{ax}	2.221(6, 66) 2.023(4, 1)	N _{eq} , N _{ax}	not given	94
(at 150 K)	or P2 ₁ ² ,2 ₁ 4	9.274(3) 12.192(15) 15.029(18)	CuN ₆	N _{eq} N _{ax}	2.192(3, 52) 2.022(1, 2)	N _{eq} , N _{ax}	not given	94
[Cu(dien) ₂]Cl ₂ ·H ₂ O (not given)	m P2 ₁ /c 4	13.62(5) 8.80(5) 13.96(5)	CuN ₆	N _{eq} N _{ax}	2.419(7, 47) 2.065(8, 47)	N _{eq} , N _{ax}	83.3(3, 2) 78.7(3, 5) ^f 156.5(3)	95
[Cu(dien) ₂]Cl(ClO ₄) (not given)	m P2 ₁ /c 4	9.23(5) 13.92(5) 14.65(5)	CuN ₆	N _{eq} N _{ax}	2.451(14, 65) 2.047(13, 42)	N _{eq} , N _{ax}	82.8(6, 8) 78.7(6, 9) ^e 155.3(5)	95
[Cu(terpy) ₂](NO ₃) ₂ (dark green)	tg I4 ₁ /a 8	12.476(6) 36.284(13)	CuN ₆	N	1.989(5, 24) 2.085(4) × 2 2.288(4) × 2	N, N	77.4(6, 1.5) ^e 92.8(6, 2.9) 154.7(6, 2.9) 180	96
[Cu(terpy) ₂]Br ₂ ·3H ₂ O (dark green)	tr P-1 2	19.763(4) 9.563(2) 8.537(1)	CuN ₆	N	1.970(9, 15) 2.147(9, 3) 2.203(9, 2)	N, N	78.0(9, 9) ^f	97
[Cu(terpy) ₂](PF ₆) ₂ (green)	tg P-42 ₁ c 2	8.930(3)	CuN ₆	N	1.977(8) × 2 2.179(7) × 4	N, N	76.9(-, -2) ^f	98
[Cu(9-aneN ₃) ₂](ClO ₄) ₂ ·2H ₂ O (blue)	m P2 ₁ /a 4	20.623 9.533(3) 16.909(8) 13.653(9)	CuN ₆	N	2.087(6) × 2 2.176(6) × 2 2.233(7) × 2	N, N	81.0(2, 8) ^f	99
[Cu(tac) ₂]Br ₂ ·4H ₂ O (not given)	or Cmcm 4	9.391(6) 11.475(7) 21.924(12)	CuN ₆	N	2.086(4) × 2 2.322(6) × 2	N, N	85.7(2) ^f 91.4(2, 2.9) 180.0(1)	100
[Cu(bpsca) ₂]·H ₂ O (green)	m P2 ₁ /c 4	8.917(1) 8.932(1) 28.794(17)	CuN ₆	N	1.979(3, 19) 2.091(3, 7) 2.299(3, 6)	N, N	78.9(1, 2.0) ^f 95.2(1, 11.6) 163.2(1.1, 1.7)	101
[Cu(metri) ₂](BF ₄) ₂ (orange)	c Pa3 4	16.621(5)	CuN ₆	N	2.100(7) × 6	N, N	82.9(3) ^f 97.1(3)	102
(at 105 K)	c Pa3 4	16.445(5)	CuN ₆	N	2.058(6) × 6	N, N	82.8(3) ^f 97.2(3)	102

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
$[\text{Cu}(\text{HB}(\text{pz})_3)_2]$ (not given)	tr P-1 2	10.309(2) 13.794(3) 11.677(3)	127.39(2) 88.46(2) 116.63(2)	CuN_6	$1.998(2) \times 2$ $2.082(2) \times 2$ $2.356(4) \times 2$	N,N 81.9(1) ^e 87.3(1.7)	103
$[\text{Cu}(\text{HB}(3,5\text{-Me}_2\text{pz})_2)_2]$ (blue green)	tr P-1	10.201(1) 10.875(1)	83.68(1) 101.62(1)	CuN_6	$2.028(6, 15)$ $2.137(6, 4)$	N,N 86.9(2, 1.7) ^d	104
$[\text{Cu}(\text{tach})_2(\text{NO}_3)_2]$ (not given)	1 m C2/m 2	8.787(1) 18.496(4) 7.289(2) 7.201(2)	117.59(1) 110.22(2)	CuN_6	$2.288(7, 36)$ $2.169(6, 5)$	N,N 87.7(1, 2.0) ^d	105
$[\text{Cu}(\text{tach})_2(\text{ClO}_4)_2]$ (not given)	or Cmca 4	9.68(1) 18.00(2) 12.14(1)		CuN_6	$2.070(6, 9)$ $2.353(7) \times 2$	N,N 85.4(2, 1.4) ^d 90.3(2)	105
$[\text{Cu}_2(\text{pz})_2(\text{py})\text{CH}_2]_2(\text{NO}_3)_2]$ (blue)	m P2 ₁ /c 2	10.207(3) 7.756(2) 18.806(11)	113.28(3)	CuN_6	$1.994(3) \times 2$ $2.020(3) \times 2$ $2.385(4) \times 2$ not given	N,N 81.8(1) 86.8(1, 1)	106
$[\text{Cu}(\text{NH}(\text{CH}_2\text{C}_7\text{H}_5\text{N}_2)_2)_2](\text{ClO}_4)_2 \cdot 2\text{H}_2\text{O} \cdot 2\text{MeOH}$ (blue)	tr P-1 1	9.576(3) 10.123(3) 11.628(5)	83.31(3) 80.93(3) 87.25(2)	CuN_6	$2.019(8, 63)$ $2.052(7)$ $2.374(10, 5)$	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{ax}}$ $\text{N}_{\text{ax}}, \text{N}_{\text{ax}}$	108
$[\text{Cu}(\text{pcam})(\text{pytrz})](\text{CF}_3\text{SO}_3)$ (green)	tr P-1 2	9.177(3) 10.735(3) 18.231(4)	102.65(2) 106.60(2) 93.89(3)	CuN_6	pcam N_{eq} pytrz N_{eq} pytrz N_{ax}	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{ax}}$ $\text{N}_{\text{ax}}, \text{N}_{\text{ax}}$	109
$\text{Cu}(\text{dbdd})(\text{NCS})_2$ (not given)	tg P4 ₁ 2 ₁ 2 4	9.614(5) 24.854(8)		CuN_6	$2.191(4) \times 2$ $2.193(4) \times 2$ $2.034(4) \times 2$	N,N 81.9(1) ^f 98.6(3.2) 179.6(2)	109
(at 140 K)	tg P4 ₁ 2 ₁ 2 4	9.677(4) 24.422(4)		CuN_6	$2.242(4) \times 2$ $2.099(4) \times 2$ $2.060(4) \times 2$	N,N 81.9(1) ^f 94.3(1, 5.4) 170.5(1, 2.7)	

[Cu(edtb)]BF ₄ (BF ₃ OEt) ₂ ·H ₂ O (blue)	m C2/c 4	14.402(2) 14.953(2) 18.846(2)	101.17(1)	CuN ₆	N	1.986(7) × 2 2.039(8) × 2 2.499(8) × 2	N,N	73.8(3, 3.7) ^e 91.9(3, 1.4) 142.8(3.2) 165.9(5)	110
[Cu(C ₂₂ H ₁₄ N ₆)] [CuCl ₄]·2H ₂ O (green)	m P2 ₁ /c 4	15.467(4) 11.287(9) 17.459(5)	106.45(2)	CuN ₆	N _{eq} N _{ax}	2.264(5, 50) 1.985(5, 2)	N,N	77.8(2, 3) ^f 155.4(2, 3) 178.2(2)	111
[Cu(ttcn)] ₂ (BF ₄) ₂ (dark brown)	or Pbca 4	21.169(3) 15.193(2) 8.729(2)		CuS ₆	S	2.435(3, 16)	S,S	87.8(2, 6) ^e 92.2(2, 6)	112
[Cu(ttcn)] ₂ (BF ₄) ₂ (MeCN) ₂ ^c (brown) (at 117 K)	m P2 ₁ /c 4	20.695(2) 14.944(1) 8.864(1)	90.797(8)	CuS ₆	S	2.429(1, 30)	S,S	88.1(1, 2) ^d 91.9(1, 2) 179.9(1, 1) 88.3(1, 4) ^d 91.7(1, 4) 179.9(1, 1)	113
[Cu(ttcn)] ₂ (BF ₄) ₂ (MeNO ₂) ₂ (brown)	or Pbca 4	19.746(2) 15.442(2) 9.227(1)		CuS ₆	S	2.343(2) × 2 2.487(2) × 2 2.504(2) × 2	S,S	87.8(1, 8) 92.2(1, 8) 180.0	113
[Cu(H ₂ O) ₅ (bim)](S'-imp)·3H ₂ O (light blue)	m P2 ₁ 2	7.013(2) 13.179(9) 14.565(9)	94.82(4)	CuO ₅ N	H ₂ O _{eq} N _{eq} H ₂ O _{ax} H ₂ O _{ax}	2.017(5, 62) 1.947(5) 2.194(6) 2.732(6)	O _{eq} N _{eq} O _{eq} O _{ax} O _{ax} O _{ax}	89.8(2, 3.9) 90.0(2, 12.2) 172.6(2)	114
[Cu(9-Meade)(H ₂ O) ₄ (SO ₄)]·H ₂ O (turquoise)	tr P-1 2	14.079(2) 7.150(3) 7.853(2)	100.52(2) 75.87(2) 107.98(2)	CuO ₅ N	H ₂ O _{eq} N _{eq} H ₂ O _{ax} O ₃ SO _{3ax}	1.968(2, 33) 1.995(2) 2.355(2) 2.551(2)	O _{eq} O _{eq} O _{eq} N _{eq} O _{eq} O _{ax} N _{eq} O _{ax} O _{ax} O _{ax}	89.19(5, 14.5) 91.13(5, 1.05) 87.16(5, 9.53) 98.25(5, 5.40) 162.94(5)	115
Cu(Mephim(ac)) ₂ (H ₂ O) ₂ (blue green)	m P2 ₁ /c 4	11.452(2) 7.595(2) 15.416(2)	101.60(1)	CuO ₅ N	H ₂ O _{eq} O _{eq} N _{eq} O _{ax} H ₂ O _{ax}	1.962(4) 1.935(4, 3) 2.049(5) 2.337(5) 2.424(4)	O _{eq} O _{eq} O _{eq} N _{eq} N _{eq} O _{ax} O _{ax} O _{ax}	95.9(2, 1.9) 77.5(3) 84.2(2, 8) 90.2(2, 4.3) 172.4(3)	116
Cu(F ₆ acac) ₂ (tmpCN)·H ₂ O (green yellow)	tr P-1 2	11.835(2) 11.942(2) 11.988(2)	111.21(1) 95.31(1) 118.65(1)	CuO ₅ N	O _{eq} CN _{ax} H ₂ O _{ax}	1.951(3, 6) 2.616(5) 2.255(4)	O _{eq} O _{eq} O _{ax} O _{ax} O _{ax} O _{ax}	92.57(11, 5) ^d 87.19(8, 27) 175.37(10)	117

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
Cu(veda)(H ₂ O) ₂ (bright blue)	m	7.50(1)	91.27(1)	CuO ₂ N	O _{eq}	87.32(11, 3.90) 174.84(11)	118
	P2 ₁ /n	28.60(1)			N _{eq}		
	4	7.50(1)			H ₂ O _{eq}		
					O _{ax}		
				H ₂ O _{ax}			
Cu(idac)(H ₂ O) ₂ (not given)	or	10.56(3)		H ₂ O _{eq}	2.00(2)		119
	Pbca	13.68(3)		O _{eq}	2.00(2, 1)		
	8	10.28(3)		N _{eq}	2.07(2)		
				O _{ax}	2.50(2)		
				H ₂ O _{ax}	2.50(2)		
Cu(F ₆ atac) ₂ (tmpCN)H ₂ O (green yellow)	tr	11.835(2)	111.21(1)	CuO ₂ N	O _{eq}	89.9(1, 32) ^d 174.6(1.8) 93.1(9, 1.2) 91.2(1) 174.8(1)	120
	P-1	11.942(2)	95.31(1)		H ₂ O _{ax}		
	2	11.988(2)	118.65(1)		CN _{ax}		
[Cu(phen) ₂ (im)(PF ₆) ₂ PF ₆ (not given)	m	16.866(5)		imN _{eq}	1.982(5)	81.9(2) ^f 92.5(2, 6) 79.1(3) ^e 99.2(2, 8, 1)	121
	P2/c	14.913(3)	105.86(2)	N _{eq}	2.028(6, 15)		
	4	12.417(3)		N _{ax}	2.219(7)		
				F _{ax}	2.782(6)		
[Cu(mbpy) ₂ (ClO ₄) ₂ ·H ₂ O (light green)	m	8.4824(22)		N _{eq}	1.966(5, 8)	100.4(8, 1.3) 80.0(7, 7) ^e 91.5(2, 8, 9) 165.8(2)	122
	P2 ₁ /c	29.1965(29)	95.836(17)	N _{ax}	2.023(5, 19)		
	4	16.7393(24)		O ₃ ClO _{4ax}	2.510(5)		
				F _{eq}	2.698(7)		
(NH ₄) ₂ [CuF ₄ (H ₂ O) ₂] (blue)	m	10.463(3)		F _{eq}	1.918(3, 1)	90.99(6) 88.68(8) 88.47(9) 178.0(1) 177.4(2)	123
	C2/c	9.245(2)	108.61(2)	H ₂ O _{eq}	1.982(2) × 2		
	4	7.368(2)		F _{ax}	2.260(2) × 2		

$\text{Cu}(\text{H}_2\text{O})_4\text{SiF}_6$ (dark blue)	m $\text{P}2_1/\text{b}$ 2	5.346(2) 7.221(2) 9.618(3)	105.21(1)	CuO_4F_2	$\text{H}_2\text{O}_{\text{eq}}$ F_{ax}	1.957(2, 3) 2.333(1) × 2	$\text{O}_{\text{eq}}, \text{O}_{\text{eq}}$ $\text{O}_{\text{eq}}, \text{F}_{\text{ax}}$	88.79(6) 88.26(5, 82)	124
$[\text{Cu}(\text{H}_2\text{O})_4(9\text{-Meade})_2]\text{Cl}_2 \cdot 2\text{H}_2\text{O}$ (green)	m $\text{C}2/\text{m}; \text{C}2$ 2	15.482(7) 6.894(10) 11.269(5)	114.42(3)	CuO_4N_2	$\text{H}_2\text{O}_{\text{eq}}$ N_{ax}	2.162(2) × 4 2.008(2) × 2	$\text{O}_{\text{eq}}, \text{O}_{\text{eq}}$ $\text{O}_{\text{eq}}, \text{N}_{\text{eq}}$	90.6(2, 0) 90.7(2, 0)	125
$[\text{Cu}(\text{H}_2\text{O})_4(\text{opca})_2](\text{NO}_3)_2$ (not given)	tr P-1 2	8.197(6) 10.107(5) 15.534(4)	116.90(5) 87.56(4) 101.30(6)	CuO_4N_2	$\text{H}_2\text{O}_{\text{eq}}$ N_{eq} $\text{H}_2\text{O}_{\text{ax}}$ N_{ax}	2.131(10, 81) 1.923(6) 2.403(11) 2.243(6)	$\text{O}_{\text{eq}}, \text{O}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{O}_{\text{eq}}, \text{O}_{\text{ax}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$ $\text{O}_{\text{ax}}, \text{O}_{\text{ax}}$	159.9(4) 89.6(4, 9, 2) 172.0(3) 89.7(4, 13, 0) 89.8(5, 9, 0) 170.5(3)	126
$[\text{Cu}(\text{H}_2\text{O})_4(\text{hmmc})_2](\text{NO}_3)_2$ (light blue)	tr P-1 1	8.154(1) 8.575(1) 8.761(1)	101.31(1) 104.67(1) 115.53(1)	CuO_4N_2	N_{eq} $\text{H}_2\text{O}_{\text{eq}}$ $\text{H}_2\text{O}_{\text{ax}}$	2.031(7) × 2 2.120(7) × 2 2.193(7) × 2	$\text{N}_{\text{eq}}, \text{O}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$ $\text{O}_{\text{eq}}, \text{O}_{\text{ax}}$	90.6(3) 90.9(3) 91.5(3)	127
$[\text{Cu}(\text{H}_2\text{O})_4(\text{sach})_2] \cdot 2\text{H}_2\text{O}$ (not given)	m $\text{P}2_1/\text{c}$ 2	8.384(2) 16.327(2) 7.327(2)	101.08(2)	CuO_4N_2	N_{eq} $\text{H}_2\text{O}_{\text{eq}}$ $\text{H}_2\text{O}_{\text{ax}}$	2.061(2) × 2 1.956(2) × 2 2.489(2) × 2	$\text{O}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{O}_{\text{eq}}, \text{O}_{\text{ax}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$	90.2(1) 94.0(1) 90.3(1)	128
$\text{Cu}(\text{H}_2\text{O})_4(\text{mtpo})_2$ (light green)	m $\text{P}2_1/\text{c}$ 2	9.497(3) 13.748(2) 6.850(4)	110.34(2)	CuO_4N_2	$\text{H}_2\text{O}_{\text{eq}}$ N_{eq} $\text{H}_2\text{O}_{\text{ax}}$	2.066(2) × 2 1.962(2) × 2 2.391(2) × 2	$\text{O}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{O}_{\text{eq}}, \text{N}_{\text{ax}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$	90.5(7, 1, 3) 90.0(8, 5, 3) 90.0(7, 1, 6)	129
$\text{Ca}[\text{Cu}(\text{H}_2\text{O})_4(\text{S}'\text{-imp})_2] \cdot 6.7\text{H}_2\text{O}$ (not given)	or $\text{C}222_1$ 4	not given		CuO_4N_2	N_{eq} $\text{H}_2\text{O}_{\text{eq}}$ $\text{H}_2\text{O}_{\text{ax}}$	2.04(1) × 2 1.97(1) × 2 2.52(1) × 2	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{eq}}$	88.1(6) 89.6(6) 177.3(6)	130
$[\text{Cu}(\text{H}_2\text{O})_4(9\text{-Mehpx})_2]\text{SO}_4$ (not given)	m $\text{P}2_1/\text{c}$ 4	8.709(3) 13.348(3) 16.778(1)	105.44(2)	CuO_4N_2	N_{eq} $\text{H}_2\text{O}_{\text{eq}}$ $\text{H}_2\text{O}_{\text{ax}}$	2.026(2, 4) 1.953(2, 6) 2.409(2, 4)	$\text{O}_{\text{eq}}, \text{O}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$ $\text{O}_{\text{eq}}, \text{O}_{\text{ax}}$ $\text{O}_{\text{ax}}, \text{O}_{\text{ax}}$	not given	131
$[\text{Cu}(\text{H}_2\text{O})_4(8\text{-aad})_2](\text{NO}_3)_2$ (not given)	m $\text{P}2_1/\text{n}$ 2	7.582(2) 11.538(2) 10.773(1)	95.02(2)	CuO_4N_2	$\text{H}_2\text{O}_{\text{eq}}$ N_{eq} $\text{H}_2\text{O}_{\text{ax}}$	1.961(2) × 2 1.961(2) × 2 2.624(2) × 2	$\text{O}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{O}_{\text{eq}}, \text{O}_{\text{ax}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$	88.8(1) 89.9(1) 89.0(1)	132

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	<i>a</i> [Å]			α [°]			Chromiophore	Cu-L [Å]	L-Cu-L [°]	Ref.
		<i>b</i> [Å]	<i>c</i> [Å]	β [°]	γ [°]						
[Cu(H ₂ O) ₄ (1-CNgud) ₂](NO ₃) ₂ (not given)	tr	3.508(2)			102.45(3)			H ₂ O _{eq}	1.96 × 2	not given	133
	P-1	10.201(3)			101.58(3)			N _{eq}	1.92 × 2		
[Cu(H ₂ O) ₄ (hpca) ₂](NO ₃) ₂ (not given)	tr	8.197(6)			116.90(5)			H ₂ O _{eq}	2.246(11, 15)	not given	134
	P-1	10.107(5)			87.56(4)			N _{ax}	2.060(8)		
[Cu(H ₂ O) ₄ (py) ₂](cas) · 10H ₂ O (blue green)	tr	15.438(4)			112.74(4)			H ₂ O _{eq}	1.923(6)	not given	135
	P-1	15.727(6)			102.02(4)			N _{eq}	2.11(2, 1)		
Cu(py) ₄ (bzs) ₂ (not given)	tr	12.121(9)			85.35(4)			H ₂ O _{ax}	2.35(2, 1)		
	C2/c	15.180(6)			96.38(4)			pyN _{eq}	2.013(9, 7)	90.0(2, 2)	136
[Cu(py) ₄ (F ₃ ac) ₂]; [Cu(py) ₂ (F ₃ ac) ₂] ^c (not given)	4	14.431(5)						O _{ax}	2.471(8) × 2	90.3(3, 2.6)	
	P2/c	15.269(6)						O _{eq}	O _{eq} , O _{ax}	87.8(2)	137
Cu(H ₂ O) ₂ (py) ₂ (act-DL) ₂ (blue)	m	10.451(5)			98.81(13)			O _{eq}	2.067(5)		
	P2/c	14.596(6)						pyN _{eq}	2.075(10, 23)		
Cu(H ₂ O) ₂ (py) ₂ (act-DL) ₂ (blue)	4	17.881(4)						O _{ax}	2.143(6)	not given	
	m	9.377(6)						pyN _{eq}	2.049(11, 7)		
Cu(H ₂ O) ₂ (py) ₂ (fm) ₂ (blue)	2	19.341(14)			123.2(2)			O _{ax}	2.343(6) × 2		
	P2 ₁ /c	11.615(7)						O _{eq}	2.02(1) × 2	87.9(6)	138
Cu(H ₂ O) ₂ (py) ₂ (fm) ₂ (blue)	4	16.493(1)			113.11(1)			pyN _{eq}	1.95(1) × 2		
	C2/c; Cc	6.320(2)						H ₂ O _{ax}	2.61(1) × 2		
Cu(H ₂ O) ₂ (py) ₂ (mco) ₂ (blue)	m	15.184(1)						O _{eq}	1.977 × 2	not given	139
	4	7.036(5)						N _{eq}	2.017 × 2		
Cu(H ₂ O) ₂ (py) ₂ (mco) ₂ (blue)	4	15.184(1)						H ₂ O _{ax}	2.455 × 2		
	tr	9.640(3)			110.05(3)			O _{eq}	1.973(1) × 2	90.00(3, 1.4)	140
Cu(H ₂ O) ₂ (py) ₂ (psac) ₂ (blue)	P-1	9.752(4)			70.95(3)			N _{eq} , O _{ax}	2.006(1) × 2	90.00(2, 2.4)	
	1	9.752(4)			104.28(5)			N _{ax}	2.509(1) × 2	90.00(3, 1.3)	
Cu(H ₂ O) ₂ (py) ₂ (psac) ₂ (blue)	m	23.58(1)			107.05(3)			O _{eq}	1.99(2, 2)	179.2(7)	189
	C2	7.747(3)						pyN _{eq}	2.02(2, 3)	90.0(7, 1.9)	
	4	16.082(8)						H ₂ O _{ax}	2.55(2, 4)	90.0(7, 3.6)	

Cu(H ₂ O) ₂ (ac) ₂ (aine) ₂ (not given)	m P2 ₁ /n 2	7.782(4) 11.137(9) 11.324(4)	97.67(4)	CuO ₄ N ₂	O _{eq} N _{eq} H ₂ O _{ax}	1.944(5) × 2 2.046(5) × 2 2.594(5) × 2	O _{ax} , O _{ax} N _{eq} , O _{ax} N _{eq} , N _{eq}	177.8(6) 90.0(7, 5.2) 178.9(8)	141
Cu(H ₂ O) ₂ (iqu) ₂ (ClO ₄) ₂ (bright green)	tr P-1	7.487(2) 7.621(2)	100.82(2) 108.33(3)	CuO ₄ N ₂	N _{eq} H ₂ O _{eq} O ₃ ClO _{ax}	1.999(2) × 2 1.999(1) × 2 2.461(2) × 2	N _{eq} , O _{eq} O _{eq} , O _{ax} N _{eq} , O _{ax}	88.4(2) 95.3(2) 91.5(2)	142
Cu(H ₂ O) ₂ (ine) ₂ (fm) ₂ (dark blue)	m P2 ₁ /a 2	7.349(4) 14.40(1) 8.738(5)	104.62(5)	CuO ₄ N ₂	N _{eq} O _{eq} H ₂ O _{ax}	2.035(7) × 2 1.978(6) × 2 2.458(6) × 2	O _{eq} , N _{eq} O _{eq} , O _{ax} N _{eq} , O _{ax}	87.8(3) 88.4(3) 93.5(3)	143
[Cu(H ₂ O) ₂ (fm) ₂ (nc) ₂].2H ₂ O (dark blue)	m C2/c 4	14.417(8) 9.856(6) 15.089(9)	109.25(5)	CuO ₄ N ₂	O _{eq} N _{eq} H ₂ O _{ax}	1.981(2) × 2 2.019(3) × 2 2.626(5) × 2	O _{eq} , N _{eq} O _{eq} , O _{ax} N _{eq} , O _{ax}	90.1(3) 98.11(1) 86.4(1)	144
Cu(H ₂ O) ₂ (NO ₃) ₂ (xan) ₂ (green)	tr P-1 1	7.111(2) 7.528(1) 8.888(1)	108.63(1) 98.29(2) 102.39(2)	CuO ₄ N ₂	H ₂ O _{eq} N _{eq} O ₂ NO _{ax}	1.938(1) × 2 2.050(1) × 2 2.540(1) × 2	O, O N, N O, N	90.0(1, 7) 180 90.0(1, 3.6)	145
[Cu(H ₂ O) ₂ (im) ₂ (ac-DL-pgly) ₂] (deep blue)	tr P-1 1	10.156(7) 9.488(6) 8.452(3)	84.45(3) 93.39(3) 63.55(3)	CuO ₄ N ₂	imN _{eq} O _{eq} H ₂ O _{ax}	1.989(3) × 2 1.956(3) × 2 2.912(5) × 2	O _{eq} , N _{eq}	88.4(2)	146
Cu(9-Mequa) ₂ (NO ₃) ₂ (H ₂ O) ₂ (not given)	m P2 ₁ /c 2	5.4087(8) 16.696(15) 11.8414(7)	109.874(15)	CuO ₄ N ₂	H ₂ O _{eq} N _{eq} O ₂ NO _{ax}	1.960(3) × 2 2.004(3) × 2 2.484(3) × 2	N _{eq} , O _{eq} N _{eq} , O _{ax} O _{eq} , O _{ax}	87.57(2) 85.92(2) 89.97(2)	147
Cu(im) ₂ (mact) ₂ (ClO ₄) ₂ (blue)	m P2 ₁ /c 2	7.720(2) 16.169(3) 8.742(2)	96.08(1)	CuO ₄ N ₂	imN _{eq} O _{eq} O ₃ ClO _{ax}	1.97(6) × 2 1.982(5) × 2 2.760(10) × 2	N _{eq} , O _{eq} N _{eq} , N _{eq} N _{eq} , O _{ax} O _{eq} , O _{ax}	90.0(2, 1) 180.0(1) 89.2(4, 4.2) 90.0(4, 9.0)	148
[Cu(H ₂ O) ₂ (bpy)(NO ₃)].NO ₃ (not given)	tr P-1 2	9.52(1) 7.80(1) 13.62(2)	110.4(2) 124.3(2) 77.3(1)	CuO ₄ N ₂	H ₂ O _{eq} N _{eq} H ₂ O _{ax} O ₂ NO _{ax}	2.01(1) × 2 2.01(1, 2) 2.36(1) 2.70(1)	N _{eq} , O _{ax} N _{eq} N _{eq} , O _{ax} N _{eq} , O _{ax}	91.0(4, 6.1)	149
[Cu(F ₃ ac)(py) ₂].[Cu(F ₃ ac) ₂ (py) ₄] ^c (not given)	m P2 ₁ /c 4	10.451(5) 14.596(6) 17.881(4)	98.81(13)	CuO ₄ N ₂	acO _{eq} N _{eq} acO _{ax}	2.067(5) × 2 2.075(10, 2.2) 2.143(6) × 2	O _{eq} , N _{eq} O _{eq} , O _{ax}	89.6(2, 5) 87.8(2) ^d	150

β -Cu(pipe) ₂ (H ₂ O) ₂ (dark blue)	m	5.875(2)	CuO ₄ N ₂	O _{eq}	1.966(2) × 2	N _{eq} O _{eq}	156	84.1(1) ^c
	P2 ₁ /n	18.618(5)		N _{eq}	2.002(2) × 2	O _{eq} O _{ax}		93.0(1)
	2	7.014(2)		H ₂ O _{ax}	2.611(4) × 2	N _{eq} O _{ax}		90.5(1)
Cu(pzo) ₂ (H ₂ O) ₂ (blue)	m	5.438(1)	CuO ₄ N ₂	O _{eq}	1.98(1) × 2	O _{eq} N _{eq}	157	82.9(2) ^c
	P2 ₁ /c	10.880(2)		N _{eq}	2.00(1) × 2	N _{eq} O _{ax}		94.0(2)
Cu(prol) ₂ (H ₂ O) ₂ (blue)	m	5.62	CuO ₄ N ₂	O _{eq}	2.03 × 2	N _{eq} O _{eq}	158	82 ^c
	P2 ₁ /n	17.85		N _{eq}	1.99 × 2	O _{eq} O _{ax}		72
Cu(mbpym) ₂ (H ₂ O) ₂ (dark green)	m	7.13	CuO ₄ N ₂	H ₂ O _{ax}	2.52 × 2	N _{eq} O _{ax}	159	93
	P2 ₁ /c	13.181(1)		O _{eq}	2.045(3) × 2	N _{eq} O _{eq}		81.39(4) ^c
Cu(bz-L-val) ₂ (H ₂ O) ₂ (blue)	m	8.973(1)	CuO ₄ N ₂	N _{eq}	1.977(2) × 2	N _{eq} O _{ax}	160	91.07(16)
	P2 ₁ /c	7.637(1)		H ₂ O _{ax}	2.285(3) × 2	O _{eq} O _{ax}		90.54(2)
Cu(thc) ₂ (H ₂ O) ₂ (not given)	tr	9.827(5)	CuO ₄ N ₂	O _{eq}	1.94(1) × 2	O ₁ N	161	85.2(7) ^c
	P-1	11.435(5)		N _{eq}	2.03(2) × 2	O ₁		94.8(7)
[Cu(tmic) ₂ (H ₂ O) ₂] ₂ ·2H ₂ O (green)	1	5.975(3)	CuO ₄ N ₂	H ₂ O _{ax}	2.82(1) × 2	O ₁	162	180.0(7)
	tr	6.670(1)		O _{eq}	1.970(12, 9)	O _{eq} N _{eq}		85.2(5, 2) ^c
[Cu(tdc) ₂ (H ₂ O) ₂] ₂ ·2H ₂ O (blue)	P-1	9.404(1)	CuO ₄ N ₂	N _{eq}	2.008(13, 8)	O _{ax} N _{eq}	163	90.3(5, 5, 6)
	1	5.608(1)		H ₂ O _{ax}	2.639(13)	O _{ax} O _{eq}		90.0(5, 6, 0)
[Cu(himp) ₂ (H ₂ O) ₂] ₂ ·2H ₂ O (pale blue)	m	9.973(2)	CuO ₄ N ₂	O _{eq}	2.02(1, 8)	N _{eq} N _{eq}	164	170.2(6)
	P2 ₁	8.947(2)		N _{eq}	2.05(1, 5)	O _{eq} O _{ax}		169.4(4)
[Cu(aoac) ₂ (H ₂ O) ₂] ₂ ·2H ₂ O (blue)	2	13.400(3)	CuO ₄ N ₂	H ₂ O _{ax}	2.31(1, 14)	O _{ax} O _{ax}	165	178.4(5)
	tr	5.203(1)		O _{eq}	1.952(2) × 2	O _{eq} N _{eq}		82.0(-, 4.1) ^c
[Cu(himp) ₂ (H ₂ O) ₂] ₂ ·2H ₂ O (pale blue)	P-1	6.583(2)	CuO ₄ N ₂	N _{eq}	2.05(1, 5)	O _{eq} O _{ax}	163	86.8(-, 3)
	1	11.835(2)		H ₂ O _{ax}	2.490(2) × 2	N _{eq} O _{ax}		89.3(1) ^c
[Cu(aoac) ₂ (H ₂ O) ₂] ₂ ·2H ₂ O (blue)	m	6.50(1)	CuO ₄ N ₂	O _{eq}	1.92 × 2	N _{eq} O _{eq}	164	84.6 ^c
	P2 ₁ /n	1.975(2)		N _{eq}	2.00 × 2	O _{eq} O _{ax}		92.0
[Cu(aoac) ₂ (H ₂ O) ₂] ₂ ·2H ₂ O (blue)	2	7.26(1)	CuO ₄ N ₂	H ₂ O _{ax}	2.57 × 2	N _{eq} O _{ax}	165	86.0
	or	5.2092(3)		O _{eq}	1.967 × 2	N _{eq} O _{eq}		94.28 ^d
[Cu(aoac) ₂ (H ₂ O) ₂] ₂ ·2H ₂ O (blue)	Pbca	18.8395(9)	CuO ₄ N ₂	N _{eq}	1.978 × 2	O _{eq} O _{ax}	165	95.35
	4	11.1252(6)		H ₂ O _{ax}	2.545(1) × 2	N _{eq} O _{ax}		89.12

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	<i>a</i> [Å] <i>b</i> [Å] <i>c</i> [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(pcb) ₂ (H ₂ O) ₂] \cdot 6H ₂ O (blue)	m P2 ₁ /n 2	6.849(4) 20.552(4) 7.291(4)	108.79(5)	CuO ₄ N ₂	O _{eq} N _{eq} H ₂ O _{ax}	82.4(1) ^c	166
[Cu(thp) ₂ (H ₂ O) ₂] \cdot 6H ₂ O (dark green)	tr P-1 1	10.713(5) 7.321(4) 6.904(4)	107.66(5) 90.55(5) 95.85(5)	CuO ₄ N ₂	O _{eq} N _{eq} H ₂ O _{ax}	82.53(8) ^e	167
[Cu(pybox) ₂ (H ₂ O) ₂] \cdot Cl ₂ (blue)	m P2 ₁ /c 2	6.32(2) 10.34(1) 14.69(1)	114.9(2)	CuO ₄ N ₂	O _{eq} N _{eq} H ₂ O _{ax}	79.5(3) ^c 89.2(3) 84.4(3)	168
[Cu(hist) ₂ (H ₂ O) ₂] \cdot (NO ₃) ₂ (blue violet)	tr P-1 1	5.4582(10) 7.1533(13) 13.8444(11)	98.617(16) 87.070(25) 109.830(6)	CuO ₄ N ₂	O _{eq} N _{eq} H ₂ O _{ax}	84.0(3) ^c 92.1(3,1) 83.5(3,3,5)	169
[Cu(phenO) ₂ (H ₂ O) ₂] \cdot (NO ₃) ₂ (not given)	m C2/c 4	15.724(2) 8.995(2) 18.432(3)	112.06(2)	CuO ₄ N ₂	O _{eq} N _{eq} H ₂ O _{ax}	86.0(2) ^d 91.6(3) 89.4(3)	170
[Cu(bzpy) ₂ (H ₂ O) ₂] \cdot (NO ₃) ₂ (green)	m P2 ₁ /c 2	7.872(1) 14.573(3) 11.132(1)	102.79(1)	CuO ₄ N ₂	O _{eq} N _{eq} H ₂ O _{ax}	90.0(1,3,2) 90.0(1,9,7) ^e 90.0(1,9)	171
[Cu(acatr) ₂ (H ₂ O) ₂] \cdot (SO ₄) \cdot 5H ₂ O (blue)	m P2 ₁ /a 4	24.834(6) 8.796(4) 9.751(4)	92.21(2)	CuO ₄ N ₂	O _{eq} N _{eq} H ₂ O _{ax}	88.8(1) ^d 89.8(1,3,7) 175.0(1) 177.0(1)	172
[Cu(Meal) ₂ (H ₂ O) ₂] \cdot 2H ₂ O (brown orange)	m P2 ₁ /c 4	8.439(8) 12.208(12) 15.37(2)	111.08(5)	CuO ₄ N ₂	H ₂ O _{eq} O _{eq} N _{ax}	88.03(8) 80.63(8, 4.77) ^e 2.040(2) \times 2 2.414(2) \times 2	173

[Cu(bpm)(ox)(H ₂ O) ₂].5H ₂ O (blue)	or P ₂ ,2,2-1 4	6.554(2) 13.838(4) 19.319(7)	CuO ₄ N ₂	bpmN _{eq} oxO _{eq} H ₂ O _{ax}	2.189(6, 13) 1.935(6, 1) 2.529(7, 26)	O _{eq} ,O _{eq} N _{eq} ,N _{eq} O _{eq} ,N _{eq} O _{eq} ,O _{ax} N _{eq} ,O _{ax}	85.5(1) ^f 81.5(3) ^e 96.5(3, 1.4) 93.1(3, 4.7) 87.1(3, 3.9)	174
Cu(Me ₆ en)(NO ₂) ₂ (green)	m P ₂ ,1/n 4	7.570(2) 12.506(2) 12.500(4)	CuO ₄ N ₂	N _{eq} O _{eq} O _{ax}	2.03(1, 1) 2.02(1, 2) 2.44(1, 3)	O _{eq} ,O _{eq} O _{eq} ,O _{ax} O _{ax} ,O _{ax} N _{eq} ,O _{ax} N _{eq} ,O _{eq} N _{eq} ,O _{eq}	89.0(6) 55.2(5, 1.0) ^f 86.4(5, 2) 127.1(6) 108.9(6, 4.5) 93.4(6, 1.1) 167.2(5, 7) 87.0(6) ^e	175
Cu(dmpz) ₂ (NO ₃) ₂ (not given)	tr P ₂ -1 2	8.708(3) 13.481(4) 9.717(3)	CuO ₄ N ₂	N _{eq} O _{eq} O _{ax}	1.955(-, 5) 2.005(-, 5) 2.46(-, 1)	O _{eq} ,N _{eq} O _{eq} ,O _{ax} O _{ax} ,O _{ax} O _{eq} ,O _{eq} N _{eq} ,O _{ax} N _{eq} ,O _{eq}	55.6(4, 7) ^f 84.6(4, 2.0) 126.4(4) 85.5(4) 96.3(4) 118.6(4, 1.2) 92.3(4, 5.7) 172.5(4, 1.3) 95.9(4)	176
Cu(hdmpz) ₂ (NO ₃) ₂ (blue)	m P ₂ ,1/c 2	7.8176(8) 14.93(2) 8.4076(8)	CuO ₄ N ₂	O _{eq} N _{eq} O _{ax}	1.991(2) × 2 1.974(2) × 2 2.561(2) × 2	N _{eq} ,N _{eq} O _{eq} ,N _{eq} O _{eq} ,O _{ax} N _{eq} ,O _{ax}	88.91(7) ^d 90.0(7, 1.09) 90.0(1, 7.04) 90.0(6, 1.81)	177
Cu(nphyro) ₂ (NO ₃) ₂ (dark blue)	m P ₂ ,1/n 2	7.768(2) 13.976(2) 9.045(3)	CuO ₄ N ₂	O _{eq} N _{eq} O ₂ NO _{ax}	1.951(3) × 2 1.963(3) × 2 2.410(3) × 2	O _{eq} ,N _{eq} O _{eq} ,N _{eq} O _{eq} ,O _{ax}	90.00(1, 6.8) ^f 90.00(1, 5) 90.00(1, 3.0)	178
[Cu(etapy) ₂ (NO ₃) ₂](NO ₃) ₂ (purple)	tr P ₂ -1 1	8.320(6) 8.639(7) 11.096(11)	CuO ₄ N ₂	N _{eq} O ₂ NO _{eq} O ₂ NO _{ax}	1.984(2) × 2 1.992(2) × 2 2.562(3) × 2	N _{eq} ,O _{eq} N _{eq} ,O _{ax} O _{eq} ,O _{ax}	88.95(9) 92.29(9) 54.85(8) ^f	179
Cu(amp) ₂ (ClO ₄) ₂ (not given)	m P ₂ ,1/a 2	9.893(1) 11.192(5) 8.367(1)	CuO ₄ N ₂	O _{eq} N _{eq} O ₃ ClO _{ax}	1.942(1) × 2 1.976(5) × 2 2.457(6) × 2	O _{eq} ,N _{eq} N _{eq} ,O _{ax} O _{eq} ,O _{ax}	84.8(2) ^e 84.0(2) 86.2(2)	180

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	<i>a</i> [Å] <i>b</i> [Å] <i>c</i> [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
Cu(pyact) ₂ (ClO ₄) ₂ (blue)	m C2/c 4	10.957(3) 12.178(6) 15.566(4)	108.74(2)	CuO ₄ N ₂	O _{eq} N _{eq} O ₃ ClO _{ax}	N _{eq} , O _{eq} N _{eq} , O _{ax} O _{eq} , O _{ax}	181
Cu(pzdc) ₂ (ClO ₄) ₂ (yellowish green)	P-1	10.959(4) 6.970(3) 7.666(3)	116.63(3) 78.14(5) 97.27(5)	CuO ₄ N ₂	O _{eq} N _{eq} O ₃ ClO _{ax}	N _{eq} , O _{eq} N _{eq} , O _{ax} O _{eq} , O _{ax}	182
Cu(mpo) ₂ (ClO ₄) ₂ (purple)	or Pbca 4	13.0888(5) 14.7638(6) 10.3534(4)		CuO ₄ N ₂	O _{eq} N _{eq} O ₃ ClO _{ax}	O _{eq} , N _{eq} N _{eq} , O _{ax} O _{eq} , O _{ax}	183
[Cu(damp) ₂ (ClO ₄) ₂ ·2H ₂ O (blue)]	m P2 ₁ /c 2	7.913(5) 11.214(5) 9.756(6)	94.86(2)	CuO ₄ N ₂	O _{eq} N _{eq} O ₃ ClO _{ax}	O _{eq} , N _{eq} N _{eq} , O _{ax} N _{eq} , O _{ax}	184
Cu(2-NO ₂ bz) ₂ (py) ₂ (not given)	tr P-1 1	8.380(8) 5.773(5) 13.095(8)	103.98(6) 91.06(7) 103.99(7)	CuO ₄ N ₂	O _{eq} N _{eq} O _{ax}	O _{eq} , N _{eq} O _{eq} , O _{ax} O _{ax} , N _{eq} O _{eq} , O _{ax}	185
Cu(tpth) ₂ (py) ₂ (green)	or Pccn 4	9.476(4) 12.455(5) 22.401(9)		CuO ₄ N ₂	O _{eq} pyN _{eq} O _{ax}	O _{eq} , N _{eq} O _{eq} , N _{eq} N _{eq} , O _{ax}	186
Cu(monp) ₂ (4-Mepy) ₂ ^c (not given)	tr P-1 2	10.537(1) 10.521(1) 15.377(2)	121.031(5) 110.958(5) 86.105(3)	CuO ₄ N ₂	pyN _{eq} O _{eq} O _{ax} pyN _{eq} O _{eq} O _{ax}	O _{eq} , N _{eq} O _{eq} , O _{ax} O _{ax} , O _{eq} O _{eq} , N _{eq} O _{eq} , N _{eq} O _{ax} , O _{eq}	187
Cu(acsal) ₂ (py) ₂ (purple)	m P2 ₁ /n 2	17.823(5) 10.903(4) 6.598(2)	95.74(2)	CuO ₄ N ₂	O _{eq} pyN _{eq} O _{ax}	O _{eq} , N _{eq} O _{eq} , O _{ax} N _{eq} , O _{ax}	188

Cu(biac) ₂ (py) ₂ (purple)	tr P-1 1	8.413(2) 8.370(3) 10.988(3)	98.14(3) 84.43(2) 119.62(2)	CuO ₄ N ₂	O _{eq} pyN _{eq} O _{ax}	2.012(3) × 2 1.993(3) × 2 2.510(5) × 2	O _{eq} N _{eq} O _{eq} O _{ax} N _{eq} O _{ax}	91.4(1) 57.5(1) ^f 89.7(2)	189
Cu(Ph ₃ ac) ₂ (py) ₂ (blue green)	m P2 ₁ /c 2	8.931(1) 12.602(1) 19.578(1)	118.17(40)	CuO ₄ N ₂	pyN _{eq} O _{eq} O _{ax}	2.015(2) × 2 1.964(1) × 2 2.673(2) × 2	N _{eq} O _{eq} N _{eq} O _{ax} O _{eq} O _{ax}	91.51(6) 80.99(6) 54.15(6) ^f	190
Cu(monp) ₂ (py) ₂ (dark green)	or Pbca 8	37.54(2) 16.76(1) 7.947(5)		CuO ₄ N ₂	O _{eq} pyN _{eq} O _{ax}	1.94(2, 1) 2.07(2, 2) 2.52(2, 6)	N _{eq} N _{eq} O _{eq} O _{eq} O _{ax} O _{ax}	178.6(2) 177.5(2) 178.1(2)	191
Cu(F ₆ acac) ₂ (py) ₂ (green)	m P2 ₁ /c 4	15.7643(4) 10.3651(3) 16.8517(6)	99.37(1)	CuO ₄ N ₂	pyN _{eq} O _{eq} O _{ax}	2.012(7, 1) 1.998(6, 1) 2.263(6, 37)	O _{eq} O _{ax} O _{ax} O _{ax} N _{eq} N _{eq}	87.5(3, 2.0) ^d 167.4(3) 93.0(3)	192
[Cu(MeOac) ₂ (py) ₂ ·4H ₂ O] (blue)	tr P-1 1	8.55(2) 9.25(2) 7.50(2)	97.5(4) 70.1(4) 86.4(4)	CuO ₄ N ₂	O _{eq} pyN _{eq} O _{ax}	1.94(1) × 2 2.01(1) × 2 2.36(1) × 2	N _{eq} O _{eq} O _{eq} O _{ax} N _{eq} O _{ax}	80.1(3) 75.5(3) ^c 91.2(3)	193
Cu(Cl ₂ ac) ₂ (2-Mepy) ₂ (blue)	m P2 ₁ /n 4	10.511(15) 16.091(23) 12.275(17)	93°24'(6')	CuO ₄ N ₂	pyN _{eq} O _{eq} O _{ax}	2.011(2, 21) 2.007(7, 8) 2.493(9) 2.711(10)	O _{eq} O _{eq} N _{eq} N _{eq} O _{eq} O _{ax} O _{eq} N _{eq} O _{ax} N _{eq}	158.3(4) 169.8(5) 55.3(3, 2.6) ^f 91.0(4, 2.5) 90.7(4, 6.5)	194
Cu(Clac) ₂ (2-Mepy) ₂ (purple)	tr P1 1	8.166(10) 9.231(12) 7.818(9)	118°25'(5') 107°40'(5') 100°26'(5')	CuO ₄ N ₂	pyN _{eq} O _{eq} O _{ax}	1.989(6) × 2 1.975(6) × 2 2.707(7) × 2	N _{eq} N _{eq} N _{eq} O _{ax} O _{eq} O _{ax}	90.5(2) 91.4(2) 53.9(2) ^f	195
Cu(ttbt) ₂ (4-Mepy) ₂ (not given)	m C2/c 4	9.357(3) 17.784(5) 18.138(5)	95.4(4)	CuO ₄ N ₂	O _{eq} pyN _{eq} O _{ax}	2.066(6) × 2 2.174(5) × 2 2.125(5) × 2	O _{eq} O _{ax} O _{eq} N _{eq} N _{eq} N _{eq} N _{eq} O _{ax}	90.3(-, 1.5) ^d 91.4(-, 2.1) 94.9 90.8	197
Cu(ac) ₂ (2, 6-Me ₂ py) ₂ (not given)	tr P-1 1	8.424(3) 8.456(2)	60.0(2) 64.41(2)	CuO ₄ N ₂	O _{eq} pyN _{eq}	1.997(4) × 2 2.046(3) × 2	N _{eq} O _{eq} N _{eq} O _{ax}	90.9(1) 91.3(1)	198

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	O_{ax} O_{eq} N_{eq} O_{ax} O_{eq} N_{eq} O_{ax} O_{eq} N_{eq} O_{ax}	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(ac) ₂ (3,4-Me ₂ py) ₂ ·H ₂ O (not given)]	1 m C2/c 4	8.736(2) 14.442(12) 12.578(11) 10.817(5)	62.34(2) 93.49(6)	CuO ₄ N ₂	O_{ax} O_{eq} N_{eq} O_{ax}	2.711(9) × 2 1.953(3) × 2 1.998(3) × 2 2.810(2) × 2	127.0(2) 84.60(8) 90.00(1, 1.30) 89.55(8, 0)	199
[Cu(pr) ₂ (3,4-Me ₂ py) ₂ ·H ₂ O (not given)]	m C2/c 4	15.024(1) 13.498(1) 11.086(1)	97.433(9)	CuO ₄ N ₂	O_{eq} N_{eq} O_{ax}	1.952(5) × 2 2.003(5) × 2 2.960(3) × 2	89.95(20, 7.5) 92.7(2) 83.2(2)	199
Cu(ac) ₂ (2,4,6-Me ₃ py) ₂ (violet)	m P2 ₁ /c 2	7.842(1) 16.384(2) 8.320(2)	101.91(1)	CuO ₄ N ₂	O_{eq} N_{eq} O_{ax}	1.963(2) × 2 2.042(2) × 2 2.788(3) × 2	90.00(8, 33) 52.02(9) ^f	200
Cu(ac) ₂ (amb) ₂ ^c (not given)	m P2 ₁ /n 4	15.302(8) 17.373(6) 18.449(10)	103.06(4)	CuO ₄ N ₂	O_{eq} N_{eq} O_{ax}	1.99(2, 1) 2.01(2, 2) 2.66(2, 75)	154.2(-1.7) 142.2	201
Cu(ac) ₂ (ahb) ₂ (not given)	tr P-1 2	9.782(5) 13.557(7) 14.188(6)	100.70(4) 89.90(4) 110.56(5)	CuO ₄ N ₂	O_{eq} N_{eq} O_{ax}	2.00(2) × 2 2.03(2, 3) 2.61(2, 4)	153.2(-1.8) 137.6	201
Cu(fm) ₂ (2-Mebzim) ₂ (purple)	m P2 ₁ /n 2	6.3306(5) 13.4855(11) 11.1863(5)	105.471(5)	CuO ₄ N ₂	fmO_{eq} fmN_{eq} $bzimO_{ax}$	2.07(1, 1) 1.95(1, 3) 2.57(1, 8)	158.2(-, 1) 136.6	202
Cu(ac) ₂ (im) ₂ (violet)	or Pccn 4	11.660(3) 12.284(4) 9.375(3)		CuO ₄ N ₂	imN_{eq} O_{eq} O_{ax}	1.985(1) × 2 1.971(1) × 2 2.668(1) × 2	54.2(1) ^f 90.0(1)	203
Cu(tgly) ₂ (Meim) ₂ (blue violet)	m P2 ₁ /n 2	13.733(3) 8.766(2) 13.641(3)	105.97(5)	CuO ₄ N ₂	imN_{eq} O_{eq} O_{ax}	2.004(8) × 2 1.921(9) × 2 2.799(8) × 2	89.97(35) 52.21(28) ^f 85.94(32)	204

[Cu(ac) ₂ (mon) ₂] ₂ ·2H ₂ O (not given)	m P2 ₁ /c 2	9.60(1) 15.93(1) 6.28(1)	113.9(1)	CuO ₄ N ₂	morN _{eq} acO _{eq} acO _{ax}	2.035(4) × 2 1.958(4) × 2 2.738(4) × 2	O _{ax} O _{eq} N _{eq} O _{ax} N _{eq} O _{eq}	52.7(1) ^f 89.6(3) 100.6(3)	205
Cu(sal) ₂ (pym) ₂ (blue)	m C2/c 4	15.786(1) 13.299(2) 12.083(1)	97.22(1)	CuO ₄ N ₂	salO _{eq} pymN _{eq} pymO _{ax}	1.981(2) × 2 1.981(1) × 2 2.331(1) × 2	N _{eq} O _{ax} N _{eq} O _{eq} O _{eq} O _{ax}	77.62(7) ^f 90.06(6) 95.13(5)	206
Cu(sal) ₂ (nc) ₂ (violet)	or Pbca 4	14.625(7) 9.572(5) 18.671(9)		CuO ₄ N ₂	ncN _{eq} salO _{eq} salO _{ax}	1.944(7) × 2 1.935(6) × 2 2.609(6) × 2	O _{eq} N _{eq} O _{eq} O _{ax} N _{eq} O _{ax}	89.4(2) 55.2(2) ^f 88.2(2)	207
Cu(pppa) ₂ (CF ₃ SO ₃) ₂ ^c (green)	tr P1 3	10.648(9) 14.080(7) 18.294(10)	81.86(4) 77.06(6) 71.21(5)	CuO ₄ N ₂	O _{eq} N _{eq} O ₂ SO _{ax}	1.926(4) × 2 2.069(6) × 2 2.356(5) × 2	O _{eq} O _{ax} O _{eq} N _{eq} O _{ax} O _{ax}	90.3(2, 4, 7) 88.2(2, 1.0) ^d 180 92.8(2)	208
Cu(2-Mepy) ₂ (NO ₃) ₂ (not given)	m P2 ₁ /c 4	8.31(3) 14.81(3) 14.14(3)	123.9(2)	CuO ₄ N ₂	ncN _{eq} O _{eq} N _{eq} O ₂ SO _{ax}	1.917(4, 2) 2.017(4, 2) 2.633(5, 167)	N _{eq} O _{eq} O _{ax} O _{ax} N _{eq} N _{eq} O _{ax} O _{ax}	174.4(2) 90.0(2, 10.9) 90.0(2, 4.2) 174.2(2)	209
Cu(2-Mepy) ₂ (NO ₃) ₂ (not given)	m P2 ₁ /c 4	8.57(3) 14.39(3) 14.20(3)	119.5(2)	CuO ₄ N ₂	pyN _{eq} O _{eq} O _{ax}	1.985(14, 15) 1.999(16, 28) 2.398(14, 91)	N _{eq} N _{eq} O _{eq} O _{ax} O _{ax} O _{ax}	53.9(5, 2.2) ^f 142.9(5, 2.3) 89.1(5) 91.1(5, 3.0) 163.2(5) 89.8(6, 8) 176.9(5)	209
					pyN _{eq} O _{eq} O _{ax}	2.000(10, 1) 1.994(10, 11) 2.534(8, 17)	O _{eq} O _{ax} O _{ax} O _{ax} O _{ax} N _{eq} O _{eq} O _{eq} N _{eq} N _{eq}	55.1(4, 2) ^f 129.8(4, 3) 74.8(4) 93.4(4, 1.9) 175.1(1) 89.8(5, 1.4) 171.5(4)	

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
Cu(F ₆ acac) ₂ (4-Mepz) ₂ (light green)	tr	9.301(3)	104.71(2)	CuO ₄ N ₂	1.988(3) × 2	O _{ax} O _{eq}	210
	P-1	10.815(2)	101.38(2)		1.975(3) × 2	N _{eq} O _{ax}	
	1	6.457(1)	75.41(2)		2.346(3) × 2	O _{eq} N _{eq}	
Cu(hip) ₂ (im) ₂ (violet)	m	10.4863(4)	102.234(1)	CuO ₄ N ₂	1.964(3) × 2	O _{eq} N _{eq}	211
	P2 ₁ /c	11.4758(3)		hipO _{eq}	1.976(3) × 2		
	2	10.3410(4)		hipO _{ax}	2.736(4) × 2		
[Cu(phen)(NO ₃ (H ₂ O)) ₂ ·NO ₃ (pale blue)]	m	7.03(1)	109.9(1)	CuO ₄ N ₂	2.007(6, 2)	N _{eq} N _{eq}	212
	P2 ₁ /c	20.20(1)		H ₂ O _{eq}	2.005(6)	N _{eq} O _{eq}	
	4	11.49(1)		O ₂ NO _{eq}	1.983(5)		
				H ₂ O _{ax}	2.261(6)	N _{eq} O _{ax}	
				O ₂ NO _{ax}	2.594(6)	O _{eq} O _{ax}	
Cu(mn) ₂ (Me ₂ CO) ₂ (dark brown)	m	8.001(5)	98.0(1)	CuO ₄ N ₂	1.988(5) × 2	O _{eq} O _{eq}	213
	P2 ₁ /c	13.806(11)		mnN _{eq}	1.951(4) × 2	N _{eq} O _{eq}	
	2	11.012(18)		mnO _{eq}	2.651(6) × 2	N _{eq} O _{ax}	
Cu(NO ₂ -bz) ₂ (ampr) ₂ (green)	or	10.933(3)		CuO ₄ N ₂	2.000(4, 10)	N _{eq} O _{eq}	214
	Pna2 ₁	23.188(9)		N _{eq}	2.018(3) × 2	89.8(1, 14.2)	
	4	9.921(2)		O _{eq}	2.457(4, 28)	176.5(1)	
Cu(tbd) ₂ (qu) ₂ (not given)	tr	9.782(2)	91.62(2)	CuO ₄ N ₂	1.924(1) × 4	O _{ax} O _{ax}	215
	P-1	9.978(2)	111.09(1)		2.545(3) × 2	O _{eq} O _{eq}	
	1	9.562(2)	104.75(2)	quN _{ax}		O _{eq} N _{ax}	
Cu(Clac) ₂ (Me ₆ en) ^c (dark blue)	or	13.013(3)		CuO ₄ N ₂	2.059(6) × 2	N _{eq} N _{eq}	216
	Pccn	19.182(12)		N _{eq}	1.970(6) × 2	O _{eq} O _{eq}	
	8	15.372(5)		O _{ax}	2.645(5) × 2	O _{eq} N _{eq}	

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(monp) ₂ (Me ₄ en)]·2H ₂ O (dark brown)	or Pbcn 4	16.160(3) 18.538(3) 8.263(1)		CuO ₄ N ₂	1.932(3) × 2 2.049(4) × 2 2.565(3) × 2	O _{eq} O _{ax} O _{eq} N _{eq} N _{eq} O _{ax} O _{eq} O _{eq} O _{ax} O _{ax} N _{eq} N _{eq}	220a
Cu(2,6-dmp) ₂ (Me ₄ en) ^c (green)	m P2 ₁ n 4	14.780(3) 10.984(2) 15.490(3)	116.63(3)	CuO ₄ N ₂	2.038(2) × 2 1.946(1) × 2 2.432(2) × 2	O _{eq} O _{ax} N _{eq} O _{ax} N _{eq} O _{eq} O _{eq} O _{eq} O _{ax} O _{ax} O _{eq} O _{ax} N _{eq} O _{ax} N _{eq} O _{eq}	220b
[Cu(fmpn) ₂ (Me ₄ en)]·4H ₂ O (not given)	m P2 ₁ c 4	8.308(3) 15.0715(6) 21.5856(9)	97.158(3)	CuO ₄ N ₂	2.024(2) × 2 1.938(2) × 2 2.436(2) × 2	enN _{eq} phO _{eq} phO _{ax} enN _{eq} O _{eq} O _{ax}	221

Cu(F ₄ acac) ₂ (Me ₄ pim) (orange)	tr P-1 2	9.317(4) 12.922(5) 13.525(6)	95.24(1) 109.71(1) 110.92(1)	CuO ₄ N ₂	N _{eq} acO _{eq} acO _{ax}	2.012(4, 9)	O _{ax} , N _{eq}	94.8(2, 2, 7)	222
						1.962(4, 6)	N _{eq} , O _{eq}	90.9(2)	
						2.288(4, 5)	N _{eq} , N _{eq}	174.2(2, 2, 8)	
[Cu(Me ₂ amet) ₂ (bz)] ₂ ·bz (blue)	or P ₂ , 2, 2, 1 4	9.166(3) 13.293(7) 19.274(7)	2.059(6, 1)	CuO ₄ N ₂	N _{eq} O _{eq} bzO _{eq} O _{ax} bzO _{ax}	2.168(6)	O _{ax} , O _{eq}	57.1(2) ^f	223
						2.126(6)	O _{ax} , N _{eq}	101.5(2, 8, 7)	
						2.179(6)	O _{ax} , O _{ax}	81.3(2) ^e	
						2.376(6)	O _{ax} , O _{ax}	93.4(2, 5, 8)	
							O _{eq} , N _{eq}	158.8(2)	
Cu(NO ₂) ₂ (Et ₂ en) (not given)	m P ₂ , 1/c 4	11.939(12) 7.699(4) 12.561(10)	99.75(9)	CuO ₄ N ₂	N _{eq} O _{eq} O _{ax}	2.026(9, 29)	O _{eq} , O _{eq}	87.8(3)	224
						2.007(8, 6)	N _{eq} , O _{ax}	115.8(4, 2)	
						2.452(11, 28)	O _{eq} , O _{ax}	54.8(3, 3) ^f	
							O _{eq} , N _{eq}	90.6(3, 3)	
							N _{eq} , N _{eq}	170.5(4, 7)	
Cu(NO ₂) ₂ (Et ₂ en) (not given)	or Pbcn 4	17.722(8) 8.126(2) 8.104(2)	1.994(7) × 2 2.235(10) × 2 2.509(12) × 2	CuO ₄ N ₂	N _{eq} O _{eq} O _{ax}	86.3(3) ^e	N _{eq} , N _{eq}	86.3(3) ^e	225
						176.1(3)	O _{ax} , O _{ax}	176.1(3)	
						88.5(3) ^e	N _{eq} , N _{eq}	88.5(3) ^e	
						47.4(3) ^f	O _{eq} , O _{ax}	47.4(3) ^f	
						86.6(3)	N _{eq} , O _{eq}	86.6(3)	
Cu(NO ₂) ₂ (Me ₄ en) (dark green)	m P ₂ , 1/n 4	7.570(2) 12.506(2) 12.500(4)	2.01(1, 2) 2.03(2, 1) 2.43(1, 4)	CuO ₄ N ₂	O _{eq} N _{eq} O _{ax}	87.0(6) ^e	N _{eq} , N _{eq}	87.0(6) ^e	226
						108.9(6, 4, 5)	N _{eq} , O _{ax}	108.9(6, 4, 5)	
						93.4(6, 1, 1)	N _{eq} , O _{eq}	93.4(6, 1, 1)	
						167.2(5, 7)	N _{eq} , O _{eq}	167.2(5, 7)	
						55.2(5, 1, 0) ^f	O _{eq} , O _{ax}	55.2(5, 1, 0) ^f	
Cu(NO ₃) ₂ (Me ₄ en) (blue)	m C ₂ /c 4	13.8971 7.4956 12.2960	2.003(2) × 2 2.007(2) × 2 2.440(2) × 2	CuO ₄ N ₂	N _{eq} (NO ₃)O _{eq} (NO ₃)O _{ax}	89.0(6)	O _{eq} , O _{eq}	89.0(6)	227
						88.0(1) ^e	N _{eq} , N _{eq}	88.0(1) ^e	
						56.9(1) ^f	O _{eq} , O _{ax}	56.9(1) ^f	
						99.8(1, 6, 3)	O _{eq} , N _{eq}	99.8(1, 6, 3)	
							O _{eq} , O _{ax}		

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å]			α [°]			Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
		b	c	β	β	γ					
Cu(NO ₃) ₂ (Me.en) (not given)	or Pnma 4	11.724(4)						CuO ₄ N ₂	2.022(5) × 2	O _{eq} , O _{eq}	228
		15.725(6)							N _{eq} (NO ₃)O _{eq} (NO ₃)O _{ax}	113.1(2), 12.0)	
Cu(NO ₃) ₂ (daco) (not given)	m P2 ₁ /c 4	8.56(2)						CuO ₄ N ₂	1.977(5, 2)	N _{eq} , N _{eq}	229
		10.39(5)			78.52(3)				N _{eq} (NO ₃)O _{eq} (NO ₃)O _{ax}	86.9(2) ^d 92.6(2, 3) 168.9(2, 1.5)	
Cu(NO ₃) ₂ (Cl ₂ aq) (not given)	m C2/c 8	20.901(5)						CuO ₄ N ₂	1.983(4, 20)	N _{eq} , N _{eq}	230
		7.439(2)			114.59(3)				O _{eq} O _{ax}	83.5(1) ^f 94.1(1, 4) 172.3(1, 2.9)	
[Cu(bimm) ₂ (NO ₃)]NO ₃ ·H ₂ O (green)	tr P-1 2	8.562(4)						CuO ₄ N ₂	2.045(1)	O _{eq} , O _{ax}	231
		9.990(4)			90.16(2)				N _{eq} (NO ₃)O _{eq} O _{ax}	55.3(2) ^f 101.6(3, 10.2)	
		13.546(1)			100.86(2)				2.055(1)	O _{eq} , O _{eq}	
					118.29(4)				2.285(1)	O _{ax} , N _{eq}	
									2.501(1)	(NO ₃)O _{ax}	
									90.7(3, 10.0)		

Cu(C ₁₀ H ₁₀ NO ₄) ₂ (deep blue)	m P2 ₁ /n 2	9.141(2) 7.335(3) 11.112(3)	103.87(2)	CuO ₄ N ₂	O _{ax} , O _{ax}	146.4(2)	232
					O _{ax} , N _{eq}	77.2(2) ^e	
					N _{eq} , N _{eq}	91.6(3, 14.4) 166.0(3)	
Cu(degly) ₂ (blue)	m P2 ₁ /b 2	9.73(2) 7.07(2) 12.22(3)	111	CuO ₄ N ₂	O _{eq}	1.949(2) × 2	233
					O _{eq}	1.991(3) × 2	
					O _{ax}	2.582(3) × 2	
[Cu(bomgly) ₂] ₂ ·2H ₂ O (pale blue)	m P2 ₁ /c 2	6.845(1) 14.095(4) 9.427(3)	91.95(2)	CuO ₄ N ₂	N _{eq}	2.14	234
					O _{eq}	1.90	
					O _{ax}	2.41	
[Cu(npac) ₂] ₂ ·2H ₂ O (not given)	m P2 ₁ /c 2	8.636(2) 14.361(4) 9.247(2)	94.78(2)	CuO ₄ N ₂	O _{eq} , O _{eq}	1.954(2) × 2	235
					O _{eq} , N _{eq}	2.062(2) × 2	
					N _{eq} , O _{ax}	2.321(2) × 2	
[Cu(Hpydca) ₂] ₂ ·3H ₂ O ^c (blue)	m Pc 4	14.74(1) 10.30(1) 13.77(1)	123.9(1)	CuO ₄ N ₂	O _{eq} , N _{eq}	2.071(2) × 2	236
					O _{eq}	1.939(2) × 2	
					O _{ax}	2.416(2) × 2	
					O _{ax} , N _{eq}	100.5(3, 6) 79.6(4, 3) ^e 103.8(4, 1.2) 77.1(4) ^e 92.5(3, 3.0) 75.4(3) ^e 152.4(3) 179.1(4) 159.1(3) 80.0(4, 6) ^e 108.0(4, 2) 107.1(4) 75.0(4, 1.0) ^e 92.6(4, 8) 150.0(4) 176.9(4) 160.0(4)	

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	<i>a</i> [Å] <i>b</i> [Å] <i>c</i> [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(purp) ₂] ₂ ·7.6H ₂ O (not given)	tr P-1 2	12.18(4) 11.889(5) 10.447(3)	73.57(3) 83.72(3) 75.09(2)	CuO ₄ N ₂	1.997(6, 24) 2.026(7, 3) 2.296(5, 7)	81.9(2) ^e 90.0(3, 8.1) 179.0(7) 76.8(2, 2) ^e 103.3(2, 1.2) 164.2(2) 91.8(2, 1.3) 153.5(2)	237
K ₂ [Cu(idac) ₂] ₂ ·H ₂ O (not given)	m B2/b 4	15.53(3) 13.17(3) 6.28(3)	95.5(5)	CuO ₄ N ₂	1.97(2) × 2 1.95(2) × 2 2.43(2) × 2	not given	238
K ₂ [Cu(pydcacH) ₂] ₂ ·7H ₂ O (not given)	or Pnna 4	20.823(5) 13.398(5) 8.183(3)		CuO ₄ N ₂	2.189(5, 29) 1.936(5)	O _{eq} , O _{eq} O _{eq} , O _{ax} O _{ax} , O _{ax}	239
[Cu(pydcac)(pydcacH)] ₂ ·H ₂ O ^c (not given)	m Pc 4	14.76(9) 10.27(3) 13.76(8)	124.0(1)	CuO ₄ N ₂	2.06(1, 1) 2.315(1, 50) 1.93(1, 5)	87.9(3, 2.5) 154.7(3) 77.6(2, 3) ^e 173.7(3)	240
[Cu(mdac) ₂] ₂ ·ClO ₄ ₂ (not given)	m P2 ₁ /c 2	6.183(2) 17.617(4) 8.900(2)	110.2(1)	CuO ₄ N ₂	2.02(1, 1) 2.375(2, 75) 1.97(1, 5)	77.5(5, 4, 3) ^e 177.1(5) 92.5(9, 3, 0) 155.0(5, 5, 5)	241
[Ba(idac) ₂ (H ₂ O) ₁₀][Cu(idac) ₂] (blue)	tr P-1 1	9.050(4) 9.437(5) 11.468(5)	108.05(3) 93.68(4) 112.55(4)	CuO ₄ N ₂	1.952(2) × 2 2.021(2) × 2 2.471(3) × 2	85.5(1) ^e 91.1(1) 71.7(1) ^e	242
[Cu(odac)(bpy)(H ₂ O)] ₂ ·4H ₂ O (light blue)	tr P1 2	14.069(9) 8.898(5) 11.078(7)	114.1(1) 52.0(2) 118.6(1)	CuO ₄ N ₂	2.018(4, 6) 1.963(3, 9) 2.458(4)	93.2(7, 1, 0) 172.7(1, 2, 1) 97.8(1, 7, 5)	243

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
Cu(tocop) (black)	m	7.032(5)	113.46(4)	CuO ₄ N ₂	2.050(3) × 2	100.9(1)	250
	P2 ₁ /c	15.533(2)			N _{eq} , O _{ax}	83.9(1)	
Cu(mug) (blue)	2	9.836(2)	106.4(1)	CuO ₄ N ₂	2.615(3) × 2	87.3(1)	251
	m	6.069(1)			O _{ax} , O _{eq}	81.1(1, 7.8) ^c	
	P2 ₁	13.617(1)			O _{eq} , O _{ax}	103.6(1, 9.0)	
	2	9.005(1)			O _{eq} , O _{ax} O _{eq} , N _{eq}	86.2(2) 84.8(2) ^e 99.6(2) ^f 168.7(2, 1)	
Cu(dzco) (not given)	tr	7.608(3)	109.51(3) 93.97(3) 81.32(3)	CuO ₄ N ₂	2.045(1) × 2	84.37(6) ^e	252
	P-1	9.064(2)			N _{eq} , O _{eq}	101.60(4)	
	1	7.000(2)			O _{ax}	86.92(5)	
[Cu(ecgm)] ₃ ·3/2H ₂ O (blue)	or	8.962(2)	109.51(3) 93.97(3) 81.32(3)	CuO ₄ N ₂	1.954(3, 8)	96.8(1, 8.3)	253
	P2 ₁ 2 ₁ 2 ₁	11.595(2)			O _{eq} , O _{ax}	106.3(1)	
	4	31.491(2)			N _{eq} , O _{eq}	78.1(1) ^e	
					O _{ax} , N _{eq}	85.2(1, 7.1)	
[Cu(megly)] ₂ ·2H ₂ O (not given)	or	13.451(2)	109.51(3) 93.97(3) 81.32(3)	CuO ₄ N ₂	1.963(6, 9)	83.8(1) ^e	254
	Pca2 ₁	9.166(1)			N _{eq} , N _{eq}	166.9(1)	
	4	13.138(2)			O _{eq} , O _{eq}	88.5(1) ^e	
					O _{ax} , O _{ax}	108.0(3)	
					89.7(3, 11.3)		
					2.021(6, 16)		
					2.343(6, 45)		
						161.4(3, 3.5)	
						171.1(2)	
						79.5(2, 6) ^f	
						101.6(3, 5.4)	
						86.8(3) ^e	

[Cu(ecgl)] ₂ ·2H ₂ O (not given)	or Pca2 ₁ 4	13.745(1) 9.301(2) 13.275(4)	CuO ₄ N ₂	O _{eq} N _{eq} O _{ax}	1.954(9, 23) 2.032(12, 15) 2.368(10, 40)	O _{eq} O _{eq} O _{eq} O _{ax} O _{eq} N _{eq} O _{ax} O _{ax} O _{ax} N _{eq} N _{eq} N _{eq}	106.7(4) 90.3(4, 10.4) 84.8(4, 1.9) ^e 162.7(5, 3.3) 171.1(2) 79.5(2, 6) ^e 101.6(3, 5.4) 87.0(4) ^e	254
Na ₂ [Cu(enphgly)]·5.5H ₂ O (dark blue)	or Pbcn 8	14.195(1) 22.607(3) 14.242(1)	CuO ₄ N ₂	O _{eq} N _{eq} O _{ax}	1.941(3, 4) 2.005(3, 6) 2.433(3, 29)		not given	255
K ₂ [Cu(esc)]·3H ₂ O (not given)	or Pbca 8	12.93(2) 22.90(3) 11.45(2)	CuO ₄ N ₂	O _{eq} N _{eq} O _{ax}	1.978(16, 45) 2.095(17, 1) 2.302(16, 78)	O _{eq} N _{eq} N _{eq} O _{ax} N _{eq} N _{eq} O _{eq} O _{eq} O _{eq} O _{ax}	86.0(-, 1.1) ^e 79.7(-, 2.0) ^e 100.0(-, 2.2) 89.0 ^e 103.3 90.2(-10.2)	256
Cu(4-NO ₂ pyNO) ₂ (H ₂ O) ₂ Cl ₂ (golden brown)	m P2 ₁ /c 2	13.540(6) 5.487(2) 11.153(5)	CuO ₄ Cl ₂	H ₂ O _{eq} Cl _{eq} O _{ax}	1.950(3) × 2 2.266(1) × 2 2.635(2) × 2	Cl _{eq} O _{eq} Cl _{eq} O _{ax} O _{eq} O _{ax}	91.0(1) 88.8(1) 94.9(1)	257
Cu(biu) ₂ Cl ₂ (blue green)	or Pnmm Pnn2 2	6.776(6) 8.025(12) 10.734(12)	CuO ₄ Cl ₂	O _{eq} Cl _{ax}	1.935(3) × 2 2.960(1) × 2	O _{eq} O _{eq} O _{eq} Cl _{ax}	90.6 ^d 94.8	258
Cu(12-aneO ₄) ₂ Cl ₂ (deep green)	or P2 ₁ 2 ₁ 2 ₁ 4	7.062(3) 13.661(8) 12.337(7)	CuO ₄ Cl ₂	O _{eq} Cl _{eq} O _{ax}	2.121(3, 8) 2.221(2, 7) 2.373(4, 30)	Cl _{eq} Cl _{eq} Cl _{eq} O _{ax} Cl _{eq} O _{eq}	95.85(7) 104.72(9, 5.08) 90.05(1, 1.05) 173.65(1, 1.15) 73.7(2, 3.0) ^e 135.2(1) 84.2(1)	259
Cu(Etsac) ₂ (H ₂ O) ₂ (deep green)	tr P-1 1	7.554(1) 9.001(1) 5.345(1)	CuO ₄ S ₂	O _{eq} S _{eq} H ₂ O _{ax}	1.916(2) × 2 2.381(1) × 2 2.457(1) × 2	S _{eq} O _{eq} S _{eq} O _{ax} O _{eq} O _{ax}	86.49(1) ^e 96.29(1) 88.97(5)	260

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
Cu(NO ₃) ₂ (dtp) ₂ (red)	or	11.55(1)		CuO ₄ NS	N _{eq}	93.3(1) ^d	261
	Pca2 ₁	7.877(2)			(NO ₃)O _{eq}	94.0(1, 5.4)	
	4	16.866(5)			S _{eq} O _{eq}	164.7(1)	
Cu(bheg)(Br)(H ₂ O) (blue)	or	8.7672(9)		CuO ₄ NBr	N _{eq} O _{ax}	169.1(1)	262
	P2 ₁ 2 ₁ -1	15.594(3)			O _{eq} O _{eq}	89.0(2)	
	4	7.4996(5)			N _{eq} O _{ax}	103.0(1, 9.3)	
					O _{eq} O _{ax}	56.5(2, 3) ^f	
					O _{ax} O _{ax}	93.8(2, 1.9)	
					S _{eq} O _{ax}	140.1(1)	
					S _{eq} O _{ax}	109.0(1)	
					Br _{eq} O _{ax}	103.8(3, 3.2)	
					Br _{eq} O _{eq}	91.1(4, 1)	
					Br _{eq} N _{eq}	174.8(4)	
Cu(cyclam)(BH ₄) ₂ (not given)	m	7.374(1)		CuN ₄ H ₂	O _{ax} O _{ax}	93.1(5, 3.3)	263
	P2 ₁ /c	12.660(2)	111.09(1)		O _{eq} O _{eq}	79.8(5)	
	2	8.858(1)			O _{ax} N _{eq}	172.0(5)	
					O _{ax} N _{eq}	76.8(4, 2.5) ^f	
					O _{eq} N _{eq}	83.9(6) ^e	
					N _{eq} N _{eq}	93.9(6)	
					N _{eq} N _{eq}	85.4(2) ^f	
					H ₃ BH _{ax}	94.6(2) ^d	
					N _{eq}	86.6(3) ^f	
					H ₃ BH _{ax}	93.4(3) ^d	
Cu(en) ₂ (BF ₄) ₂ (violet)	tr	7.42(2)	100°54'	CuN ₄ F ₂	N _{eq} N _{eq}	86.4(5) ^e	264
	P-1	8.22(2)	105°12'		N _{eq} N _{eq}	86.4(5) ^e	
	1	5.92(2)	160°0'		N _{eq} F _{ax}	93.9(5, 8)	
					N _{eq} F _{ax}	2.56(1) × 2	

Cu(trien)(PF ₆) ₂ (dark blue)	or B ₂ mb 4	7.362(1) 14.708(5) 15.551(6)	CuN ₄ F ₂	N _{eq} F _{ax}	1.982 2.596(-, 6)	not given	265
Cu(Bu ₂ -cyclam)(PF ₆) (pink)	tr P-1 1	9.357(2) 9.297(2) 8.552(2)	CuN ₄ F ₂	N _{eq} F _{ax}	2.034(2, 29) 2.840(2) × 2	86.8(8) ^e	266
[Cu(en) ₂ (PF ₆)(H ₂ O)]PF ₆ (dark blue)	or B ₂ mb 4	8.66036(1) 14.51489(5) 13.25635(6)	CuN ₄ FO	N _{eq} F _{ax} H ₂ O _{ax}	2.003(-, 28) 2.7371 2.3968	84.35 ^f	267
Cu(NH ₃) ₄ (nact) ₂ (blue violet)	tr P-1 1	6.827(1) 7.842(1) 8.018(1)	CuN ₄ O ₂	N _{eq} O _{ax}	2.008(1) × 4 2.577(1) × 2	90.0(5, 2) 90.0(5, 7, 9) 180	268
Cu(py) ₄ (F ₃ ac) ₂ (not given)	tr P-1 2	9.156(3) 9.505(4) 17.160(6)	CuN ₄ O ₂	pyN _{eq} acO _{ax}	2.044(4, 8) 2.355(4, 5)	90.5(1, 3) 178.5(2, 3) 90.0(1, 3, 5)	137 150
Cu(py) ₄ (F ₃ ac) ₂ (green)	tr P-1 2	9.528(6) 9.218(5) 18.471(10)	CuN ₄ O ₂	pyN _{eq} acO _{ax}	2.046(7, 4) 2.367(7, 2)	90.0(3, 1, 1) 178.5(3) 90.0(3, 3, 3) 172.9(3)	269
Cu(py) ₄ (CF ₃ SO ₃) ₂ (blue)	or Pbcn 4	10.5618(7) 16.287(1) 16.830(2)	CuN ₄ O ₂	N _{eq} O _{ax}	2.039(6, 19) 2.425(4) × 2	90.00(15, 1, 59) 90.00(11, 1, 90) 178.4(3, 1, 6) 179.8(2)	270
Cu(py) ₄ (PhPO ₃ H) ₂ (blue)	m P ₂ 2	9.821(1) 17.776(1) 10.688(1)	CuN ₄ O ₂	N _{eq} O _{ax}	2.071 2.320(7, 22)	90.0 90.0 176.9(3)	271
[Cu(py) ₄ (PhPO ₃ H) ₂ ·2MeOH (blue)]	tr P-1 2	9.148(3) 11.366(4) 17.466(5)	CuN ₄ O ₂	N _{eq} O _{ax}	2.049 2.490(4, 11)	90.0 90.0 177.7(1)	271
Cu(3-Mepy) ₂ (5, 5'-Et ₂ barb) ₂ (H ₂ O) ₂ (light purple)	m P ₂ i,c 2	11.196(5) 15.124(5) 9.449(5)	CuN ₄ O ₂	pyN _{eq} N _{eq} H ₂ O _{ax}	2.18(3) × 2 1.980(5) × 2 2.857(4) × 2	90.0(1, 3) 90.0(5, 38, 7)	272
[Cu(3-Mepy) ₄ (H ₂ O) ₂](ClO ₄) ₂ (blue)	m C ₂ /c 4	15.292(4) 9.672(2) 20.579(5)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	2.028(5, 13) 2.493(12, 23)	90.3(2) 90.9(2, 1, 7) 180	273

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
$\text{Cu}(\text{qu})_4(\text{CF}_3\text{SO}_3)_2^{\text{c}}$ (blue)	tr P-1 4	10.88(2)(3) 18.219(8) 22.384(3)	67.49(2) 88.64(2) 83.34(2)	CuN_4O_2	N_{eq} O_{ax}	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$	141
$[\text{Cu}(\text{im})_4(\text{H}_2\text{O})_2]_2\text{F}_2$ (dark blue)	m C2/c 4	12.708(3) 10.299(2) 13.915(3)	106.63(2)	CuN_4O_2	N_{eq} O_{ax}	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$	274
$\text{Cu}(\text{im})_4(\text{NO}_3)_2$ (purple)	or Pna2 ₁ 4	13.85(1) 9.83(1) 13.39(1)		CuN_4O_2	N_{eq} $\text{O}_2\text{NO}_{\text{ax}}$	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{O}_{\text{ax}}, \text{O}_{\text{ax}}$	275
$\text{Cu}(\text{im})_4(\text{MeOac})_2$ (not given)	m P2 ₁ /a 2	8.35(3) 9.85(3) 14.77(4)	94.8	CuN_4O_2	N_{eq} O_{ax}	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{O}_{\text{ax}}, \text{O}_{\text{ax}}$	276
$\text{Cu}(\text{im})_4(\text{tsval})_2$ (pink red)	m C2/c 4	18.252(5) 12.372(2) 17.534(3)	93.77(2)	CuN_4O_2	N_{eq} O_{ax}	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{O}_{\text{ax}}, \text{O}_{\text{ax}}$	277
$\text{Cu}(\text{im})_4(\text{ac})_2$ (blue violet)	m C2/c 4	13.187(2) 8.591(1) 17.644(2)	104.13(1)	CuN_4O_2	imN_{eq} acO_{ax}	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{O}_{\text{ax}}, \text{O}_{\text{ax}}$	278
$\text{Cu}(\text{im})_4(\text{P}(\text{PhO})_2\text{O}_2)_2$ (violet blue)	m P2 ₁ /c 2	8.264(3) 25.339(8) 10.216(3)	114.82(4)	CuN_4O_2	N_{eq} O_{ax}	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{O}_{\text{ax}}, \text{O}_{\text{ax}}$	279
$[\text{Cu}(\text{Meim})_4(\text{H}_2\text{O})_2]_2[\text{Cl}_2 \cdot \text{H}_2\text{O}]^{\text{c}}$ (dark green)	tr P-1 2	9.198(1) 9.338(1) 15.520(2)	99.41(1) 105.59(1) 103.36(1)	CuN_4O_2	N_{eq} $\text{H}_2\text{O}_{\text{ax}}$	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$	280

Cu(4-Meim) ₄ (ClO ₄) ₂ (purple)	m C2/c 4	15.742(5) 10.052(3) 16.654(5)	111.8(2)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	2.000(5, 7) 2.659(4) × 2	N _{eq} N _{eq} N _{eq} O _{ax}	90.0(2, 1, 2) 180.0(1) 91.2(2, 1)	281
Cu(1, 2-Me ₂ im) ₄ (ClO ₄) ₂ (blue purple)	m P2 ₁ /n 4	15.9011(7) 16.730(3) 10.919(1)	98.413(7)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	1.999(10, 12) 2.752(20, 32)	N _{eq} N _{eq} N _{eq} O _{ax} O _{ax} O _{ax}	90.0(2, 13) 178.1(2, 4) 90.0(4, 60) 174.1(3)	282
Cu(pz) ₄ (F ₃ ac) ₂ (violet)	tr P-1 1	9.600(2) 10.666(3) 8.703(2)	115.52(2) 113.58(2) 59.71(1)	CuN ₄ O ₂	N _{eq} acO _{ax}	2.013(3, 1) 2.452(3) × 2	N _{eq} N _{eq} N _{eq} O _{ax}	90.5(1) 88.8(1, 3, 4)	283
[Cu(5, 7-Me ₂ tpym) ₄ (H ₂ O) ₂](PF ₆) ₂ (blue)	m P2 ₁ /c 2	11.826(2) 9.463(2) 17.673(3)	97.50(1)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	2.035(3, 16) 2.456(3) × 2	N _{eq} N _{eq} N _{eq} O _{ax}	90.1(1) 93.5(1, 1, 3)	284
Cu(H ₂ O) ₂ (NH ₃) ₂ (mtpo) ₂ (purple)	m P2 ₁ /c 2	8.917(2) 13.998(2) 7.234(5)	106.77(3)	CuN ₄ O ₂	N _{eq} H ₃ N _{eq} H ₂ O _{ax}	2.016(3) × 2 1.997(3) × 2 2.554(3) × 2	N _{eq} N _{eq} N _{eq} O _{ax}	90.0(2, 8) 90.0(2, 4)	128
Ca[Cu(en)(5'-qmp) ₂ (H ₂ O) ₂] ₂ ·8H ₂ O (blue)	m P2 ₁ 2	10.997(2) 21.693(4) 8.872(2)	94.69(1)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	2.02(2, 4) 2.04(2, 2) 2.76(2, 2, 6)	N _{eq} N _{eq} N _{eq} O _{ax} O _{ax} O _{ax}	84.9(7) ^d 90.1(8, 5, 1) 175.0(10, 4, 4) 89.9(8, 11, 8) 161.6(9)	129
[Cu(en) ₂ (H ₂ O) ₂][F ₂ ·4H ₂ O] (mauve)	m C2/m 2	8.179(2) 15.854(2) 7.123(3)	101.98(3)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	2.021(5, 2) 2.571(6) × 2	N _{eq} N _{eq} N _{eq} O _{ax}	84.6(2) ^e 90.0(2, 2, 3)	285
[Cu(en) ₂ (H ₂ O) ₂](Ph ₄ B) ₂ ·2Me ₂ SO (not given)	tr P-1 1	9.871(2) 10.077(3) 14.612(3)	95.75(2) 91.59(2) 111.13(2)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	2.038(10, 17) 2.693(9) × 2	N _{eq} N _{eq} N _{eq} O _{ax}	84.9(4) ^e 86.5(5, 7, 1)	286
[Cu(en) ₂ (H ₂ O) ₂](4-amsal) (not given)	or Pbca 4	29.196(6) 10.788(2) 7.318(1)		CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	2.004(4, 1) 2.659(3) × 2		not given	287
[Cu(en) ₂ (H ₂ O) ₂](pbpc)·2MeOH (lilac)	tr P-1 1	9.943(1) 13.740(1) 15.968(1)	69.26(1) 83.27(1) 68.98(1)	CuN ₄ O ₂		not given		not given	288
Cu(en) ₂ (acts) ₂ (not given)	tr P-1 1	8.196(7) 9.094(4) 9.352(3)	84.51(3) 74.83(6) 63.57(5)	CuN ₄ O ₂	enN _{eq} O _{ax}	2.005(3, 4) 2.652(1) × 2	N _{eq} N _{eq} N _{eq} O _{ax}	84.77(6) ^f 97.43(5, 2)	82b

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(Meen) ₂ (H ₂ O) ₂](ox) (not given)	m P2 ₁ /n 2	7.198(3) 12.674(4) 7.693(2)	97.32(3)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax} 2.032(5, 24) 2.503(5) × 2	85.9(2) ^e 90.9(18, 3.6)	289
[Cu(Meen) ₂ (H ₂ O) ₂](mal) (not given)	or Pbcn 8	23.861(16) 10.312(6) 12.663(10)		CuN ₄ O ₂	N _{eq} H ₂ O _{ax} 2.031(7, 32) 2.513(7) 2.681(6)	85.4(5) ^f 90.1(5, 8.1) 174.4(4)	290a
Cu(Meen) ₂ (ClO ₄) ₂ (not given)	m P2 ₁ /c 2	8.351(7) 11.850(3) 8.439(6)	113.93(5)	CuN ₄ O ₂	N _{eq} O _{ax} 2.035(5, 31) 2.575(6) × 2	84.6(2) ^f 87.45(20, 2.75)	290b
[Cu(Meen) ₂ (H ₂ O) ₂](adip) (not given)	tr P-1 1	7.241(2) 8.271(3) 8.304(3)	89.92(3) 114.87(2) 90.45(2)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax} 2.028(2, 20) 2.561(2) × 2	85.4(1) ^e 91.5(1, 2.6)	291a
Cu(Meen) ₂ (NO ₃) ₂ (not given)	m P2 ₁ /c 2	8.143(7) 11.939(4) 7.509(4)	111.18(3)	CuN ₄ O ₂	N _{eq} O _{ax} 2.03(1, 2) 2.54(1) × 2	86.2(4) ^e 93.9(4, 6)	291b
[Cu(Et ₃ en) ₂ (H ₂ O) ₂](NO ₃) ₂ (not given)	m P2 ₁ /c 2	7.123(3) 15.264(6) 10.011(3)	92.21(3)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax} 2.068(4, 13) 2.498(4) × 2	85.1(2) ^e 92.2(2, 2.3)	292
[Cu(1, 3-pn) ₂ (H ₂ O) ₂]F ₂ (mauve)	or Pccn 4	16.350(2) 11.383(4) 6.983(4)		CuN ₄ O ₂	N _{eq} H ₂ O _{ax} 2.025(6, 5) 2.637(6) × 2	90.0(2, 8) ^d 87.8(2, 6)	293
[Cu(bip) ₂ (H ₂ O) ₂]Br ₂ (dark blue)	m C2/c 4	9.865(2) 12.790(2) 20.878(2)	101.98(1)	CuN ₄ O ₂	not given	not given	294
[Cu(Meims) ₂ (H ₂ O) ₂](ClO ₄) ₂ ·2H ₂ O (blue)	m P2 ₁ /n 2	7.696(2) 13.875(3) 13.186(2)	95.98(2)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax} 1.997(3, 7) 2.501(4) × 2	88.7(1) ^d	295
[Cu(dim) ₂ (H ₂ O) ₂](ClO ₄) ₂ ·2H ₂ O (not given)	m C2/m	11.520(2) 19.517(3)	105.24(2)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax} 1.996(4) × 4 2.624(3) × 2	90.0(2, 6) ⁱ 180	296

2	7.131(1)		CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	2.006(3, 15) 2.325(3) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	81.1(1) ^c 91.2(1, 4)	297
m P2 ₁ /c 2	13.190(2) 8.934(2) 7.649(1)	103.81(2)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	2.005(6, 39) 2.365(6) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	82.8(5) ^c 90.7(5, 1.9)	298
m P2 ₁ /n 2	13.579(2) 10.904(3) 10.329(2)	103.5	CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	2.005(6, 4) 2.46(1) × 2	N _{eq} , N _{eq}	88.7(2) ^d 90.0(3, 1.3)	299
m P2 ₁ /n 2	8.460(3) 8.503(3) 20.341(8)	98.41(3)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	1.988(10, 8) 2.497(9) × 2	N _{eq} , N _{eq}	86.5(4) ^d 93.5(5) 180 90.0(4, 4, 7) 180	300
tr P-1 1	9.856(1) 12.724(1) 9.401(1)	101.69(8) 96.717(9) 70.145(8)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	2.016(11, 19) 2.426(9) × 2	N _{eq} , N _{eq}	85.5(4) ^d 94.5(4) 180	300
m P2 ₁ /n 2	12.663(4) 9.573(3) 18.813(4)	107.16(2)	CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	2.005(60, 35) 2.770(70) × 2	O _{ax} , O _{ax} O _{ax} , N _{eq} N _{eq} , N _{eq}	90.0(4, 2.3) 83.0(2, 1) ^e 84.0(2)	301
m P2 ₁ /a 2	13.730(14) 6.240(14) 11.670(10)	102° 33'(13')	CuN ₄ O ₂	N _{eq} H ₂ O _{ax}	2.015(5, 19) 2.570(5) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	90.6(2) ^d 92.4(2, 1.8)	302
tr P-1 1	6.894(1) 8.478(1) 9.553(1)	98.61(1) 116.13(1) 35.88(1)	CuN ₄ O ₂	N ₃ N _{eq} H ₂ O _{ax}	1.989(4) × 2 2.005(3) × 2 2.467(3) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	90.0(1, 2.2) 74.1(1) ^e 95.0(1, 10.9)	303
tr P-1 1	7.598(4) 7.655(3) 10.796(5)	96.24(3) 110.10(3) 90.24(3)	CuN ₄ O ₂	N ₃ N _{eq} O _{ax}	1.943(2) × 2 1.984(2) × 2 2.521(3) × 2	N _{eq} , O _{eq} O _{eq} , N _{ax} N _{eq} , N _{ax}	89.06(8) ^d 91.78(7) 92.20(8)	304
or Pbca 4	9.669(6) 12.192(3) 14.266(3)		CuN ₄ O ₂	N _{eq} O _{eq} SCN _{ax}	1.956(5) × 2 1.995(5) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	88.1(2) 70.9(2) ^e	305
m P2 ₁ /n	8.696(9) 11.052(4)	85.24(5)	CuN ₄ O ₂	SCN _{eq} N _{eq}				
			[Cu(ammur) ₂ (H ₂ O) ₂] (deep green)					
			[Cu(odpy) ₂ (H ₂ O) ₂](ClO ₄) ₂ (green)					
			[Cu(Meims) ₂ (H ₂ O) ₂]·2MeSO ₄ (blue)					
			[Cu(papt) ₂ (H ₂ O) ₂](BF ₄) ₄ (brown)					
			[Cu(papt) ₂ (H ₂ O) ₂](BF ₄) ₄ (green)					
			K ₂ [Cu(ggty) ₂ (H ₂ O) ₂]·4H ₂ O (red purple)					
			[Cu(hist) ₂ (H ₂ O) ₂]·4H ₂ O (dark blue)					
			Cu((bzpy) ₂ (N ₃) ₂) (dark green)					
			Cu(acatr) ₂ (NCS) ₂ (grass green)					
			[Cu(hb-bzim) ₂ (NCS) ₂]·2Me ₂ CO (green)					

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
Cu(dmen) ₂ (NO ₃) ₂ (blue violet)	2 m P2 ₁ /c	19.105(7) 7.425(2) 8.533(2)	105.56(3)	CuN ₄ O ₂	O _{ax} N _{eq} O ₂ :NO _{ax}	93.3(2) 84.8(3) ^e 89.9(3, 4.4)	306
Cu(1, 2-pn) ₂ (NO ₃) ₂ (not given)	2 m P2 ₁ /c	13.072(3) 8.793(10) 10.193(4)	105.02(7)	CuN ₄ O ₂	N _{eq} O ₂ :NO _{ax}	87(1) ^f 91(1, 2)	307
Cu(2Me-1,2-pn) ₂ (NO ₃) ₂ (not given)	2 m P2 ₁ /c	7.960(6) 9.263(2) 18.441(6)	92.55(2)	CuN ₄ O ₂	not given	not given	308
Cu(1, 3-pn) ₂ (NO ₃) ₂ (not given)	4 or Pbca	9.296(3) 8.180(5) 16.335(16)		CuN ₄ O ₂	N _{eq} O ₂ :NO _{ax}	94.9(5) ^d 94.6(5, 2.3)	309
Cu(Me ₂ -1,3-pn) ₂ (NO ₃) ₂ (blue violet)	4 m P2 ₁ /u	10.240(4) 5.951(1) 19.333(3)	91.00(1)	CuN ₄ O ₂	N _{eq} O ₂ :NO _{ax}	92.56(2) ^d 90.28(2, 7.35)	310
Cu(Mepicam) ₂ (NO ₃) ₂ (blue)	2 m P2 ₁ /c	7.457(2) 7.04(2) 14.60(4)	123.4(1)	CuN ₄ O ₂	N _{eq} O ₂ :NO _{ax}	81.9(4) ^e 91.8(3, 3.1)	311
Cu(3, 5Me ₂ guapz) ₂ (NO ₃) ₂ (blue)	2 tr P-1	10.08(3) 9.708(3) 5.464(2)	92.99(3) 101.24(3)	CuN ₄ O ₂	N _{eq} O ₂ :NO _{ax}	101.0(2) 92.6(2, 6)	312
Cu(Meen) ₂ (ClO ₄) ₂ (not given)	1 m P2 ₁ /c	9.367(3) 8.351(7) 11.850(3)	108.54(3) 113.93(5)	CuN ₄ O ₂	N _{eq} O ₃ :ClO _{ax}	84.6(2) ^e 87.5(2, 2.8)	313a
Cu(Eten) ₂ (ClO ₄) ₂ (violet)	2 tr P-1	8.439(6) 8.131(8) 8.762(13)	65.33(16) 65.98(11)	CuN ₄ O ₂	N _{eq} O ₃ :ClO _{ax}	85.0(1) ^e 85.9(1, 4.3)	313b
Cu(ampv) ₂ (ClO ₄) ₂	1 tr	9.786(12) 7.630(2)	63.34(8) 106.67(3)	CuN ₄ O ₂	N _{eq} O _{ax} :O _{ax}	177.0(4)	314

(blue)	P1	7.980(1) 7.972(5)	108.75(3) 95.90(2)		O ₃ ClO _{ax}	2.598(9, 34)	O _{ax} , N _{eq} N _{eq} , N _{eq}	90.0(3, 12.3) 81.3(3, 3) ^e 98.6(3, 1.2) 176.6(5, 7)	315
Cu(aepy) ₂ (ClO ₄) ₂ (not given)	tr P-1	8.275(3) 9.772(4) 7.894(4)	126.88(2) 82.20(3) 111.22(3)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	2.015(2, 10) 2.883(2) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	86.54(7) ^d 93.46(7) 90.00(7, 10.98)	316
Cu(1, 2-pn) ₂ (ClO ₄) ₂ (not given)	tr P1, P-1	5.82(1) 7.66(1) 9.42(2)	102.0(1) 102.8(1) 77.6(1)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	2.03(3, 1) 2.61(2) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	86(1) ^d 93(1, 4)	317
Cu(1, 3-bn) ₂ (ClO ₄) ₂ (red violet)	m P2 _{1/c}	6.807(3) 15.952(7) 8.524(3)	105.92(2)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	2.028(7, 2) 2.676(10) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	88.9(3) ^d 89.2(3, 2.9)	317
Cu(1, 3-bn) ₂ (ClO ₄) ₂ (blue violet)	m P2 _{1/c}	9.626(9) 11.545(4) 7.856(4)	102.46(2)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	2.027(6, 10) 2.579(6) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	83.3(2) ^d 90.0(2, 2.1)	318
Cu(Me ₂ -1,3-pn)(ClO ₄) ₂ (light blue)	m P2 _{1/c}	9.724(2) 6.287(1) 16.19(3)	97.66(1)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	2.025(3, 4) 2.602(3) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	88.1(1) ^d 90.0(1, 1.9) 88.9(1, 3.0)	319
Cu(athal) ₂ (ClO ₄) ₂ (purple)	m P2 _{1/c}	11.562(3) 9.282(2) 10.028(3)	94.20(1)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	2.008(8, 11) 2.571(8) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	82.9(3) ^e	320
Cu{N(py) ₃ } ₂ (ClO ₄) ₂ (yellowish brown)	m P2 _{1/n}	18.638(2) 8.552(3) 9.822(1)	91.44(1)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	1.999(3, 7) 2.593(8) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	84.5(1) ^d 92.7(2, 3.6)	321
Cu(Me ₂ en) ₂ (F ₆ acac) ₂ (purple)	tr P-1	8.174 8.495 11.910	120.32 109.02 94.65	CuN ₄ O ₂	N _{eq} O _{ax}	2.06(2, 5) 2.76(1) × 2		not given	322
[Cu(1, 3-pn) ₂ (3- <i>l</i> bz)(H ₂ O)]·(3- <i>l</i> bz) (dark blue)	or P2 _{1,2}	34.98(2) 11.839(2) 6.362(2)		CuN ₄ O ₂	N _{eq} bzO _{ax} H ₂ O _{ax}	2.04(8, 4) 2.67(3) 2.51(5)	N _{eq} , N _{eq} N _{eq} , O _{ax} N _{eq} , O _{ax}	92(2, 3) ^d 90(2, 5) 91(8, 5)	323
[Cu(Meen) ₂ (trr)(H ₂ O)]·H ₂ O (violet)	or P2 _{1,2}	9.561(7) 9.578(8) 18.037(9)		CuN ₄ O ₂	N _{eq} O _{ax} H ₂ O _{ax}	2.032(8, 22) 2.630(7) 2.439(8)	N _{eq} , N _{eq} N _{eq} , O _{ax} O _{ax} , O _{ax}	84.6(3, 1) ^e 90.0(3, 5.8) 173.0(3)	

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	N_{eq} O_{ax}	Cu-L [Å]	N_{eq} , N_{eq} N_{eq} , N_{eq} N_{eq} , O_{ax}	L-Cu-L [°]	Ref.
Cu(1,3-pn) ₂ (bz) ₂ (blue)	m P2 ₁ /c 2	11.498(2) 8.295(2) 11.383(2)	91.84(1)	CuN ₄ O ₂	N_{eq} bzO _{ax}	2.035(3,8) 2.468(3) × 2	N_{eq} , N_{eq}	94.4(5) ^d	324
Cu(imp-1,2-pn) ₂ (sal) ₂ (not given)	m P2 ₁ /c 2	8.774(6) 19.885(12) 8.771(6)	97.15(7)	CuN ₄ O ₂	N_{eq} sulO _{ax}	2.019(4,30) 2.606(4) × 2	N_{eq} , N_{eq} N_{eq} , O_{ax}	83.3(3) ^f 93.8(1,4,3)	325
Cu(1,3-pn) ₂ (2-Mebz) ₂ (blue)	or Pbca 4	10.188(3) 21.821(12) 10.567(4)		CuN ₄ O ₂	N_{eq} O _{ax}	2.037(5,8) 2.501(4) × 2	N_{eq} , N_{eq} N_{eq} , O_{ax}	87.2(2) ^d 87.6(2,2)	326
Cu(1,3-pn) ₂ (3-NO ₂ bz) ₂ (blue green)	tr P-1 1	8.337(2) 11.281(3) 6.737(2)	98.72(2) 90.43(2) 111.63(2)	CuN ₄ O ₂	N_{eq} O _{ax}	2.031(2,2) 2.665(2) × 2	N_{eq} , N_{eq} N_{eq} , O_{ax}	89.3(1) ^d 93.1(1,2)	327
Cu(1,3-pn) ₂ (3-Mebz) ₂ (not given)	m P2 ₁ /c 2	12.571(4) 8.243(2) 11.480(4)	102.39(2)	CuN ₄ O ₂	N_{eq} O _{ax}	2.042(10,15) 2.486(6) × 2	N_{eq} , N_{eq} N_{eq} , O_{ax}	87.4(3) ^d 87.0(3,7)	328
Cu(1,3-pn) ₂ (4-Mebz) ₂ (not given)	m P2 ₁ /c 2	12.408(5) 8.466(2) 11.199(3)	90.049(49)	CuN ₄ O ₂	N_{eq} O _{ax}	2.034(7,4) 2.546(5) × 2	N_{eq} , N_{eq} N_{eq} , O_{ax}	87.9(3) ^d 86.1(2,2,3)	329
Cu(1,3-pn) ₂ (4-NO ₂ bz) ₂ (not given)	m P2 ₁ /c 2	12.473(1) 8.174(1) 11.586(1)	92.20(1)	CuN ₄ O ₂	N_{eq} O _{ax}	2.044(7,4) 2.528(8) × 2	N_{eq} , N_{eq} N_{eq} , O_{ax}	87.3(3) ^d 86.7(3,3,1)	330
Cu(1,3-pn) ₂ (3-Clbz) ₂ (blue)	m P2 ₁ /c 2	12.514(6) 8.198(4) 11.449(5)	103.10(4)	CuN ₄ O ₂	N_{eq} O _{ax}	2.04(1) × 4 2.48(1) × 2	N_{eq} , N_{eq} N_{eq} , O_{ax}	87.6(3) ^d 94.0(3)	331
Cu(1,3-pn) ₂ (4-Clbz) ₂ (blue)	m P2 ₁ /c 2	12.458(3) 8.989(3) 10.605(3)	91.72(2)	CuN ₄ O ₂	N_{eq} O _{ax}	2.04(1,1) 2.53(1) × 2	N_{eq} , N_{eq} N_{eq} , O_{ax}	87.8(5) ^d 95.5(5)	331
Cu(1,3-pn) ₂ (3-Brbz) ₂ (bluish green)	m P2 ₁ /c	12.72(1) 8.24(1)	104.02(8)	CuN ₄ O ₂	N_{eq} O _{ax}	2.051(9,1) 2.481(8) × 2	N_{eq} , N_{eq} N_{eq} , O_{ax}	89.2(4) ^d 93.3(3,1,6)	332

Cu(1,3-pn) ₂ (4-Brbz) ₂ (not given)	2 m P2 ₁ /c 2	11.50(1) 14.503(5) 11.093(4) 7.636(3)	102.54(3)	CuN ₄ O ₂	N _{eq} O _{ax}	2.036(10,3) 2.496(11) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	88.9(4) ^d 92.9(4,6)	333
Cu(1,3-pn) ₂ (3-lbz) ₂ (blue)	m P2 ₁ /a 2	11.717(2) 8.283(2) 13.000(1)	107.79(1)	CuN ₄ O ₂	N _{eq} bzO _{ax}	2.035(3,6) 2.500(3) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax} O _{ax} , O _{ax}	87.9(1) ^d 90.0(1,2,1) 86.7(1,9) 180	334
Cu(1,3-pn) ₂ (4-Fbz) ₂ (dark blue)	m P2 ₁ /c 2	11.808(7) 8.632(4) 10.979(4)	90.29(6)	CuN ₄ O ₂	N _{eq} O _{ax}	2.035(6,1) 2.503(5) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	87.2(2) ^d 93.2(2,3,0)	335
Cu(1,3-pn) ₂ (4-lbz) ₂ (not given)	m P2 ₁ /c 2	14.80(1) 11.15(1) 7.74(1)	105.38(8)	CuN ₄ O ₂	N _{eq} O _{ax}	2.014(21,9) 2.548(18) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	89.1(8) ^d 92.15(80,1.55)	336
Cu(8-Mesthp) ₂ (py) ₂ (red violet)	m P2 ₁ /c 2	7.502(2) 20.352(4) 9.757(2)	98.67(3)	CuN ₄ O ₂	pyN _{eq} N _{eq} O _{ax}	2.032(2) × 2 1.999(2) × 2 2.825(2) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax} O _{ax} , O _{ax}	90.0(1,8) 74.5(1) ^e 90.0(1,15.5) 180.0(1)	337
Cu(4-Mebzcs) ₂ (py) ₂ (blue)	tr P-1 1	7.150(4) 9.314(5) 10.523(7)	93.55(5) 97.15(5) 109.06(4)	CuN ₄ O ₂	pyN _{eq} N _{eq} O _{ax}	2.010(2) × 2 1.956(2) × 2 2.616(4) × 2	N _{eq} , N _{eq} N _{eq} , N _{eq} O _{ax} , O _{ax}	90.2(2)	338
Cu(mocpz) ₂ (py) ₂ (violet)	m P2 ₁ 2	11.441(2) 9.339(3) 12.493(3)	99.95(1)	CuN ₄ O ₂	pyN _{eq} N _{eq} O _{ax}	2.057(2) × 2 1.950(2) × 2 2.680(2) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax} O _{ax} , O _{ax}	89.28(2) 72.21(8) ^e 83.07(8)	339
Cu(5,5'-Et ₂ barb) ₂ (py) ₂ (not given)	m P2 ₁ /c 2	11.235(5) 9.958(4) 12.624(5)	102.6(2)	CuN ₄ O ₂	pyN _{eq} N _{eq} O _{ax}	2.032(5) × 2 1.983(5) × 2 2.723(5) × 2	N _{eq} , N _{eq} N _{eq} , N _{eq} O _{ax} , O _{ax}	90.0(2,5)	340
Cu(pic) ₂ (pyam) ₂ (dark green)	m P2 ₁ /c 4	12.311(5) 15.435(5) 15.320(6)	115.95(3)	CuN ₄ O ₂	pyamN _{eq} picO _{eq} picN _{ax}	2.051(6,3) 1.951(7,10) 2.599(7,13)	O _{eq} , N _{ax} N _{ax} , N _{ax} N _{eq} , N _{eq} N _{eq} , N _{ax} O _{eq} , N _{eq} O _{eq} , O _{eq}	73.8(5,3) ^e 106.2(5,2.5) 176.9(4) 178.8(4) 86.1(5) 90.3(5,1.8) 177.1(4)	341

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
Cu(bipyam) ₂ (tcne) ₂ (golden brown)	tr P-1 1	10.075(2) 10.911(2) 7.841(2)	107.98(2) 104.78(2) 108.40(2)	CuN ₄ O ₂	2.003(6, 14) 2.250(4) × 2	86.28(23) ^e 89.78(17, 48)	342
[Cu(prim) ₄](ClO ₄) ₂ ·EtOH (purple)	or Aba2 ₁ Acam 4	20.022(2) 12.263(1) 11.073(1)		CuN ₄ O ₂	N _{eq} O _{ax} 1.998(6, 6) 2.839(6, 63)	90.1(3, 3) 176.9(3) 52.1(2, 1.5) ^f	343
[Cu(bpy) ₂ (NO ₂) ₂][BF ₄] (not given)	m P2 ₁ /n 4	10.839(3) 12.367(3) 16.219(3)	105.25(5)	CuN ₄ O ₂	(NO ₂)O _{eq} bpyN _{eq} bpyN _{ax} 2.117(6) 2.097(5, 44) 1.998(5, 8)	80.1(2, 7) ^e 99.4(2, 1.6) 178.6(1) 98.5(2) 90.4(2, 3.2) 164.1(1) 149.2(1) 111.9(2) 97.3(2) 52.7(2) ^f	344
[Cu(2-pira) ₂ (NO ₂)][NO ₃] (not given)	m P2 ₁ /n 4	14.817(4) 14.872(3) 11.595(2)	94.28(2)	CuN ₄ O ₂	(NO ₂)O _{eq} N _{eq} N _{ax} 2.317(9, 75) 2.085(9, 16) 1.974(9, 5)	54.3(4) ^f 145.8(4, 2.6) 95.2(4, 14.8) 87.0(4, 11.3) ^e 179.4(4)	345
[Cu(bpy) ₂ (NO ₂)][NO ₃] (not given)	m P2 ₁ /n 4	11.101(18) 12.058(23) 15.385(30)	99.10(10)	CuN ₄ O ₂	(NO ₂)O _{eq} bpyN _{eq} bpyN _{ax} 2.283(10, 45) 2.083(10, 18) 1.993(10, 15)	103.0(4) 80.7(4, 4) ^e 99.0(4, 4) 52.5(4) ^f 102.4(4, 3.2) 154.6(4, 3.5) 90.2(4, 3.4) 179.6(4)	346
[Cu(bpy) ₂ (NO ₂)][NO ₃] (not given)	m P2 ₁ /n	11.225(2) 12.035(5)	99.55(2)	CuN ₄ O ₂	bpyN _{eq} (NO ₂)O _{eq} 2.083(10, 18) 2.284(10, 46)	80.7(4, 4) ^e 89.9(4, 9.6)	347 348

4	15.109(5)		bpyN _{ax}	1.993(11, 13)	O _{eq} , N _{ax} O _{eq} , O _{eq} O _{eq} , N _{eq}	90.5(4, 3.7) 52.5(4) ^f 98.0(4, 8.9) 154.5(4, 3.4) 103.0(4) 179.6(4)
m	11.217(2)	CuN ₄ O ₂	bpyN _{eq} (NO ₃)O _{eq}	2.085(2, 14)	N _{eq} , N _{ax} N _{eq} , N _{ax}	80.24(9, 27) ^e
P2 ₁ /n 4	11.936(5) 14.969(3)		bpyN _{ax}	2.278(3, 74) 2.987(2, 3)	N _{eq} , N _{ax} N _{eq} , O _{eq}	99.79(9, 26) 102.85(9) 102.02(9, 3.01) 154.84(9, 3.18) 179.96(9)
m	11.168(4)	CuN ₄ O ₂	bpyN _{eq} (NO ₃)O _{eq}	2.085(2, 25)	N _{ax} , N _{ax} N _{ax} , O _{eq} O _{eq} , O _{eq}	89.98(9, 3.45) 53.5(9) ^f
P2 ₁ /n 4	11.945(5) 14.909(6)		bpyN _{ax}	2.155(2) 2.414(2) 1.989(2, 3)	N _{eq} , N _{ax} N _{eq} , N _{eq} N _{eq} , O _{eq}	80.37(8, 3.4) ^e 99.69(8, 62) 102.08(8) 102.36(8, 3.45) 155.22(7, 3.69) 179.63(8)
m	11.067(2)	CuN ₄ O ₂	bpyN _{eq} (NO ₃)O _{eq}	2.085(2, 57)	N _{ax} , N _{ax} N _{ax} , O _{eq} O _{eq} , O _{eq}	89.97(8, 3.51) 53.62(7) ^f
P2 ₁ /n 4	12.014(3) 14.855(7)		bpyN _{ax}	2.051(2) 2.536(2) 1.990(2, 8)	N _{eq} , N _{ax} N _{eq} , N _{eq} N _{eq} , O _{eq}	80.39(7, 84) ^e 99.87(7, 1.62) 100.68(7) 103.02(6, 4.55) 155.82(7, 4.91) 178.91(7)
m	7.854(1)	CuN ₄ O ₂	N _{eq}	2.119(6, 91)	N _{ax} , N _{ax} N _{ax} , O _{eq} O _{eq} , O _{eq}	89.86(8, 2.89) 53.83(4) ^f
P-1 2	14.303(2) 11.852(2)		N _{ax}	2.180(7) 2.569(8) 2.000(5, 12)	N _{eq} , N _{ax} O _{eq} , N _{ax} O _{eq} , O _{eq} O _{eq} , N _{eq}	84.2(2, 4.0) ^e 99.4(2, 1) 89.3(3, 6.1) 53.1(2) ^f 91.4(2) 114.3(2)
		[Cu(bpy) ₂ (NO ₃) ₂] (not given) (at 165 K)				
		[Cu(bpy) ₂ (NO ₃) ₂] (not given) (at 100 K)				
		[Cu(bpy) ₂ (NO ₃) ₂] (not given) (at 20 K)				
		[Cu(4, 4', 5, 5'-Me ₄ dirn) ₂ (NO ₃) ₂] (red purple)				

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(bpy) ₂ (NO ₃)]NO ₃ ·H ₂ O (not given)	tr P-1 2	7.488(2) 10.017(2) 15.061(2)	106.183(4) 91.335(3) 89.662(3)	CuN ₄ O ₂	bpyN _{eq} (NO ₃)O _{eq} bpyN _{ax}	140.4(2) 163.8(2) 104.8(2) 178.3(4) 81.3(2, 3) ^e 101.9(2, 1.9) 170.7(1) 140.7(1) 87.9(2, 3.9) 127.5(2)	350
[Cu(bpy) ₂ (NO ₃)]NO ₃ ·H ₂ O (not given)	tr P-1 2	7.464(3) 10.002(3) 15.503(6)	111.50(5) 91.43(5) 90.47(4)	CuN ₄ O ₂	bpyN _{eq} (NO ₃)O _{eq} bpyN _{ax}	140.2(3) 47.7(3) ^f 86.2(3, 5.9) 133.7(3, 6.4) 85.9(3, 4.1) 81.5(3, 4) ^e 101.8(3, 1.8) 170.9(3)	351
[Cu(bpy) ₂ (fm)]BF ₄ ·0.5H ₂ O (blue)	tr P-1 2	7.570(2) 9.742(2) 14.994(2)	100.376(4) 96.806(3) 86.107(3)	CuN ₄ O ₂	bpyN _{eq} fmO _{eq} bpyN _{ax}	113.2(2) 79.6(2, 7) ^e 98.3(2, 8) 175.8(1) 91.4(2, 1.5) 100.5(2) 146.1(1)	352
[Cu(phen) ₂ (fm)]ClO ₄ (turquoise)	m C2/c 4	16.772(6) 11.5806(3) 12.5735(3)	111.370(2)	CuN ₄ O ₂	N _{eq} fmO _{eq} N _{ax}	81.3(1) ^e 99.0(1) 89.8(2, 5) 179.4(2)	353
[Cu(bpy) ₂ (ac)]BF ₄ (purple blue)	m P2 ₁ /c	7.333(2) 28.078(4)	111.42(5)	CuN ₄ O ₂	N _{eq} O _{eq}	79.5(2, 1.2) ^e 99.8(2, 2.0)	354

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	<i>a</i> [Å] <i>b</i> [Å] <i>c</i> [Å]	α [°] β [°] γ [°]	Chromophore	<i>Cu-L</i> [Å]	<i>L-Cu-L</i> [°]	Ref.
[Cu(phen) ₂ (ac)BF ₄ ·2H ₂ O (not given)	m P2/c 2	9.639(2) 8.237(2) 17.575(4)	108.66(2)	CuN ₄ O ₂	2.123(4) × 2 2.261(5) × 2 2.000(4) × 2	N _{eq} ¹ N _{eq} O _{eq} ¹ N _{ax} N _{eq} ¹ N _{ax}	357
[Cu(phen) ₂ (NO ₂)BF ₄ (not given)	tr P-1 2	10.442(4) 16.022(3) 7.408(2)	93.12(2) 101.86(2) 110.13(2)	CuN ₄ O ₂	2.167(3) × 2 2.072(4) 2.597(4) 2.022(4, 27)	N _{ax} ¹ N _{ax} N _{eq} ¹ N _{ax} N _{eq} ¹ N _{eq} N _{eq} ¹ O _{eq}	357
[Cu(phen) ₂ (ac)ClO ₄ ·2H ₂ O (blue)	m P2/c 2	9.671(3) 8.282(3) 17.595(4)	109.63(2)	CuN ₄ O ₂	2.124(4) × 2 2.257(5) × 2 1.994(5) × 2	O _{eq} ¹ N _{ax} N _{ax} ¹ N _{ax} O _{eq} ¹ O _{eq}	358
[Cu(phen) ₂ (ac)NO ₃ ·2H ₂ O (not given)	tr P-1 2	14.728(2) 10.499(2) 8.603(2)	104.19(5) 83.97(5) 96.81(5)	CuN ₄ O ₂	2.127(3, 44) 2.123(5) 2.448(5) 2.010(3, 10)	N _{eq} ¹ N _{eq} O _{eq} ¹ O _{eq}	358
Cu(Clqo) ₂ (bpy) ^c (dark red)	tr P-1 4	15.499(14) 14.260(13) 11.097(15)	71.26(8) 75.25(4) 81.07(5)	CuN ₄ O ₂	2.010(6, 11) 2.028(4) 2.055(6) 2.181(5) 2.320(4)	N _{eq} ¹ N _{eq} O _{eq} ¹ O _{eq} bpy ¹ N _{eq} bpy ¹ N _{ax} O _{ax}	359

Cu(esop) ₂ (phen) (green)	tr P-1 2	12.414(2) 12.882(2) 11.371(2)	112.55(2) 92.58(1) 83.96(1)	CuN ₄ O ₂	N _{eq} O _{eq} bpyN _{eq} bpyN _{ax} O _{ax}	2.051(7, 63) 2.106(5) 2.012(5) 2.203(5) 2.262(4)	N _{eq} , O _{eq} N _{ax} , O _{ax} O _{eq} , O _{ax} N _{eq} , N _{ax} N _{eq} , N _{ax} N _{eq} , O _{ax} N _{eq} , O _{eq} 94.4(3) 170.9(3) 169.6(3) 89.6(3) 98.2(7)	162.5(3) 80.1(3) ^e 88.6(3, 8.5) 164.6(2) 90.0(3) 93.2(3, 1.2) 172.2(4) 78.2(3) ^e 96.7(3, 6) 77.6(3) ^e 92.8(3, 1.4) 79.8(3) ^e	360																				
										Cu(dien)(im)(ClO ₄) ₂ (purple)	or P _{ma} 4	14.043(2) 9.322(2) 12.347(2)	112.55(2) 92.58(1) 83.96(1)	CuN ₄ O ₂	imN _{eq} N _{eq} O ₃ ClO _{ax}	1.970(5) 2.009(7, 3) 2.667(8, 96)	O _{eq} , O _{eq} N _{eq} , N _{eq} O _{ax} , N _{eq} O _{ax} , O _{ax}	83.6(3) ^e 90.0(3, 6.4) 170.2(3, 4.0) 90.1(3, 11.8) 177.6(3)	361										
																				Cu(dien)(1-Etim)(ClO ₄) ₂ (purple)	m P ₂ /c 4	8.326(2) 14.600(3) 15.055(3)	94.88(2)	CuN ₄ O ₂	imN _{eq} N _{eq} O ₃ ClO _{ax}	1.971(7) 2.012(9, 4) 2.628(18, 58)	N _{eq} , N _{eq} N _{eq} , N _{eq} O _{ax} , O _{ax} N _{eq} , N _{eq} N _{eq} , N _{eq} O _{ax} , O _{ax}	82.9(4, 7) ^e 89.7(4, 7.5) 172.0(4, 5.6) 90.0(5, 9.4) 172.0(5)	361

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(pcsm)(pyf)(ClO ₄) ₂ ·H ₂ O (dark blue)]	m P2 ₁ /n 4	8.443(4) 27.145(12) 9.781(5)	100.40(6)	CuN ₄ O ₂ pcsmN _{eq} pyrN _{eq} pyfO _{ax} O ₃ ClO _{ax}	2.012(4, 11) 1.931(4) 2.316(3) 2.726(3)	N _{eq} , N _{eq} N _{eq} , O _{ax}	362
Cu(pcsm)(pybox)(SO ₃ CF ₃) (dark blue)	m P2 ₁ /n 4	9.927(4) 26.727(5) 8.594(4)	98.72(2)	CuN ₄ O ₂ pcsmN _{eq} pyboxN _{eq} pyboxO _{ax} O _{ax}	1.991(5, 56) 2.000(4) 2.381(4) 2.709(6)	O _{ax} , O _{ax} N _{eq} , N _{eq} N _{eq} , O _{ax}	108
Cu(12-aneN ₄ ac) ₂ (light blue)	or Pccn 4	9.444(2) 15.300(6) 13.109(6)		CuN ₄ O ₂ O _{eq} N _{eq} N _{ax}	1.965(5) × 2 2.107(5) × 2 2.318(6) × 2	N _{eq} , N _{eq} N _{eq} , N _{ax} N _{ax} , N _{ax} O _{eq} , O _{eq} O _{eq} , N _{ax} N _{eq} , O _{eq}	363
[Cu(bcmgly) ₂]·2H ₂ O (pale blue)	m P2 ₁ /c 2	6.845(1) 14.095(4) 9.427(3)	91.95(2)	CuN ₄ O ₂ O _{eq} N _{eq} O _{ax}	1.954(2) × 2 2.062(2) × 2 2.321(2) × 2	N _{eq} , O _{ax} N _{eq} , O _{ax} O _{ax} , O _{ax}	364
[Cu(2-Clbquam) ₂]Cl ₂ ·2H ₂ O (deep blue)	tr P1 1	6.82(1) 7.19(1) 8.25(1)	95.3(3) 103.0(3) 110.0(3)	CuN ₄ O ₂ N _{eq} O _{ax}	2.017(-, 25) 2.635 × 2	N _{eq} , N _{eq} O _{ax} , O _{ax} O _{ax} , O _{ax}	365
[Cu(bpykhh) ₂]Cl ₂ ·4H ₂ O (not given)	m C2/c 4	14.504(4) 12.333(8) 14.630(3)	90.92	CuN ₄ O ₂ N _{eq} O _{ax}	2.000(5, 6) 2.465(5) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	366

$\{\text{Cu}(\text{bpykh})_2(\text{NO}_3)_2 \cdot 2\text{H}_2\text{O}$ (not given)	m P ₂ /n 2	7.60(5) 11.977(4) 14.463(6)	93.10(8)	CuN_4O_2	N_{eq} O_{ax}	2.013(5, 23) 2.464(4) × 2	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$	87.3(2) ^d 73.8(2, 3) ^e	366
$[\text{Cu}(\text{aenol})_2](\text{NCS})_2$ (violet)	tr P-1 1	6.624(2) 9.050(4) 9.532(4)	62.78(3) 80.64(3) 70.20(3)	CuN_4O_2	N_{eq} O_{ax}	2.038(7, 16) 2.441(8) × 2	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$	87.3(3) ^d 83.0(3, 4, 6) ^e	367
$\{\text{Cu}(\text{EtenOH})_2\}(\text{ClO}_4)_2$ (dark blue)	m P ₂ /n 4	15.914(3) 12.710(1) 9.041(2)	101.39(2)	CuN_4O_2	N_{eq} O_{ax}	2.027(4, 15) 2.460(3, 50)	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$	84.9(2, 1) ^e 96.9(1, 1, 9) 165.9(2, 2) 75.7(1, 1) ^e 85.2(1, 4, 5) 95.5(1, 3, 2)	368
$[\text{Cu}(\text{pslh}(\text{psalh}))(\text{ClO}_4)_2 \cdot \text{EtOH}$ (green)	m P ₂ /n 4	21.230(9) 11.493(5) 12.091(5)	100.47(4)	CuN_4O_2	N_{eq}	1.930(4) 2.062(4, 19) 2.062(3) 2.277(4) 2.334(4)	$\text{O}_{\text{ax}}, \text{O}_{\text{ax}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$	79.0(2) ^e 99.6(1) 71.5(2) ^e 86.1(1) 101.3(1) 78.2(1) ^e 102.8(1) 97.3(1) 93.9(1) 75.3(2) ^e 95.7(2) 111.9(2)	369
$[\text{Cu}(\text{EtenOH})_2](\text{suc})_2$ (not given)	or Pbcn 4	9.589(5) 14.003(9) 12.596(10)		CuN_4O_2	N_{eq} O_{ax}	2.057(4, 24) 2.510(2) × 2	$\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$	83.5(1) ^e 92.5(2, 9, 0) 172.6(1) 77.3(1) ^e 89.9(1, 12, 6)	370
$[\text{Cu}(\text{peme})_2][\text{Ag}_2\text{I}_4]$ (pale brown)	tr P-1 1	8.5402(14) 10.0286(16) 10.584(2)	110.221(10) 95.476(10) 93.420(12)	CuN_4O_2	N_{eq} O_{ax}	2.000(3) × 4 2.393(3) × 2	$\text{O}_{\text{ax}}, \text{O}_{\text{ax}}$ $\text{N}_{\text{eq}}, \text{N}_{\text{eq}}$ $\text{O}_{\text{ax}}, \text{O}_{\text{ax}}$ $\text{N}_{\text{eq}}, \text{O}_{\text{ax}}$	158.0(1) 86.9(1) ^e 93.1(1) 180 75.4(1) ^e 105.9(1, 1, 3)	371

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	N _{eq} , N _{ax} O _{ax}	L-Cu-L [°]	Ref.
[Cu(perme) ₂][Cu ₂ I ₄] (red brown)	tr P-1 1	8.574(2) 9.860(3) 10.441(3)	109.10(2) 95.04(2) 94.54(2)	CuN ₄ O ₂	2.011(4, 8) 2.380(4) × 2	N _{eq} , N _{ax} O _{ax}	86.7(1) ^d 99.3(1) 180 74.3(1, 1.5) ^e 105.8(1, 1.5)	372
[Cu(cyclam)(H ₂ O) ₂][F ₃ -4H ₂ O] (purple)	m C2/m 2	8.179(2) 15.854(2) 7.123(3)	101.98(3)	CuN ₄ O ₂	2.012(5) × 4 2.484(6) × 2	N _{eq} , N _{ax} H ₂ O _{ax}	90.0(2, 1.3) 94.4(3) ^d 85.6(3) ^e	285
[Cu(tim)(H ₂ O) ₂](NO ₃) ₂ (red violet)	or Cmca 4	12.007(2) 15.690(4) 10.869(5)		CuN ₄ O ₂	1.969(2) × 4 2.556(5) × 2	N _{eq} , N _{ax} H ₂ O _{ax}	90.0(1, 8.9) ^e 86.5(1)	373
[Cu(Mecyclam)(H ₂ O) ₂](ClO ₄) ₂ (red)	m P2 ₁ /c 2	8.201(1) 16.082(3) 10.432(1)	103.30(1)	CuN ₄ O ₂	2.033(4, 2) 2.815(5) × 2	N _{eq} , N _{ax} H ₂ O _{ax}	94.5(1) ^d 85.5(1) ^e 80.1(1, 2.3)	374
[Cu(tpy)(H ₂ O)(NO ₃)]NO ₃ ·H ₂ O (not given)	tr P-1 2	13.472(1) 10.861(8) 7.863(3)	98.7 78.6 103.6	CuN ₄ O ₂	1.996(-, 44) 2.286(3) 2.695(3)	N _{eq} , N _{ax} H ₂ O _{ax} O ₂ NO _{ax}	80.0(-, 2) ^e 119.4 90.1(1, 7.1)	375
Cu(trioe)(ClO ₄) ₂ ^c (red)	tr P-1 2	8.572(2) 8.499(3) 15.204(3)	80.42(5) 73.57(3) 69.82(4)	CuN ₄ O ₂	2.020(14, 35) 2.757(13) × 2	N _{eq} , N _{ax} O ₃ ClO _{ax}	172.5(6) 87.7(6) ^e 92.3(6) ^d 90.0(5, 1.3) 85.6(5) ^e 94.4(5) ^d	376
Cu(EtMecyclam)(ClO ₄) ₂ (red)	or Pbca 4	14.516(6) 11.674(7) 16.225(9)		CuN ₄ O ₂	2.018(11, 22) 2.660(19) × 2 2.032(6, 2) 2.716(11) × 2	N _{eq} , N _{ax} O ₃ ClO _{ax}	90.0(5, 1.4) 86.3(2) ^e 93.7(2) ^d	377
Cu(cyclam)(ClO ₄) ₂ (not given)	tr P-1 1	8.744(6) 8.022(5) 8.677(6)	118.73(2) 56.94(2) 113.50(2)	CuN ₄ O ₂	2.02(4) × 4 2.57(4) × 2	N _{eq} , N _{ax} O ₃ ClO _{ax}	86.0(2) ^e 94.0(2) ^d	378

Cu(pyha)(ClO ₄) ₂ (blue)	or P2 ₁ 2 ₁ 2 ₁ not given	14.295(6) 13.143(6) 9.068(6)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	1.99(2, 4) 2.66(2, 15)	N _{eq} , N _{eq} O _{ax} , O _{ax}	83.7(8, 2.7) ^e 109.5(8) 165.3(7, 4.1) 172.5(6)	379
Cu(indap)(ClO ₄) ₂ (blue purple)	or P2 ₁ cn 4	9.083(2) 15.198(3) 16.025(3)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	1.976(17, 44) 2.726(20, 53)	N _{eq} , N _{eq} O _{ax} , O _{ax}	83.5(7, 10) ^e 92.5(6) ^d 103.8(1) 165.0(9, 3)	380
Cu(mmeyclam)(ClO ₄) ₂ (not given)	m P2 ₁ /c 4	14.061(1) 8.908(1) 19.710(1)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	2.036(4, 15) 2.677(5, 20)	N _{eq} , N _{eq}	85.7(2, 2) ^e 179.7(2, 2) 94.3(2, 3) ^d 90.0(4, 8.1) 168.2(2)	381
Cu(bpdz)(ClO ₄) ₂ (blue violet)	or Pmm2 ₁ 8	13.746(1) 26.388(2) 11.225(1)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	2.02(3, 1) 2.67(3, 3)	N _{eq} , O _{ax} O _{ax} , O _{ax} N _{eq} , N _{eq}	83.5(10, 5) ^e 93(1) ^d 98(1, 5) 166.5(10, 25)	382
Cu(pydoc)(ClO ₄) ₂ (deep blue)	m C2/c 4	14.5156(6) 9.7118(4) 15.8993(6)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	2.016(5, 2) 2.693(4) × 2	N _{eq} , O _{ax} O _{ax} , O _{ax} N _{eq} , N _{eq}	90(1, 15) 171(1) 84.6(2) ^e 94.2(2) ^d 92.2(2, 7.6) 157.0(2)	383
Cu([15]aneN ₄)(ClO ₄) ₂ (not given)	tr P-1 1	8.663(3) 8.443(3) 7.804(3)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	2.035(36, 48) 2.591(4) × 2	N _{eq} , N _{eq}	86.1(9, 2.2) ^e 102.0(9) 168.9(8, 7) 78.7(4)	384
Cu(mmeyclam)(ClO ₄) ₂ (red)	m P2 ₁ /c 2	8.47(1) 9.26(1) 16.61(2)	CuN ₄ O ₂	N _{eq} O ₃ ClO _{ax}	2.042(3, 20) 2.594(5) × 2	N _{eq} , N _{eq} O _{ax} , O _{ax}	91.1(6, 1.3) 86.4(1, 3.5) 85.7(1) ^e	385
Cu(ethc)(ClO ₄) ₂ (not given)	m P2 ₁ /c 4	8.968(3) 8.435(3) 32.007(11)	CuN ₄ O ₂	N _{eq} O _{ax}	2.054(4, 28) 2.31(1, 1)	N _{eq} , N _{eq}	72.9(2) ^e 94.6(3, 2.8) ^d 163.2(3, 7) 96.9(3, 11.1)	386

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(amcyclam)(ClO ₄) ₂](ClO ₄) ₂ 6H ₂ O (blue)	m P2 ₁ /c 2	8.77(2) 14.08(1) 13.59(2)	107.5(1)	CuN ₄ O ₂ N _{eq} O ₃ ClO _{ax}	2.024(6, 3) 2.523(7) × 2	94.3(2) ^d 91.7(2, 3, 8)	387
[Cu(umte)(NO ₃)]NO ₃ ·H ₂ O (olive green)	m P2 ₁ /c 4	13.136(6) 17.294(7) 17.391(4)	127.66(2)	CuN ₄ O ₂ N _{eq} (NO ₃)O _{eq} N _{ax}	2.002(6, 4) 2.580(8, 77) 1.961(6, 4)	90.6(2, 1, 6) ^d 151.7(2) 175.5(2) 85.1(2, 7, 0) 125.7(3, 1) 46.5(2) ^f 77.4(2) 98.6(2)	388
[Cu(cyclax)(ClO ₄)ClO ₄ (not given)]	m P2 ₁ 2	8.718(3) 13.977(2) 9.102(9)	105.93(6)	CuN ₄ O ₂ N _{eq} O _{ax} O ₃ ClO _{ax}	2.016(7, 13) 2.703(6) 2.582(8)	93.7(3, 2) ^d 91.2(3, 2) 149.7(3) 161.2(3) 73.1(2) ^e 92.6(3, 16, 6) 164.2(2)	389
[Cu(mntcp)(ClO ₄)ClO ₄ (blue purple)]	m P2 ₁ /c 4	8.635(8) 12.755(3) 19.716(8)	105.08(5)	CuN ₄ O ₂ N _{eq} O _{ax} O ₃ ClO _{ax}	2.037(6, 56) 2.443(5) 2.62(3)	85.7(2, 9) ^e 94.9(2, 7) ^d 171.9(2, 7) 90.2(7, 16, 7) 81.4(2) ^e 92.6(2, 6, 2) 169.6(7, 2, 2)	389
[Cu(6-NO ₂ cyclam)](EtOH)·H ₂ O (blue)	tr P-1 2	8.537(4) 8.684(6) 21.614(15)	99.48(5) 98.69(5) 90.51(5)	CuN ₄ O ₂ O _{eq} N _{eq} N _{ax}	2.005(6, 1) 2.005(6, 11) 2.398(6, 31)	86.6(2, 4) ^e 93.4(2) ^d 179.4(3) 91.4(2, 2, 1) 179.3(2)	390

Ba[Cu(cyclamac)]·6H ₂ O ^c (blue)	tr P-1 2	8.928(3) 10.893(4) 15.889(2)	91.98(2) 101.96(1) 117.28(3)	CuN ₄ O ₂	N _{eq} O _{ax}	2.104(6, 81) 2.302(5) × 2	N _{eq} , O _{eq} O _{eq} , N _{eq} N _{eq} , O _{ax}	90.6(2, 2.8) 85.4(2) ^e 178.0(2) 86.7(2) ^e 83.6(2, 3.1) ^e	391
[Cu(en) ₂ (H ₂ O)Cl]Cl (purple)	m P2 ₁ /n 4	6.18(2) 15.16(4) 11.80(3)	99.0(1)	CuN ₄ O ₂	N _{eq} O _{ax}	2.097(7, 56) 2.278(6) × 2	N _{eq} , N _{eq} N _{eq} , O _{ax}	87.2(3) ^e 83.5(2, 3.8) ^e	392
[Cu(C ₅₀ H ₂₈ N ₄ S)(H ₂ O)(ClO ₄) ₂] (not given)	or P2 ₁ , 2 ₁ 4	16.793(3) 14.530(2) 10.393(2)		CuN ₄ OCl	enN _{eq} H ₂ O _{ax} Cl _{ax}	2.00(-, 2) 2.62 2.81	N _{eq} , N _{eq} N _{eq} , O _{ax} N _{eq} , N _{eq}	85.2(-, 1) ^e	393
Cu(1-viim) ₄ Cl ₂ (not given)	m P2 ₁ /c 2	7.40 14.70 11.26	112.1	CuN ₄ Cl ₂	N _{eq} Cl _{ax}	2.04(-, 1) 2.80 × 2	N _{eq} , N _{eq} N _{eq} , Cl _{ax}	93(2) 90(2)	394
[Cu(mic) ₄ Cl ₂]·2H ₂ O (blue)	tr P-1 1	8.888(8) 15.196(9) 15.297(4)	95.59(3) 102.42(2) 93.20(4)	CuN ₄ Cl ₂	N _{eq} Cl _{ax}	2.025(10, 5) 3.060(4) × 2	N _{eq} , N _{eq}	89.6(5)	395
Cu(nphy) ₂ Cl ₂ (green)	m C/2c 4	13.786(1) 8.192(1) 16.396(1)	122.5(1)	CuN ₄ Cl ₂	N _{eq} Cl _{eq} N _{ax}	2.028(3) × 2 2.253(1) × 2 2.764(3) × 2	N _{eq} , N _{eq} N _{eq} , Cl _{ax} Cl _{eq} , Cl _{eq}	87.0(2) 90.4(1) 94.6(1)	396
Cu(OH·1,3-pm) ₂ Cl ₂ (not given)	tr P-1 1	7.196(2) 7.585(2) 5.864(2)	98.51(2) 102.99(2) 92.97(3)	CuN ₄ Cl ₂	N _{eq} Cl _{ax}	2.038(4, 17) 2.953(1) × 2	N _{eq} , N _{eq} N _{eq} , Cl _{ax}	87.4(2) ^d 84.6(1, 4, 2)	397
Cu(dafone) ₂ Cl ₂ (blue)	m P2 ₁ /n 2	6.5911(6) 13.0712(5) 11.4813(9)	95.575(4)	CuN ₄ Cl ₂	N _{eq} Cl _{eq} N _{ax}	1.978(4) × 2 2.292(1) × 2 2.773(4) × 2	N _{eq} , Cl _{eq} N _{eq} , N _{ax} Cl _{eq} , N _{ax}	91.3(1) 76.6(1) ^e 93.11(9)	398
Cu(EtenOH) ₂ Cl ₂ (not given)	tr P1 1	6.569(3) 9.472(5) 6.158(3)	98.07(3) 101.57(3) 80.76(4)	CuN ₄ Cl ₂	N _{eq} Cl _{ax}	2.039(5, 21) 2.831(2) × 2	N _{eq} , N _{eq} N _{eq} , Cl _{ax}	86.3(2) ^e 93.4(2, 3.7)	399
[Cu(2Me-1,2-pm) ₂ Cl ₂]·0.5MeOH (not given)	tg P4/nmm 8	11.827(2) 11.827(1) 21.361(7)		CuN ₄ Cl ₂		not given			308

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	N_{eq} Cl_{eq} N_{ax}	Cu-L [Å]	N_{eq}, N_{eq} N_{eq}, N_{ax} N_{ax}, N_{ax} Cl_{eq}, N_{ax} Cl_{eq}, N_{eq} Cl_{eq}, Cl_{eq} N_{eq}, N_{eq} N_{eq}, N_{ax}	L-Cu-L [°]	Ref.
[Cu(12-aneN ₄)Cl ₂]·H ₂ O (light green)	or Pbca 8	13.452(2) 14.9319(6) 18.2366(14)		CuN ₄ Cl ₂	N_{eq} Cl_{eq} N_{ax}	2.067(1, 22) 2.292(1, 9) 2.348(2, 15)	N_{eq}, N_{eq} N_{eq}, N_{ax} N_{ax}, N_{ax} Cl_{eq}, N_{ax} Cl_{eq}, N_{eq} Cl_{eq}, Cl_{eq} N_{eq}, N_{eq} N_{eq}, N_{ax}	78.86(5) 77.64(6, 78) ^e 147.81(5) 100.77(4, 1.25) 92.67(4, 1.66) 95.81(2) 90.3(2) ^d 87.8(2, 3.6)	400
Cu(OH)·1,3-pn) ₂ (NCS) ₂ (red)	or Pbca 4	8.612(3) 10.926(3) 15.725(5)		CuN ₄ S ₂	N_{eq} NCS _{ax}	2.040(6, 18) 2.971(2) × 2	N_{eq}, N_{eq} N_{eq}, N_{ax}	90.3(2) ^d 87.8(2, 3.6)	401
Cu(1,3-pn) ₂ (NCS) ₂ (not given)	m C2/c 4	13.01(1) 6.89(1) 16.12(1)	92.3(2)	CuN ₄ S ₂	N_{eq} NCS _{ax}	2.017(5, 12) 3.154(2) × 2	N_{eq}, N_{eq} N_{eq}, N_{ax}	86.5(2) ^d 94.9(2, 4.5)	402
Cu(hd) ₂ (tu) ₂ (dark brown)	m C2/c 4	15.065(4) 10.216(3) 15.513(4)	103.86(2)	CuN ₄ S ₂	hdN _{eq} tuS _{ax}	2.951(1, 1) 2.941(1) × 2	N_{eq}, N_{eq}	99.8(1)	403
Cu(en) ₂ (Et ₃ sp) ₂ (dark violet)	m P2 ₁ 2	13.597(4) 6.945(1) 16.071(1)	121.22(1)	CuN ₄ S ₂	enN _{eq} S _{ax}	2.03(3, 3) 2.97(2, 7)	N_{eq}, N_{eq}	83.5(-, 3.3) ^e	404
[Cu(Meimthp) ₂](BF ₄) ₂ ·2EtOH (not given)	m P2 ₁ /c 2	8.262(2) 9.609(3) 22.582(5)	100.04(2)	CuN ₄ S ₂	N_{eq} S _{ax}	2.01(2, 2) 2.790(8) × 2	N_{eq}, N_{eq} N_{eq}, S_{ax}	90.0(9, 1.5) 80.4(7, 6) ^e 99.7(7, 6)	405
[Cu(Meimthb) ₂]Cl ₂ ·2H ₂ O (purple)	m P2 ₁ /c 2	7.742(1) 14.845(4) 9.936(2)	104.80(2)	CuN ₄ S ₂	N_{eq} S _{ax}	2.026(3, 8) 2.774(2) × 2	N_{eq}, N_{eq} N_{eq}, S_{ax}	90.5(2) 81.1(2, 1.8) ^e	406
[Cu(Phimthp) ₂](ClO ₄) ₂ ·5MeOH (orange brown)	m P2 ₁ /c 2	12.08(2) 11.98(2) 18.93(3)	96.19(4)	CuN ₄ S ₂	N_{eq} S _{ax}	2.020(9, 1) 2.824(5) × 2	N_{eq}, N_{eq} N_{eq}, S_{ax}	89.0(3) 78.6(3, 4) ^e	407
[Cu(9-aneN ₂ S) ₂](NO ₃) ₂ (purple)	or Pbca 4	14.50(1) 10.34(1) 12.95(1)		CuN ₄ S ₂	N_{eq} S _{ax}	2.077(3, 20) 2.707(1) × 2	N_{eq}, N_{eq} N_{eq}, S_{ax}	84.1(1) ^e 95.9(1) 82.0(1, 2, 2) ^e 98.0(1, 2, 2)	408

[Cu(daes) ₂](NO ₃) ₂ ^c (dark blue)	tr P-1 2	10.50(1) 9.96(1) 8.55(1)	90.34(5) 99.67(5) 89.24(5)	CuN ₄ S ₂	N _{eq} S _{ax}	2.058(4, 12) 2.745(1) × 2	N _{eq} , N _{eq} N _{eq} , S _{ax}	90.0(2, 1.2) 81.3(1, 8) ^e 98.7(1, 8)	408
Cu(N ₄ S ₂) ₂						2.057(4, 4) 2.772(1) × 2	N _{eq} , N _{eq} N _{eq} , S _{ax}	90.0(2, 1.6) 81.5(1, 1.2) ^e 98.6(1, 1.2)	409
Cu(cyclam)(F ₄ Sph) ₂ (red)	tr P-1 1	7.140(4) 8.320(5) 12.033(9)	75.02(3) 88.97(4) 71.07(5)	CuN ₄ S ₂	N _{eq} S _{ax}	2.008(3, 3) 2.940(1) × 2	N _{eq} , N _{eq}	85.2(1) ^e 94.8(1) 180	409
[Cu(2Me-1,2-pn) ₂ Br ₂] <cdot 0.6h<sub="">2O (not given)</cdot>	tg I4cm 8	12.171(2) 12.171(2) 21.435(9)		CuN ₄ Br ₂		not given	N _{eq} , S _{ax} S _{ax} , S _{ax}	90.0(1, 7.1) 180	308
Cu(dmpr) ₂ (NCSe) ₂ (not given)	or Pbca 4	8.852(4) 10.888(6) 15.947(8)		CuN ₄ Se ₂	N _{eq} NCS _{ax}	2.033(10, 13) 3.109(2) × 2	N _{eq} , N _{eq} N _{eq} , S _{ax}	90.2(4) ^d 87.4(3, 3.9)	410
Cu(2Me-1,2-pn) ₂ I ₂ (not given)	m Pc: P2/c 8	12.815(8) 12.215(5) 20.777(12)	104.36(6)	CuN ₄ I ₂		not given			308
(cbq) ₂ [CuCl ₄ (H ₂ O) ₂] (not given)	m P2 ₁ /c 2	6.522(3) 11.218(3) 11.790(3)	110.66(3)	CuCl ₄ O ₂	H ₂ O _{eq} Cl _{eq} Cl _{ax}	1.992(3) 2.305(1) 2.791(1)	O _{eq} , Cl _{eq} O _{eq} , Cl _{ax} Cl _{eq} , Cl _{ax}	81.57(8) 89.46(8) 88.07(3)	411
(NH ₄) ₂ [Cu(NH ₃) ₂ Cl ₄] (not given)	c Pm3 1/4	3.91(1)		CuCl ₄ N ₂	H ₃ N Cl	1.96(1) 2.76(1)		not given	412
[Cu(ach) ₂ Cl ₄] <cdot h<sub="">2O (olive green)</cdot>	m C2/c 4	15.153(14) 6.962(14) 18.274(16)	121.47(6)	CuCl ₄ N ₂	N _{eq} Cl _{eq} Cl _{ax}	2.049(3) × 2 2.258(2) × 2 2.967(2) × 2	N _{eq} , Cl _{eq} N _{eq} , Cl _{ax} Cl _{eq} , Cl _{ax}	97.2(1) 88.9(1) 90.0(1)	413
[Cu(dsc) ₂](ClO ₄) ₂ (purple)	m P2 ₁ /c 2	6.811(3) 9.225(4) 16.575(5)	90.54(3)	CuS ₄ O ₂	S O	2.334(1, 1) 2.350(1) × 2	S S S, O	91.0(1) ^d 80.7(1, 1.0) ^e	414
[Cu(14-aneS ₄)(ClO ₄) ₂] (blue)	m P2 ₁ /c 2	7.904(3) 8.830(4) 16.840(6)	126.33(2)	CuS ₄ O ₂	S _{eq} O ₃ ClO _{ax}	2.303(1, 6) 2.652(4) × 2	S _{eq} , S _{eq} S _{eq} , O _{ax}	89.9(4) ^e 90.1(4) ^d 92.4(2, 5.4)	415

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
Cu(15-ane-S ₄)(ClO ₄) ₂ (not given)	m P2 ₁ /c 2	7.994(1) 9.167(3) 13.978(3)	100.96(5)	CuS ₄ O ₂ S _{eq} O ₃ ClO _{ax}	2.318(3,5) 2.53(1) × 2	90.3(1) ^d 89.2(2, 3.9)	416
Cu(16-aneS ₄)(ClO ₄) ₂ (not given)	m P2 ₁ /c 2	7.931(3) 9.613(5) 14.885(5)	110.71(3)	CuS ₄ O ₂ S _{eq} O ₃ ClO _{ax}	2.359(1, 28) 2.482(5) × 2	90.37(5) ^d 88.8(2, 2.9)	416
[Cu(18-aneS ₄ N ₂)](ClO ₄) ₂ ·H ₂ O (not given)	or Pcab 8	12.020(8) 17.733(17) 21.999(23)		CuS ₄ N ₂ S _{eq} N _{ax}	2.543(5, 56) 2.022(13, 15)	86.11(16) ^f 84.23(17) ^e 95.77(16, 54) 169.57(17, 11) 84.83(4, 1.23) ^e 95.03(5, 37) 178.8(6)	417
(NH ₄) ₂ [Cu(NH ₃) ₂ Br ₄] (not given)	c Pm3 1/4	4.070(1)		CuBr ₄ N ₂ N Br	2.035(1) × 2 2.878(1) × 4	not given	418
Cu(ptrao) ₂ (qu) ₃ (blue)	tr P-1 2	11.55(1) 16.912(6) 9.729(9)	100.32(5) 109.96(13) 79.50(5)	CuO ₃ N ₃ O N	1.990(7, 30) 2.642(10) 2.047(9, 3) 2.359(11)	55.1(2) ^f 178.8(2) 98.1(3, 3.9) 163.7(3)	419a
[Cu(pmca)(H ₂ O) ₃](NO ₃) ₂ ·2H ₂ O (blue)	tr P-1 2	13.626(3) 6.512(1) 10.497(2)	101.48(1) 109.32(1) 94.83(1)	CuO ₃ N ₃ N _{eq} H ₂ O _{eq} H ₂ O _{ax}	2.000(17, 62) 1.976(2) 2.011(2, 9)	81.79(7, 22) ^e 163.50(9) 98.25(7, 2.48) 173.78(8) 92.04(7, 7.79) 79.90(7) 165.54(8)	419b
[Cu(azem)(H ₂ O) ₂](SO ₄) ₂ ·3H ₂ O (deep green)	m Cc	13.652(1) 14.630(1)	95.34(8)	CuO ₃ N ₃ H ₂ O _{eq} N _{eq}	1.968(-, 19) 1.966(-, 6)	93.50(6, 5.47) 87.11(6) ^e	420

4	6.795(1)	N_{ax} O_3SO_{ax}	2.793 2.500	N_{ax}, O_{ax} N_{eq}, N_{eq} N_{eq}, O_{eq}	89.35(6) 172.90(5) 80.56(7) ^e 95.58(7, 4) 171.77(8, 3.73)
m P2 ₁ /c 4	12.661(3) 13.660(7) 16.403(4)	N_{eq} H_2O_{eq} O_3ClO_{ax}	2.003(5, 4) 1.968(5) 2.729(5, 92)	O_{eq}, O_{eq} O_{eq}, O_{ax} O_{eq}, N_{ax} N_{eq}, N_{eq}	88.65(8) 91.28(7, 91) 87.11(7, 5.62) 85.5(2, 6) ^e 164.7(2) 94.8(2, 9) 72.9(1) 84.6(1, 2, 4) 97.1(2, 1, 1) 120.3(1) 83.4(2, 1, 2) 150.3(2)
m P2 ₁ /c 4	7.341(3) 13.919(2) 18.081(3)	H_2O_{eq} terpy/ N_{eq} fm/ O_{eq} fm/ O_{ax}	2.247(3) 2.011(4, 64) 1.935(3) 2.815(4)	O_{eq}, O_{ax} O_{ax}, O_{ax} N_{eq}, O_{eq} O_{ax}, O_{ax}	51.9(1) ^f 88.4(1) 140.0(1) 94.7(1, 6, 0) 120.5(1) 99.4(1, 6) 172.4(1) 80.0(2, 1) ^e 158.5(1)
tr	15.204(3)	terpy/ N_{eq}	1.987(9, 67)	O_{eq}, O_{ax}	80.4(2) ^e
P-1 2	12.825(2) 11.006(2)	croc/ O_{eq} croc/ O_{ax} H_2O_{ax}	1.962(7) 2.393(6) 2.420(7)	N_{eq}, O_{eq} N_{eq}, O_{ax} N_{eq}, N_{eq}	90.1(3) 100.2(3, 6) 172.4(2) 92.2(3, 5, 1) 79.1(3) ^e 80.6(4) 159.7(3) 170.5(3)

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromiophore	Cu-L [Å]	L-Cu-L [°]	Ref.	
[Cu(terpy)(H ₂ O)(ox)]·4H ₂ O (green)	tr P ₂ ₁ 2	10.192(2) 12.319(2) 8.397(3)	86.65(3) 96.80(3) 106.14(1)	CuN ₃ O ₂ terpyN _{eq} croCO _{eq} croCO _{ax}	2.043(7,97) 2.318(7) 1.940(7)	N _{eq} ¹ N _{eq} N _{eq} ¹ O _{eq} N _{eq} ¹ O _{ax} O _{eq} ¹ O _{ax} O _{eq} ¹ O _{ax}	80.1(3,2) 99.1(3,2) 172.4(4) 89.1(3) 106.0(3,4) 81.9(3)	423
[Cu(hist)(thr)(H ₂ O)(H ₂ O) (blue)	m P2 ₁ 2	5.843(10) 12.249(10) 11.049(10)	102.46	CuO ₃ N ₃ thrN _{eq} thrO _{eq} histN _{eq} histO _{ax}	2.00 1.97 1.98(-,3) 2.58	N _{eq} ¹ O _{ax} N _{eq} ¹ O _{ax} O _{ax} ¹ O _{ax} O _{eq} ¹ N _{eq}	79.64(9) ^e 88.39(10) 96.66(10, 10.14) 79.58(10, 7) ^e 158.89(1) 167.41(9) 100.43(10, 50) 179.24(10)	424
Cu(ibpca)(ac)(H ₂ O) (green)	m P2 ₁ /c 4	7.130(2) 10.918(3) 21.686(7)	105.59(2)	CuO ₃ N ₃ N _{eq} acO _{eq} acO _{ax} H ₂ O _{ax}	1.981(5,59) 1.987(4) 2.636(5) 2.518(4)	O _{eq} ¹ O _{ax} O _{ax} ¹ O _{ax} N _{eq} ¹ O _{ax} N _{eq} ¹ O _{eq}	54.1(2) ^f 103.8(2) 157.7(1) 93.9(2, 11.1) 93.6(2, 4.9) 159.1(2) 81.5(2) ^e 94.1(2) ^d 170.3(2)	425
Cu(asp)(hist)(H ₂ O) (light blue)	m P2 ₁ 2	12.654(4) 11.623(4) 5.931(2)	100.47(3)	CuO ₃ N ₃ aspN _{eq} aspO _{eq} histN _{eq} histO _{ax}	2.019(8) 1.972(6) 1.980(7, 26) 2.366(7)	N _{eq} ¹ O _{ax} N _{eq} ¹ O _{ax} O _{eq} ¹ O _{ax} O _{ax} ¹ O _{ax}	74.3(3) ^e 91.8(3, 8.1) 100.2(3) 87.7(2)	426

[Cu(ptcbx)(bpy)(H ₂ O)]·3H ₂ O (not given)	m P ₂ ₁ /n 4	7.046(1) 26.459(3) 11.001(2)	102.90(1)	CuO ₃ N ₃	H ₂ O _{ax}	2.642(7)	N _{eq} ,N _{eq} N _{eq} ,O _{eq}	170.8(2) 92.8(3) ^d 93.6(3, 8) 172.1(3, 9) 91.4(3) 82.2(3) ^e	427
[Cu(ptcbx)(bpy)(H ₂ O)]·3H ₂ O (not given)	m P ₂ ₁ /n 4	7.046(1) 26.459(3) 11.001(2)	102.90(1)	CuO ₃ N ₃	H ₂ O _{eq} bpy,N _{eq} N _{eq} O _{ax}	1.972(3) 1.999(3, 6) 2.013(3) 2.445(3, 54)	N _{eq} ,N _{eq} N _{eq} ,O _{ax}	81.1(1) ^e 97.0(1) 177.1(1) 85.8(1, 2, 8) 106.2(1, 1, 8) 74.3(1) ^e 73.1(1) ^e 93.0(1) 171.1(2) 146.7(1) 92.8(1, 8)	427
[Cu(salen)(cyt)(NO ₃)·H ₂ O (deep blue)	m P ₂ ₁ /c 4	7.453(3) 12.555(2) 20.36(7)	110.07(3)	CuO ₃ N ₃	salN _{eq} salO _{eq} cytN _{eq} cytO _{ax} O ₂ NO _{ax}	1.993(1, 55) 1.922(1) 2.008(1) 2.772(1) 2.806(1)	O _{ax} ,O _{ax} O _{eq} ,O _{ax} N _{eq} ,N _{eq} N _{eq} ,O _{eq}	83.4(1) ^f 94.4(1) 174.6(1) 90.9(1, 2, 3) 88.6(1) ^d 175.5(1)	428
Cu(phepy)(NO ₃) ₂ (dark green)	m P ₂ ₁ /c 4	11.1956(2) 14.3477(12) 16.2180(15)	123.36(1)	CuO ₃ N ₃	N _{eq} (NO ₃)O _{eq} O ₂ NO _{ax} (NO ₃)O _{ax}	1.998(3, 85) 1.968(3) 2.213(4) 2.533(5)	N _{eq} ,N _{eq} N _{eq} ,O _{eq} N _{eq} ,O _{ax} O _{eq} ,O _{ax}	79.2(1) ^e 80.0(1) 158.8(1) 98.5(1) 161.2(2) 97.1(2, 9, 5) 55.9(1) ^f 92.1(1) 147.9(1)	429
[Cu(bimpa)(NO ₃)(MeOH)]NO ₃ (not given)	or P _{bc} _a 8	16.136(5) 15.316(5) 18.901(3)		CuO ₃ N ₃	N _{eq} (NO ₃)O _{eq} (NO ₃)O _{ax}	1.949(5, 4) 2.102(5) 1.982(4) 2.602(5)	O _{ax} ,O _{ax} N _{eq} ,N _{eq} N _{eq} ,O _{ax}	82.5(2, 2) ^f 164.6(2) 89.6(2, 3, 6) 109.3(2, 9, 2)	430

[Cu(asp)(phen)(H ₂ O)]·4H ₂ O (blue)	or P ₂ , 2 ₁ , 2 ₁ 4	23.852(2) 11.731(1) 6.969(1)	CuO ₃ N ₃	phenN _{eq} aspN _{eq} aspO _{ax} H ₂ O _{ax}	2.002(—, 18) 1.989 2.560 2.446	N _{eq} , N _{eq} N _{eq} , O _{eq} N _{eq} , O _{ax} O _{eq} , O _{ax} O _{ax} , O _{ax}	176.4(1) 100.3(1) 82.0(1) ^e 84.1(1) ^e 93.8(1) 174.3(1) 78.2(1) ^e 93.4(1, 11.1) 87.9(1, 4.2) 169.2(1)	433
[Cu(terpy)(odac)]·2H ₂ O (green)	m P ₂ /c 4	10.681(6) 22.44(1) 11.534(6)	CuO ₃ N ₃	terpyN _{eq} odacO _{eq} odacO _{ax}	2.004(4, 62) 2.070(5) 2.257(5, 55)	N _{eq} , N _{eq} O _{eq} , O _{ax} O _{ax} , O _{ax} N _{eq} , O _{ax} N _{eq} , O _{eq}	80.0(1, 0) ^e 73.4(2, 1.0) ^e 146.6(1) 97.5(2, 18.3) 99.9(4, 3) 170.0(2)	434
[Cu(opai)(ClO ₄) ₂]·0.5H ₂ O (deep blue)	m C ₂ /c 8	32.777(8) 9.395(2) 14.14(6)	CuO ₃ N ₃	N _{eq} O _{eq}	2.003(3, 27) 2.008(3)	N _{eq} , N _{eq} N _{eq} , O _{eq}	88.6(1) 176.9(1) 83.7(1) 174.5(1)	435
[Cu(ac)(C ₆ H ₂ 1N ₄ O ₂)]·0.5H ₂ O ^e (green)	tr P-1 4	15.873(15) 9.693(9) 13.104(22)	CuO ₃ N ₃	acO O N	1.979(3) 2.765(3) 2.527(3) 1.998(3, 39)	O, O N, N O, N	52.48(10) ^f 94.6 146.9 82.04(14) ^e 94.88(13) 168.28(14) 78.67(11) ^e 97.1(1, 17.1) ^f 55.74(10) ^f 94.8 150.5 82.68(13) ^e 96.41(13) 165.43(13) 77.27(11) ^e 96.5(1, 14.9)	436

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(17-aneN ₃ O ₂)(H ₂ O)](ClO ₄) ₂ (not given)	m P2 ₁ /n 4	12.350(27) 13.879(16) 15.954(30)	101.16(16)	CuO ₃ N ₃ H ₂ O _{eq} N _{eq} O _{ax}	2.02(5) 2.01(5.2) 2.49(4.19)	O _{ax} , O _{ax} O _{ax} , O _{eq} N _{eq} , O _{ax}	437
Cu(dacac)(N ₃) (violet blue)	or Pccn 8	32.289(4) 13.187(2) 8.679(1)		CuO ₃ N ₃ (N ₃)N _{eq} N _{eq} O _{eq} O _{ax}	1.950(4) 2.057(3.25) 1.925(3) 2.779(2.3)	N _{eq} , O _{ax} N _{eq} , N _{eq} N _{eq} , O _{eq}	438
[Cu(dipa) ₂](H ₂ O) (blue)	tr P-1 2	6.443(2) 9.524(4) 16.308(8)	101.20(4) 93.89(4) 104.51(3)	CuO ₃ N ₃ N _{eq} O _{eq} O _{ax} N _{ax}	2.049(2.7) 1.974(2.2) 2.518(2) 2.336(2)	O _{eq} , O _{ax} O _{ax} , O _{ax} N _{eq} , N _{eq} N _{eq} , O _{eq} N _{eq} , O _{ax} O _{eq} , N _{ax} O _{ax} , N _{ax}	439
[Cu(ottd)]·5H ₂ O (blue)	m P2 ₁ /c 4	11.19(3) 12.19(2) 16.22(6)	110.16(9)	CuO ₃ N ₃ O _{eq} N _{eq} O _{ax}	1.973(6) 2.150(7.44) 1.983(7) 2.414(6) 2.171(6)	N _{eq} , N _{eq} N _{eq} , N _{eq} N _{eq} , O _{ax} O _{eq} , N _{eq}	440

Cu(dacac)Cl (dark blue)	or P _{bca} 8	12.734(2) 11.965(2) 23.252(3)	CuO ₃ N ₂ Cl	O _{eq} N _{eq} Cl _{eq} O _{ax}	1.929(3) 2.044(3, 22) 2.261(1) 2.760(3, 7)	O _{eq} N _{eq} Cl _{eq} O _{ax}	85.4(3, 6) 149.5(1)	441
[Cu(pith)(H ₂ O)(ClO ₂) ₂ ·H ₂ O] (deep green)	m P ₂ /c 4	9.212(7) 14.439(6) 15.117(9)	CuO ₃ N ₂ S	N _{eq} S _{eq} H ₂ O _{eq} O ₃ ClO _{ax}	1.972(3, 29) 2.320(1) 1.931(2) 2.525(4, 35)	N _{eq} S _{eq} H ₂ O _{eq} O ₃ ClO _{ax}	97.05(8) 91.43(10, 9.53) 89.78(8, 4.16) 84.70(8) ^e 163.01(8) 82.56(10) ^e 96.54(10) 175.29(11) 87.90(12, 4.34) 175.68(11)	442
[Cu(fpsem)(H ₂ O)(ClO ₂) ₂ ·2H ₂ O] (green black)	m P ₂ /c 4	12.9615(5) 9.7638(4) 14.9964(8)	CuO ₃ N ₂ S	N _{eq} S _{eq} H ₂ O _{eq} O ₃ ClO _{ax}	1.975(3, 42) 2.278(1) 1.935(2) 2.652(5, 58)	N _{eq} S _{eq} H ₂ O _{eq} O ₃ ClO _{ax}	80.5(1) ^e 85.2(1) ^e 165.7(1) 96.2(1) 175.9(2) 82.1(1, 6.7) 98.1(1) 101.1(1, 2.2) 93.4(2, 5) 155.9(2)	443
[Cu(bpy)(tdac)(H ₂ O)]·4H ₂ O (light blue)	m P ₂ /c 4	11.136(7) 21.13(1) 8.999(6)	CuO ₃ N ₂ S	bpyN _{eq} tdacO _{eq} tdacS _{ax}	2.022(4, 3) 1.975(4, 2) 2.697(1)	bpyN _{eq} tdacO _{eq} tdacS _{ax}	93.2(2, 1.5) 172.3(2, 4) 80.2(1, 4) ^e	444

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(Me-pyrpy)Cl]ClO ₄ (blue)	or P2 ₁ 2 ₁ 2 ₁ 4	11.057(1) 12.643(1) 16.799(3)		H ₂ O _{ax} CuN ₃ O ₂ Cl	2.353(3)	102.5(1, 4, 1) 93.1(2) 87.8(1, 4, 7) 80.9(2) ^c 91.1(2, 1) 161.0(1)	445
[Cu(OH-pyrpy)Cl]ClO ₄ ·H ₂ O (blue)	or P2 ₁ 2 ₁ 2 ₁ 4	11.724(1) 11.628(1) 17.174(2)		N _{eq} CuN ₃ O ₂ Cl	2.106(5, 7) 1.930(4) 2.2248(14) 2.494(5, 28)	179.00(14) 91.32(14, 1.62) 79.73(23, 29) ^e 159.55(18) 78.24(20, 57) ^f 94.98(21, 8.91) 176.81(17)	445
[Cu(bz-pyrpy)Cl]ClO ₄ ^c (blue)	or P2 ₁ 2 ₁ 2 ₁ 8	11.724(1) 30.346(5) 17.871(2)		N _{eq} CuN ₃ O ₂ Cl	2.110(10, 3) 1.938(9) 2.204(3) 2.563(8) 2.664(8)	100.7(3, 5) 175.0(3) 87.9(2, 5) 79.5(4, 8) ^e 158.6(4) 77.1(3, 1.4) ^f 97.9(3, 10.1) 173.5(3) 99.5(3, 3) 177.7(4)	445

[Cu(pdto)(1-Meim)(ClO ₄) ₂](ClO ₄) (deep blue)	m P ₂ ₁ /c 4	8.996(2) 18.435(4) 13.687(3)	116.84(1)	CuN ₃ S ₂ O	imN _{cu} N _{eq} S _{eq} S _{ax} O ₃ ClO _{ax}	Cl _{eq} O _{ax}	2.196(4) 2.609(11,77)	Cl _{eq} O _{ax} N _{eq} N _{ax}	90.9(3,2,5) 80.6(4,1,2) ^c 161.1(4) 77.2(4,4) ^c 95.8(4,7,7) 178.3(4)	446
Cu(imdh)(NO ₃)(NCS) (green)	m P ₂ ₁ /n 4	9.641(2) 8.040(3) 25.480(6)	92.56(2)	CuN ₃ S ₂ O	SCN _{eq} N _{eq} S _{eq} S _{ax} O ₂ NO _{ax}	Cl _{eq} O _{ax}	1.968(9) 1.969(9,1) 2.370(4) 2.746(4) 2.70(1)	N _{eq} N _{eq} S _{ax} O _{ax} N _{eq} O _{ax} S _{eq} N _{eq}	84.0(1) ^c 99.2(3,1,6) 80.1(3) ^c 88.6(2) 89.5(4,3,7) 83.0(3) ^c 89.0(3) 174.6(3) 94.1(4,2) 171.9(4) 166.1(2)	447
[Cu(dtum)Br] ₂ ·Br·H ₂ O (not given)	m P ₂ ₁ /m 2	5.559(7) 10.009(11) 14.720(13)	90.0(1)	CuS ₃ N ₂ Br	N _{eq} S _{eq} S _{ax} Br _{ax}	N _{eq} O _{ax}	2.03(2) × 2 2.340(6) × 2 2.576(9) 3.130(5)	N _{eq} S _{eq} N _{eq} S _{ax} N _{eq} Br _{ax} N _{eq} N _{eq} S _{eq} Br _{ax} S _{eq} S _{ax} S _{ax} Br _{ax} S _{ax} Br _{ax}	85.1(5) ^c 170.6(5) 102.2(5) 80.6(5) 95.4(5) 86.8(2) ^c 90.3(2) 92.9(2) 175.7(2)	448
[Cu(nc) ₂ (H ₂ O) ₂ F ₂]·4H ₂ O (blue)	tr P-1 1	8.895(1) 8.008(1) 7.222(1)	110.57(1) 103.11(1) 90.36(1)	CuF ₂ O ₂ N ₂	N _{eq} F _{eq} H ₂ O _{ax}	N _{eq} O _{ax} F _{eq} O _{ax}	2.009(3) × 2 1.911(3) × 2 2.560(4) × 2	N _{eq} F _{eq} O _{ax} N _{eq} F _{eq} O _{ax}	90.7(5) 87.1(5) 90.4(5)	449

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(nc) ₂ (H ₂ O) ₂ (BF ₄) ₂ ·2H ₂ O (blue)]	tr F-1 4	6.773(0) 16.483(1) 19.340(2)	98.114(2) 92.973(2) 92.539(2)	CuF ₂ O ₂ N ₂	N _{eq} H ₂ O _{eq} F _{ax}	O _{eq} N _{eq} O _{eq} F _{ax} N _{eq} F _{ax}	450 87.9(5) 88.8(5) 88.9(5)
Cu(bzph) ₂ (BF ₄) ₂ (green brown)	m P2 ₁ /c 2	13.540(3) 7.772(1) 12.858(2)	109.00(2)	CuF ₂ O ₂ N ₂	N _{eq} O _{eq} F _{ax}	O _{eq} N _{eq} N _{eq} F _{ax} O _{eq} F _{ax}	451 86.9(1) ^d 81.6(1) 96.8(1)
Cu(mesim) ₂ (BF ₄) ₂ (not given)	m C2/c 4	13.844(4) 11.251(1) 15.753(4)	111.39(2)	CuF ₂ N ₂ S ₂	N _{eq} S _{eq} F _{ax}	S _{eq} N _{eq} S _{eq} F _{ax} N _{eq} F _{ax} N _{eq} N _{eq} F _{ax} F _{ax}	452 84.2(2) ^e 95.8(2) 90.0(1, 7.3) 180 90.0(2, 1.1) 180 180
Cu(deam) ₂ (H ₂ O) ₂ Cl ₂ (blue green)	tr P-1 1	5.42(1) 6.45(1) 9.31(1)	74.5(2) 80.4(3) 84.7(3)	CuO ₂ N ₂ Cl ₂	H ₂ O _{eq} N _{eq} Cl _{ax}		453 not given
[Cu(9-Meop) ₂ (H ₂ O) ₂ Cl ₂]·H ₂ O (deep blue)	m C2/c 4	16.858(4) 8.541(4) 14.293(11)	91.02(5)	CuO ₂ N ₂ Cl ₂	N _{eq} H ₂ O _{eq} Cl _{ax}	N _{eq} Cl _{ax} O _{eq} Cl _{ax} N _{eq} O _{eq}	454 88.46(5) 86.67(5) 87.27(5)
Cu(hemq) ₂ (H ₂ O) ₂ Cl ₂ (green)	m C2/c 4	19.416(2) 10.086(1) 13.524(2)	111.17(1)	CuO ₂ N ₂ Cl ₂	N _{eq} H ₂ O _{eq} Cl _{ax}	O _{eq} Cl _{ax} N _{eq} O _{eq} N _{eq} Cl _{ax}	455 92.0 90.8(1) 84.5(1)
[Cu(ctr) ₂ Cl ₂]·2H ₂ O (not given)	tr P-1 1	6.915(3) 8.765(5) 6.845(3)	100.55(4) 108.50(4) 106.58(5)	CuO ₂ N ₂ Cl ₂	Cl _{eq} N _{eq} O _{ax}	N _{eq} Cl _{eq} N _{eq} O _{ax} Cl _{eq} O _{ax}	131 90.8(1) 73.1(1) ^e 89.2(1)
Cu(mnim) ₂ Cl ₂ (deep blue)	m P2 ₁ /a 2	7.685(2) 13.587(4) 7.796(2)	114.80(2)	CuO ₂ N ₂ Cl ₂	Cl _{eq} N _{eq} O _{ax}	Cl _{eq} N _{ax} N _{ax} N _{ax} Cl _{eq} Cl _{eq}	456 90.0(2, 4) 180 180

Cu(cyt) ₂ Cl ₂ (light blue)	m P2 _i /c 4	8.359(2) 13.744(5) 13.660(4)	128.00(1)	CuO ₂ N ₂ Cl ₂	Cl _{eq} N _{eq} O _{ax}	2.287(-, 20) 1.96(1, 1) 2.81(-, 7)	N _{eq} O _{ax}	52.9(6, 9) ^f 107.6(6) 146.6(6)	457		
										Cl _{eq} ^a , O _{ax} N _{eq} , Cl _{eq} N _{ax} , N _{eq} Cl _{eq} , Cl _{eq}	91.0(4, 9, 3) 91.9(5, 9) 90.5(5, 1, 9) 165.7(3)
Cu(sim) ₂ Cl ₂ (brownish yellow)	or Pbcn 4	19.530(4) 8.040(1) 16.006(5)		CuO ₂ N ₂ Cl ₂	Cl _{eq} O _{eq} N _{ax}	2.29(2) × 2 2.78(2) × 2 2.00(2) × 2	Cl _{eq} , Cl _{eq} Cl _{eq} , N _{ax} N _{ax} , N _{ax}	125.0(1, 1) 97.0(1, 1, 2) 152.0(1, 1)	458		
Cu(atr) ₂ Cl ₂ (brownish yellow)	m C2/c 4	23.465(3) 8.059(1) 15.427(1)	106.19(1)	CuO ₂ N ₂ Cl ₂	Cl _{eq} O _{eq} N _{ax}	2.28(2) × 2 2.75(2) × 2 2.00(2) × 2	Cl _{eq} , Cl _{eq} Cl _{eq} , N _{ax} N _{ax} , N _{ax}	124.0(1, 1) 98.0(1, 1, 2) 148.0(1, 1)	458		
[Cu(pmttz) ₂ Cl ₂](pmttz) (not given)	tr P-1 2	12.190(3) 15.970(3) 12.092(3)	111.45(2) 101.88(3) 95.93(3)	CuO ₂ N ₂ Cl ₂	Cl _{eq} O _{eq} N _{ax}	2.268(5, 12) 2.917(9) 2.757(9) 2.013(11, 3)	Cl _{eq} , Cl _{eq} Cl _{eq} , N _{ax} N _{ax} , N _{ax}	157.5(1) 91.4(3, 3, 2) 165.4(4)	459		
[Cu(hys) ₂ Cl ₂]·2H ₂ O (blue)	m P2 ₁ 2	5.189(1) 16.988(3) 11.482(2)	93.57(1)	CuO ₂ N ₂ Cl ₂	N _{ax} N _{eq} Cl _{ax}	1.989(10, 14) 1.949(9, 5) 2.823(6) × 2	O _{eq} , Cl _{ax} N _{eq} , Cl _{ax} N _{eq} , O _{eq}	90.1(4, 4, 0) 90.0(4, 8, 8) 84.8(4, 1) ^d 94.9(4, 5) 176.1(4) 169.0(4)	460		
[Cy(orm) ₂ Cl ₂]·2H ₂ O (royal blue)	m P2 ₁ 2	5.218(3) 15.636(2) 12.058(4)	93.37(2)	CuO ₂ N ₂ Cl ₂	O _{eq} N _{eq} Cl _{ax}	1.958(4, 12) 1.976(5, 13) 2.906(2, 11)	O _{eq} , N _{eq} O _{eq} , Cl _{ax} N _{eq} , Cl _{ax} O _{ax} , O _{eq} Cl _{ax} , Cl _{ax} N _{eq} , N _{eq}	85.0(11, 4) ^e 95.2(11, 9) 90.0(10, 1, 5) 90.0(10, 3, 7) 177.6(12) 178.9(3) 176.1(15)	461		
[Cu(orm) ₂ Cl ₂]·2H ₂ O (deep blue)	m P2 _i /c 2	5.18 15.57 11.90	93.3	CuO ₂ N ₂ Cl ₂	O _{eq} N _{eq} Cl _{ax}	1.976(5) × 2 1.992(6) × 2 2.89(2) × 2	O _{eq} , N _{eq}	87.7 ^e	462		
Cu(sc) ₂ Cl ₂ (blue)	or Pn2 ₁ a 4	6.90 10.22 8.26		CuO ₂ N ₂ Cl ₂	O _{eq} N _{eq} Cl _{ax}	1.97 × 2 1.99 × 2 2.84 × 2	N _{eq} , O _{eq} O _{eq} , Cl _{ax} N _{eq} , Cl _{ax}	83.0 ^e 89.1 86.5	463		

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	pyN _{eq} O _{eq} Cl _{ax} pyN _{eq} O _{eq} Cl _{ax} imN _{eq} O _{eq} Cl _{ax} Cl _{eq} N _{eq} O _{ax}	C _u -L [Å]	L-Cu-L [°]	Ref.
Cu(2, 4, 6-Cl ₃ ph) ₂ (py) ₂ (reddish brown)	tr	9.576(6)	125.61(8)	CuO ₂ N ₂ Cl ₂	pyN _{eq}	2.036(4) × 2	89.0(1)	187
	P-1	9.017(7)	101.73(4)		O _{eq}	1.911(2) × 2	89.0(2)	
	1	9.129(7)	94.02(6)		Cl _{ax}	2.916(4) × 2	77.0(1) ^c	
Cu(2, 4, 6-Cl ₃ ph) ₂ (py) ₂ (brown)	tr	8.292(2)	85.91(2)	CuO ₂ N ₂ Cl ₂	pyN _{eq}	2.036(3) × 2	90.8(1)	464
	P-1	9.002(2)	72.44(2)		O _{eq}	1.909(2) × 2		
	1	9.587(2)	63.48(2)		Cl _{ax}	2.912(1) × 2		
[Cu(2, 4, 6-Cl ₃ ph) ₂ (im) ₂]-H ₂ O (yellow brown)	or	11.711(5)		CuO ₂ N ₂ Cl ₂	imN _{eq}	1.98(1) × 2	89.7(6)	465
	Pbnb	24.829(6)			O _{eq}	1.97(1) × 2	75.1(6) ^c	
	4	8.257(4)			Cl _{ax}	2.83(1) × 2	88.2(6)	
Cu(18-aneO ₄ N ₂)Cl ₂ (not given)	m	12.209(19)		CuO ₂ N ₂ Cl ₂	Cl _{eq}	2.311(2, 27)	76.2(2, 1) ^e	466
	P2 ₁ /c	10.813(16)	93° 39'(8')		N _{eq}	2.035(6, 5)	103.3(2, 1.8)	
	4	12.588(5)			O _{ax}	2.732(5, 23)	90.0(2, 2.3)	
							90.0(1, 8.1)	
							164.4(2)	
[Cu(9-aneN ₂ Sac)Cl]-ClO ₄ (blue)	m	13.808(2)		CuO ₂ N ₂ ClS	N _{eq}	2.031(5, 21)	96.97(7)	467
	P2 ₁	8.195(2)	104.68(2)		O _{eq}	2.058(4)	97.2(2)	
	2	8.702(2)			Cl _{eq}	2.241(2)	174.0(1)	
					O _{ax}	2.554(4)	94.2(2)	
					S _{ax}	2.635(2)	92.5(1)	
							82.9(1) ^c	
							86.6(1)	
					155.9(1)			
					110.0(1)			
					88.0(2)			
					74.6(2) ^c			
					84.1(2)			
					162.8(2)			
					81.7(2) ^c			
					O _{eq} O _{ax}	90.6(2)		

[Cu(mesim) ₂ (H ₂ O) ₂](NO ₃) ₂ (dark green)	m P ₂ /n 2	10.451(3) 13.430(5) 8.757(4)	109.24(3)	CuO ₂ N ₂ S ₂	N _{eq} S _{eq} H ₂ O _{ax}	1.934(2) × 2 2.449(1) × 2 2.440(2) × 2	N _{eq} , S _{eq} S _{eq} , O _{ax} N _{eq} , O _{ax}	84.11(7) ^c 95.89(7) 90.0(5, 5.8) 90.0(8, 9)	468
Cu(mceam) ₂ (ClO ₄) ₂ (purple)	m P ₂ /c 2	5.756(2) 16.353(4) 8.592(3)	93.31(3)	CuO ₂ N ₂ S ₂	N _{eq} S _{eq} O ₃ ClO _{ax}	1.977(2) × 2 2.366(1) × 2 2.599(2) × 2	S _{eq} , N _{eq} O _{ax} , N _{eq} S _{eq} , O _{ax}	86.8(1) ^e 84.8(1) 100.7(1)	469
Cu(prepy) ₂ (ClO ₄) ₂ (not given)	tr P-1 1	8.955(1) 8.794(1) 8.517(1)	104.37(1) 109.723(8) 78.86(1)	CuO ₂ N ₂ S ₂	N _{eq} S _{eq} O ₃ ClO _{ax}	2.015(3) × 2 2.366(1) × 2 2.501(3) × 2	S _{eq} , N _{eq} S _{eq} , O _{ax} N _{eq} , O _{ax}	83.8(1) ^e 85.3(1) 85.7(1)	470
Cu(prcim) ₂ (ClO ₄) ₂ (brown)	m P ₂ /n 2	10.131(1) 8.229(1) 13.663(1)	90.40(1)	CuO ₂ N ₂ S ₂	N _{eq} S _{eq} O ₃ ClO _{ax}	1.940(6) × 2 2.397(2) × 2 2.594(6) × 2	S _{eq} , N _{eq} S _{eq} , O _{ax} N _{eq} , O _{ax}	84.1(2) ^e 100.1(1) 95.0(2)	471
Cu(sbzh) ₂ (NO ₃) ₂ (not given)	m P ₂ /a 2	22.056(4) 5.774(2) 7.756(3)	88.28(2)	CuO ₂ N ₂ S ₂	N _{eq} S _{eq} (NO ₃)O _{ax}	1.978(6) × 2 2.293(2) × 2 2.617(5) × 2	S _{eq} , N _{eq}	86.1(1) ^e	472
Cu(acatr) ₂ (SCN) ₂ (brownish green)	tr P-1 1	9.057(3) 6.442(2) 7.602(3)	75.79(2) 73.16(2) 77.82(2)	CuO ₂ N ₂ S ₂	N _{eq} O _{eq} NCS _{ax}	1.932(3) × 2 2.019(3) × 2 2.864(2) × 2	N _{eq} , O _{eq} N _{eq} , S _{ax} O _{eq} , S _{ax}	89.1(1) ^d 90.0(1) 90.3(1)	304
Cu(bhdhx)(NO ₃) ₂ (green)	or Pbcn 4	14.351(5) 8.554(3) 13.057(4)		CuO ₂ N ₂ S ₂	O ₂ NO _{eq} S _{eq} N _{ax}	2.280(3) × 2 2.495(1) × 2 1.933(4) × 2	S _{eq} , S _{ax} S _{eq} , O _{eq}	83.6(1) ^e 90.6(1) 99.41(9) 169.38(8)	473
Cu(14-aneS ₂ N ₂)(ClO ₄) ₂ (dark purple)	m P ₂ /n 4	15.008(5) 8.606(3) 17.967(4)	113.79(2)	CuO ₂ N ₂ S ₂	N _{eq} S _{eq} O ₃ ClO _{ax}	2.024(6, 7) 2.296(2, 2) 2.459(7) 2.697(12)	N _{eq} , N _{eq} N _{eq} , S _{eq} S _{eq} , O _{eq}	95.1(2) ^d 87.7(2, 1) ^e 175.1(2, 2) 89.28(7) ^d	474
[Cu(quisp)(NO ₃)(H ₂ O)]NO ₃ (green)	m P ₂ /n 4	6.480(3) 17.430(5) 21.248(5)	97.81(3)	CuO ₂ N ₂ S ₂	H ₂ O _{eq} N _{eq} S _{eq} S _{ax}	2.030(6) 1.990(7, 4) 2.343(3) 2.507(3)	S _{eq} , S _{eq} S _{eq} , S _{ax} N _{eq} , S _{ax} S _{eq} , O _{eq}	103.5(2) ^d 83.5(3) ^e 100.2(3) 160.3(2)	475

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å]			α [°]			Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
		b [Å]	c [Å]	β [°]	γ [°]						
[Cu(pyos)(ClO ₄) ₂ ClO ₄] (green)	m P2 ₁ /c 4	14.388(2) 9.990(5) 16.076(3)		107.919(12)	O ₂ , NO _{ax}	2.677(7)	S _{eq} , O _{ax}	2.014(6, 7)	80.9(2)		476
							O _{eq} , N _{eq}				
							O _{eq} , O _{ax}				
							N _{eq} , S _{eq}				
							O _{eq} , N _{eq}				
							O _{eq} , S _{ax}				
							N _{eq} , O _{ax}				
							S _{ax} , O _{ax}				
							N _{ax} , N _{eq}				
							N _{eq} , N _{eq}				
S _{eq} , S _{eq}											
O _{ax} , O _{ax}											
N _{eq} , S _{eq}											
N _{eq} , O _{ax}											
S _{eq} , O _{ax}											
N _{eq} , O _{ax}											
S _{eq} , O _{ax}											
Cu(eiphim) (brown red)	m P2 ₁ /c 4	8.887(3) 28.825(1) 10.168(4)		106.51(2)	S _{eq}	2.548(3, 60)	84.0(1) ^f		477		
					O _{eq}						
					N _{ax}						
					S _{eq} , S _{eq}						
					S _{eq} , O _{eq}						
S _{eq} , N _{ax}											
O _{eq} , O _{eq}											
O _{eq} , N _{ax}											
N _{ax} , N _{ax}											
N _{ax} , N _{ax}											
Cu(bzpy) ₂ Br ₂ (dark green)	m P2 ₁ /n 2	7.401(3) 20.690(7) 7.403(3)		105.29(3)	O _{eq}	2.407(4) × 2	74.6(2) ^f	303			
					N _{eq}						
					Br _{ax}						
					O _{eq} , Br _{ax}						
					N _{eq} , Br _{ax}						

[Cu(gly) ₂ Br ₂][Li ₂ (gly) ₂ ·2H ₂ O (royal blue)	m P2 ₁ /n 2	9.011(1) 5.061(1) 20.659(3)	91.54(5)	CuO ₂ N ₂ Br ₂	O _{eq}	1.984(3) × 2	N _{eq} , O _{eq}	97.1(2) ^d	478
					N _{eq}	1.956(4) × 2	N _{eq} , Br _{ax}	90.5(1)	
					Br _{ax}	2.987(1) × 2	O _{eq} , Br _{ax}	91.4(1)	
Cu(bzimox)Br ₂ (not given)	or Pccn 4	8.940(1) 10.634(1) 20.455(2)		CuO ₂ N ₂ Br ₂	N _{eq}	1.987(7) × 2	N _{eq} , Br _{ax}	89.5(2, 1.3)	479
					O _{eq}	2.743(6) × 2	O _{eq} , Br _{ax}	95.2(1, 4.4)	
					Br _{ax}	2.465(1) × 2	N _{eq} , O _{eq}	67.2(3) ^e	
Cu(18-aneO ₄ N ₂)Br ₂ (not given)	m C2/c 4	15.895(6) 10.593(17) 10.336(20)	98°37'(6')	CuO ₂ N ₂ Br ₂	O _{eq}	2.772(8) × 2	N _{eq} , O _{eq}	75.5 ^f	480
					N _{eq}	2.008(9) × 2	O _{eq} , Br _{ax}	103.9(3)	
					Br _{ax}	2.476(3, 38)	N _{eq} , Br _{ax}	90.0(2, 8.6)	
Cu(esbim) ₂ Cl ₂ (blue green)	m P2 ₁ /c 2	9.707(1) 14.977(2) 8.525(1)	112.03(1)	CuN ₂ Cl ₂ S ₂	Cl _{eq}	2.322(2) × 2	Cl _{eq} , S _{eq}	85.70(6)	481
					S _{eq}	3.001(2) × 2	Cl _{eq} , N _{ax}	91.2(2)	
					N _{ax}	2.000(5) × 2	S _{eq} , N _{ax}	76.2(1) ^f	
Cu(10-aneN ₂ S ₂)Cl ₂ (brown)	or Pbca 8	12.927(3) 18.933(5) 14.469(2)		CuN ₂ Cl ₂ S ₂	Cl _{eq}	2.395(1)	Cl _{eq} , S _{eq}	86.31(3, 69)	482
					S _{eq}	2.457(1)	N _{ax} , N _{ax}	159.99(4, 1.13)	
					N _{ax}	2.886(1)	Cl _{eq} , Cl _{eq}	168.6(1)	
Cu(mindth)Cl ₂ (green)	or Pbca 8	17.917(2) 14.502(4) 12.905(2)		CuN ₂ Cl ₂ S ₂	S _{eq}	2.970(1)	Cl _{eq} , Cl _{eq}	112.08(3)	483
					N _{ax}	1.952(2, 3)	S _{eq} , S _{eq}	76.81(3) ^d	
					Cl _{eq}	2.475(3, 87)	Cl _{eq} , N _{ax}	92.5(1, 1.8)	
					S _{eq} , N _{ax}	76.3(1, 4) ^e	94.5(1, 1)		
					Cl _{eq} , N _{eq}	93.9(1, 15)			
					Cl _{eq} , Cl _{eq}	2.475(3, 87)	108.03(8)		
					S _{eq}	2.714(3, 91)	86.3(1, 1.6)		
					N _{ax}	1.971(7, 9)	165.1(1, 1.7)		
					Cl _{eq} , N _{ax}		93.4(2, 2.0)		
					S _{eq} , S _{eq}		79.6(1) ^e		
					S _{eq} , N _{ax}		80.2(3.7) ^e		
							90.9(2.4)		
							168.5(3)		

TABLE I (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
Cu(bhdhx)Cl ₂ (not given)	or	12.738(3)		CuN ₂ Cl ₂ S ₂	2.589(1)	93.7(1, 1.1)	484
	Pbca	17.218(4)			2.371(1)	79.8(1, 2.1) ^c	
	8	13.894(7)			2.523(1)	91.3(1, 2.0)	
{Cu(18-aneN ₂ S ₂)Cl ₂ }.0.5tol ^c (light green)	tr	12.828(6)	93.15(2)	CuN ₂ Cl ₂ S ₂	2.165(10, 95)	176.6(2)	485
	P-1	21.963(4)	109.35(2)		2.293(6, 93)	90.15(3) ^d	
	4	10.872(3)	82.32(2)		2.594(5)	90.0(2, 2.3)	
					2.951(6)	164.4(4)	
						178.8(2)	
						89.98(3, 7.78)	
[Cu(imdth)(BF ₄)Cl] (green)	m	13.975(5)		CuN ₂ Cl ₂ S ₂	2.185(10, 75)	176.1(2)	447
	P2 ₁ /n	8.347(3)	116.34(2)		2.402(6, 7)	90.0(2, 2.6)	
	4	17.156(6)			2.547(5)	89.83(40, 48) ^d	
					2.932(5)	179.1(1)	
						90.0(4, 5.4)	
						169.2(5)	

S_{ax}, F_{ax}	163.0(1)
S_{eq}, N_{eq}	82.8(1) ^c
	89.4(1)
S_{eq}, F_{ax}	80.9(1)
N_{eq}, N_{eq}	171.8(1)
N_{eq}, F_{ax}	88.1(1, 2.6)

^a Where more than one chemically equivalent distance or angle is present, the mean value is tabulated. The first number in parenthesis is c.s.d., the second is a maximum deviation from the mean value. ^b The chemical identity of coordinated atom/ligand is specified in these columns. ^c There are two crystallographically independent molecules. ^d Six-membered metalocyclic ring. ^e Five-membered metalocyclic ring. ^f Four-membered metalocyclic ring. ^g There are three crystallographically independent molecules. ^h Eight-membered metalocyclic ring. ⁱ Seven-membered metalocyclic ring.

The ligands used range from unidentate to hexadentate, accounting for the variations in distortion about the Cu(II) atoms. The mean Cu–L_{eq} bond distances for the unidentate ligands increases in the sequence: 1.915 Å(F) < 1.99 Å(OL) < 2.02 Å(NL) < 2.31 Å(Cl) < 2.41 Å(Br), which follows the covalent radius of the coordinated atom. The mean Cu–L_{ax} bond distances are somewhat longer than those of the Cu–L_{eq} bond distances and increases in the order: 2.265 Å(H) < 2.46 Å(H₂O) < 2.47 Å(NL) < 2.56 Å(OL) < 2.56 Å(F) < 2.73 Å(Br) < 2.85 Å(Cl) < 2.97 Å(SL). For bidentate ligands, Cu–L_{eq} vs. Cu–L_{ax} bond distances are: 2.05 vs. 2.32 Å(N) < 2.06 vs. 2.45 Å(O). For tridentate ligands the values are: 2.00 vs. 2.325 Å(O), 2.065 vs. 2.31 Å(N) and 2.44 vs. 2.44 Å(S). The most common donor atoms are O and N.

In Table IA are summarized Cu–L bond distances for the trigonal-bipyramidal, square-pyramidal and tetragonal-bipyramidal configuration about Cu(II) atoms. In the series of trigonal-bipyramidal configuration, the Cu–L_{eq} bond distances are somewhat longer than those of Cu–L_{ax} bond distances; in the remaining two configurations (square-pyramidal and tetragonal-bipyramidal) the opposite is observed, Cu–L_{eq} bond distances are somewhat shorter than those of Cu–L_{ax}. The mean Cu–L_{eq} bond distance increases in the sequence: square-pyramidal < tetragonal-bipyramidal < trigonal-bipyramidal and Cu–L_{ax}: trigonal-bipyramidal < square-pyramidal < tetragonal-bipyramidal.

Other than the ligand donor atoms of the same type, a wide range of hetero-donors also exist. There are bidentates: O+N; O+S; N+S and even O+Cl; tridentates: 2O+N; 2O+S; O+2N; 2N+S and O+2S; tetradentates: 3O+N; 2O+2N; O+3N and 2N+2S; pentadentates: 3O+2N; 2O+3N; O+4N; 4N+S; 2N+3S; 2O+2N+S; O+2N+2S; and hexadentates: 4O+2N; 3O+3N; 2O+4N; 2N+4S and 2O+2N+2S donor atoms.

TABLE IA Summary of the mean Cu–L_{eq} vs. Cu–L_{ax} bond distances

<i>Coord. atom</i>	<i>Trigonal-Bipyramid (ref. 2)</i>	<i>Square-Pyramid (ref. 2)</i>	<i>Tetragonal-Bipyramid</i>
O	2.13 vs. 1.96 2.23 vs. 2.01	1.97 vs. 2.33 2.01 vs. 2.32	1.99 vs. 2.51 2.06 vs. 2.45 ^a
N	2.04 vs. 2.00 2.12 vs. 1.995 2.06 vs. 2.01	1.97 vs. 2.26 2.015 vs. 2.22 2.005 vs. 2.25	2.02 vs. 2.47 2.05 vs. 2.32 ^a 2.065 vs. 2.31 ^b
F		1.89 vs. 2.50	1.915 vs. 2.565
Cl	2.32 vs. 2.25	2.285 vs. 2.51	2.31 vs. 2.85
Br	2.50 vs. 2.43	2.41 vs. 2.63	2.41 vs. 2.73

^aBidentate ligands. ^bTridentate ligands.

An interesting feature of the chelate rings is that the bite angle (the angles formed by two donor atoms and the central Cu(II) atom) may vary widely. The effect of both electronic and steric hindrances of the ligands can be seen in the opening of the L–Cu–L angles of the respective metallo-cyclic rings. The metallo-cyclic intra-ligand O–Cu–O ring angle opens with increasing size of the ring, for example: 54.0(7.5; 3.5)° (four-membered) < 77.5(6.0; 8.0)° (five-membered) < 89.8(4.8; 3.3)° (six-membered) < 90.5° (seven-membered). The first number in parenthesis is the mean minus the smallest value and the second is the largest value minus the mean. The N–Cu–N rings follow this trend: 78.0(5.0; 2.5)° (five-membered, unsaturated ligands) < 84.5(4.5; 5.5)° (five-membered, saturated ligands) < 90.2(4.8; 6.3)° (six-membered) < 98.0° (seven-membered). The values for S–Cu–S rings are: 86.0(6.4; 3.9)° (five-membered) < 90.5(13.7; 13.0)° (six-membered) and for O–Cu–N rings: 52.0(1.5)° (four-membered) < 76.0(8.8; 3.5)° (five-membered, unsaturated ligands) < 84.0(4.0; 6.0)° (five-membered, saturated ligands) < 90.0(6.0; 5.2)° (six-membered) < 99.6° (seven-membered) < 101.8° (eight-membered).

Table IB summarizes L–Cu–L bond angles of the metallo-cyclic rings for tetra-, penta-, hexa- and hepta-coordinate Cu(II) derivatives. It can be seen that the L–Cu–L bond angles of the respective metallo-cyclic rings increase

TABLE IB Summary of the L–Cu–L angles[°] of the metallo-cyclic rings^a

<i>Metallo-cyclic ring</i>	<i>4-Coordination</i> ²	<i>5-Coordination</i> ²	<i>6-Coordination</i>	<i>7-Coordination</i>
4-membered				
O–Cu–O	64.3	57.6	54.0	51.0
O–Cu–N	—	56.9	52.1	
N–Cu–N	67.4	62.0	—	
5-membered				
O–Cu–O	85.0	84.0	77.5	72.0
O–Cu–N	—	83.0	76.0	
N–Cu–N	79.5 ^b	80.0 ^b	77.5 ^b	69 ^b
S–Cu–S	85.5 ^c	84.5 ^c	84.0 ^c	80 ^c
S–Cu–O	92.0	88.5	85.0	
S–Cu–O	—	82.5	80.0	
S–Cu–N	—	85.5	82.5	
6-membered				
O–Cu–O	93.0	92.0	90.0	
O–Cu–N	—	92.0	90.0	
N–Cu–N	93.3	95.0	90.5	
N–Cu–S	—	91.0	91.5	
7-membered				
N–Cu–N	96.0	98.0	98.0	

^aThe mean values are tabulated. ^bUnsaturated ligands. ^cSaturated ligands.

with increasing covalent radius of the donor atoms and consequently decreasing with coordination number about Cu(II). In general, the mean Cu–L bond distance increases with coordination number, which means that, at greater Cu–L bond lengths, the angle must be smaller, and at shorter it must be larger. The data given in Table IA and IB follow this trend.

Some hexa-coordinate Cu(II) compounds were studied at two different temperatures, $\text{Rb}_2[\text{Cu}(\text{H}_2\text{O})_6](\text{SO}_4)_2$,⁵ $[\text{Cu}(\text{H}_2\text{O})_6]\text{SiF}_6$,¹² $\text{K}_2\text{Pt}[\text{Cu}(\text{NO}_2)_6]$,⁷¹ $[\text{Cu}(\text{en})_3]\text{SO}_4$,^{85,86} $[\text{Cu}(\text{dien})_2](\text{NO}_3)_2$ ⁹⁴ and $\text{Cu}(\text{debt})(\text{NCS})_2$.¹⁰⁹ The mean Cu–L bond distances decrease with decreasing temperature. Notably, $[\text{Cu}(\text{en})_3]\text{SO}_4$ is hexagonal at 295K⁸⁵ but at 120K is triclinic.⁸⁶

There are several examples which contain two monomeric forms $[\text{Cu}(\text{ahmi})_6][\text{Cu}(\text{ahmiCl}_3)_2]$ with CuO_6 and CuCl_3O chromophores; $[\text{Cu}(\text{H}_2\text{O})_6][\text{Cu}(\text{C}_8\text{H}_8\text{O}_5)_2]$ ^{85,86} with CuO_6 , $[\text{Cu}(\text{C}_{22}\text{H}_{34}\text{N}_6)[\text{CuCl}_4]$ ¹¹ with CuN_6 and CuCl_4 chromophores, $[\text{Cu}(\text{F}_3\text{ac})_2(\text{py})_2][\text{Cu}(\text{py})_4(\text{F}_3\text{ac})_2]$ ^{137,150} with CuO_4N_2 and CuN_4O_2 chromophores; $[\text{Cu}(\text{croc})(\text{terpy}(\text{H}_2\text{O}))][\text{Cu}(\text{croc})(\text{terpy})]\cdot 4\text{H}_2\text{O}$ ⁴²² with the CuO_3N_3 and CuN_3O_2 chromophores.

Some hexacoordinate Cu(II) compounds exist in two isomeric forms differing mostly by degree of distortion involving both Cu–L distances and L–Cu–L angles. For example, $\text{Cu}(\text{NO}_3)_2(\text{apy})_2$ exists in monoclinic⁵¹ and triclinic⁵² crystal classes, with the ratios of $(\text{Cu}-L_{\text{eq}})/(\text{Cu}-L_{\text{ax}})$ (degree of tetragonality) at 0.78 and 0.81, respectively. Two monoclinic forms of $[\text{Cu}(\text{phen})_3](\text{ClO}_4)_2$ ⁹⁰ differ by unit cell dimensions (Cu–N distances and N–Cu–N angles are not given). Another two monoclinic forms, α - and β - $\text{Cu}(\text{pipc})_2(\text{H}_2\text{O})_2$ ¹⁵⁶ differ mostly by unit cell dimensions and by the ratio of $(\text{Cu}-L_{\text{eq}})/(\text{Cu}-L_{\text{ax}})$ with values of 0.835 and 0.76, respectively. The ratio in orthorombic²¹⁶ and monoclinic forms of $\text{Cu}(\text{Clac})_2(\text{Me}_4\text{en})$ are equal to 0.76. For the $\text{Cu}(\text{NO}_3)_2(\text{Et}_2\text{en})$ complexes which also exist in monoclinic²²⁴ and orthorombic²²⁵ forms, the ratio is 0.82 and 0.84, respectively. In another pair, monoclinic²²⁷ and orthorombic²²⁸ forms of $\text{Cu}(\text{NO}_3)_2(\text{Me}_4\text{en})$, the ratio is 0.82 and 0.77, respectively. A somewhat higher degree of tetragonality (0.80) was found for the triclinic brown complex than for the monoclinic green complex (0.83) of $[\text{Cu}(\text{papt})_2(\text{H}_2\text{O})_2](\text{BF}_4)_4$.³⁰⁰ For the $\text{Cu}(1,3\text{-bn})_2(\text{ClO}_4)_2$ complex,³¹⁷ which exists in two monoclinic forms, red violet and blue violet, the degree of tetragonality is 0.76 and 0.79, respectively.

Two crystallographically independent molecules differing by degree of distortion and coexisting in the same crystal have been found in several species.^{15,16,43,50,67,72,112,141,187,201,208,216,220b,236,240,280,359,376,391,408,445,485} In blue $[\text{Cu}(\text{en})_3]\text{Cl}_2\cdot 0.75\text{en}$ ⁸⁶ three such independent molecules are present.

Almost all Cu(II) complexes are blue or green. Exceptions are generally caused by strong ultraviolet charge-transfer bands tailing off into the blue

end of the visible spectrum and causing the substances to appear red or brown. There are examples, which are red^{102,173,222,249,261,266,277,301,317,337,359,372,374,376,377,385,401,409} as well as brown.^{91,112,113,187,213,220a,257,300,371,403,407,458,465,471,477,482} There are three examples^{17,64,68} which are colourless, and even two^{83,250} which are black.

Mononuclear hexacoordinate Cu(II) compounds "prefer" the monoclinic crystal class, then triclinic and orthorhombic.

HEPTACOORDINATE Cu(II) COMPOUNDS

Crystallographic and structural data for mononuclear Cu(II) compounds with coordination number seven are collected in Table II. The structures are tabulated by the increasing number of different coordinated atoms (ligands). There are eleven examples⁴⁸⁶⁻⁴⁹⁵ in which Cu(II) atoms are surrounded by seven donor atoms. There are only three types of donor atoms: O, N and Cl, which form a pentagonal bipyramidal geometry about Cu(II). There are no examples with unidentate ligands. The most prevalent structure involves Cu(II) surrounded by three ligands (one pentadentate and two unidentate) in the form of a distorted pentagonal bipyramid, the equatorial plane of which consists of a pentadentate ligand and apical positions are occupied by unidentate ligands.^{486,487,490-492,494,495} In $[\text{Cu}(\text{shb})]^{2+}$, the heptadentate N-donor ligand sbh creates a distorted pentagonal bipyramid about the Cu(II) atom.⁴⁸⁸ In $[\text{Cu}(\text{NO}_3)_3]^{2-}$, the Cu(II) is surrounded by four ligands (three bidentate and one unidentate) again in the form of a distorted pentagonal bipyramid.⁴⁸⁹ Finally, in $\text{Cu}(\text{NO}_3)_2\text{py}_3$ a distorted pentagonal bipyramid is formed by two bidentate NO_3 groups and three unidentate pyridine molecules.⁴⁹³

Inspection of the data in Table II reveals that Cu-L_{ax} bond distances except $[\text{Cu}(\text{dapsc})(\text{H}_2\text{O})\text{Cl}]^+$ (see Ref. 495) are shorter than the Cu-L_{eq} bond distances, consistent with the two apical sites of a pentagonal bipyramid being less sterically hindered than the five equatorial sites. The mean Cu-L_{ax} vs. Cu-L_{eq} bond distances are: 2.06 vs. 1.98 Å(NL); 2.40 vs. 2.21 Å(bidentate O donors); 2.43 vs. 1.995 Å(heptadentate N donors); 2.35 vs. 2.40 Å(Cl). The mean Cu-OH_{2(ax)} bond distance is 1.93 Å and Cu-O_{eq} bond distance (pentadentate) is 2.25 Å.

Another factor of interest is the sum of all interatomic distances around Cu(II), the observed values ranging from 14.96 to 15.97 Å with an average value 15.5 Å. The sum increasing from 15.0 Å in CuO_7 , through 15.1 Å in CuO_5N_2 to 15.9 Å in CuO_5Cl_2 , which corresponds well with the increasing size of the atoms of the coordinated ligands.

(lc) ₂ [Cu(NO ₃) ₃ Cl] (not given)	m P2 ₁ /n 4	CuO ₆ Cl	10.554(2) 24.402(7) 14.124(3)	100.02(2)	CuO ₆ Cl	O _{eq}	2.231(12,70)	O _{eq} , O _{eq} O _{eq} , Cl _{eq} O _{eq} , O _{ax}	53.9(4) ^c 152.9(3) 48.2(5, 4.2) ^c	489
							2.474(10)			
							2.3538(24) 2.123(7, 18)			
[Cu(15-crown-5)(MeCN) ₂][Cu ₃ Cl ₈] (orange)	tr P-1 1	CuO ₅ N ₂	9.762(2) 9.644(2) 9.353(2)	95.02(2) 97.80(2) 122.53(2)	CuO ₅ N ₂	O _{eq} MeCN _{ax}	2.24(2, 10) 1.95(1) × 2	O _{ax} , Cl _{ax} O _{ax} , O _{ax} O _{eq} , O _{eq}	72.3(7, 1.7) ^c 143.5(7, 2.9)	490
[Cu(15-crown-5)(MeCN)Cl] ₂ · [Cu ₂ Cl ₆] (orange)	m P2 ₁ /a 4	CuO ₅ NCl	14.732(5) 11.357(4) 13.520(3)	112.25(2)	CuO ₅ NCl	O _{eq} MeCN _{ax} Cl _{ax}	2.261(3, 99) 1.981(4) 2.209(2)	O _{eq} , N _{ax} N _{ax} , N _{ax} O _{eq} , O _{eq}	72.2(2, 2.2) ^f 142.8(3, 3.5) 86.5(3, 5.8) 93.5(3, 5.5) 179.2(2)	490
[Cu(15-crown-5)(MeCN)Cl]· [CuCl ₃] (red)	m P2 ₁ /c 4	CuO ₅ NCl	11.367(6) 13.521(7) 14.751(9)	112.40(4)	CuO ₅ NCl	O _{eq} N _{ax} Cl _{ax}	2.261(5, 90) 1.987(5) 2.212(2)	O _{eq} , O _{eq} O _{eq} , N _{ax} O _{eq} , Cl _{ax}	72.2(2, 1.9) ^c 143.0(2, 3.0) 86.5(2, 6.4)	491
[Cu(C ₁₄ H ₂₀ O ₅ Cl ₂) ₂]·CHCl ₃ (red)	m P2 ₁ /c 4	CuO ₃ Cl ₂	14.650(5) 11.268(7) 13.623(4)	106.94(3)	CuO ₃ Cl ₂	O _{eq} Cl _{ax}	2.285(5, 52) 2.248(2, 6)	O _{eq} , O _{eq} μCl ₁ , μCl ₁ O _{eq} , O _{eq}	72.3(6, 4.2) ^c 89.32(8)	492
Cu(NO ₃) ₂ (py) ₃ (not given)	m C2/c 4	CuO ₅ N ₃	12.783(6) 9.199(5) 16.055(8)	CuO ₅ N ₃ 108.35(3)	CuO ₅ N ₃	O _{eq} pyN _{eq} pyN _{ax}	2.154(7) × 2 2.732(9) × 2 2.064(9) 2.018(8, 2)	O _{eq} , O _{eq} O _{eq} , N _{eq} O _{eq} , N _{ax} N _{eq} , N _{ax} N _{ax} , N _{ax}	50.0(3) ^f 91.4(3) 141.3(3) 84.4(2) 134.2(2) 89.4(3, 4.2) 92.0(2) 176.0(3)	493

TABLE II (Continued)

Compound	Cryst. Cl. Space G. Z	a [Å] b [Å] c [Å]	α [°] β [°] γ [°]	Chromophore	Cu-L [Å]	L-Cu-L [°]	Ref.
[Cu(dapso)(H ₂ O) ₂ · (NO ₃) ₂ · H ₂ O (not given)]	m Pc(P2/c) 2	10.977(4) 13.930(3) 6.977(3)	104.55(3)	CuO ₄ N ₃	2.350(4) × 2 2.260(6, 5) 1.922(3) × 2	not given	494
[Cu(dapso)(H ₂ O)Cl]Cl · 2H ₂ O (pale yellow)]	tr P-1 2	7.903(4) 7.768(3) 18.477(12)	86.37(4) 84.90(5) 67.00(4)	CuO ₃ N ₃ Cl	2.341(4, 11) 2.051(5, 73) 2.269(5) 2.701(2)	not given	495
[Cu(dapso)(H ₂ O) _{2-x} Cl _x]Cl _{2-x} · N ₃ ^a (golden yellow)]	m I2/a 4	17.980(5) 13.192(2) 7.956(4)	99.95(3)	CuO ₃ N ₃ Cl	2.269(3) 2.172(5, 24) 1.976(8) 2.625(3)	not given	495
(AsPh ₄) ₂ [Cu(NO ₃) ₄] · 10CH ₂ Cl ₂ (not given)]	m P2 ₁ /n 2	10.714 16.753 14.082	92.58	CuO ₈	1.946(8, 27) × 4 2.733(9, 60) × 4	not given	496
[Cu(6-am) ₄](ClO ₄) ₂ (not given)]	tg P-4 1	10.60 7.72		CuO ₈	1.97 × 4 2.79 × 4	not given	497
[Cu(bet)] · (NO ₃) ₂ (blue violet)]	tg I4-2m 2	11.443(2) 11.672(2)		CuO ₈	1.948(3) × 4 2.833(4) × 4	51.3(3) ^e 84.1 90.2(4) 173.2(4)	498
Cu(prometone) ₂ (NO ₃) ₂ (green)]	or Pbcn 4	23.809(5) 8.855(2) 14.978(5)		CuO ₆ N ₂	2.75 × 2 2.00 × 2 1.99 × 2 2.65 × 2	not given	499

^a Where more than one chemically equivalent distance or angle is present, the mean value is tabulated. The first number in parenthesis is e.s.d., the second is a maximum deviation from the mean value. ^b The chemical identity of coordinated atom/ligand is specified in these columns. ^c The five-membered metallocyclic ring. ^d There are ten crystallographically independent molecules. ^e The four-membered metallocyclic ring.

One of the important shape characteristics for describing a pentagonal bipyramidal geometry is the ratio of bond distances $(M-L_{ax})/(M-L_{eq})$. The ratio ranges from 0.84 to 0.93 for the series in which the mean $Cu-L_{eq}$ bond distance is longer than that of $Cu-L_{ax}$; but in $[Cu(dapsc)(H_2O)Cl]^+$ where $Cu-L_{eq} < Cu-L_{ax}$ the ratios are 1.04 and 1.15.

However, we also note that any distortions from ideal geometry will be reflected in the value of the bond angles: $L_{ax}-M-L_{eq}$; $L_{ax}-M-L_{ax}$; $L_{eq}-M-L_{eq}$. In the ideal pentagonal bipyramid, the values of these angles are: $L_{ax}-M-L_{eq}$ 90° , $L_{ax}-M-L_{ax}$ 180° and $L_{eq}-M-L_{eq}$ 72 and 144° , respectively.

Inspection of the data in Table II reveals that $L_{eq}-Cu-L_{eq}$ angles in the pentagonal base differ from the ideal pentagonal angles of 72 and 144° respectively, by twenty degrees for the former and by ten degrees for the latter, the observed values ranging from 50.0 to 91.4° with an average value of 72.5° , and from 134.2 to 152.9° with an average value of 143° , respectively. Smaller $L_{eq}-Cu-L_{eq}$ angles occur between the intraligand coordinated atoms while the larger ones involve interligand bonded atoms. The $L_{ax}-Cu-L_{eq}$ bond angles ranging from 44 to 99° with an average value of 86° . The average value of 176° for $L_{ax}-Cu-L_{ax}$ angles (range from 165 to 180°) is again smaller than in the ideal pentagonal bipyramid (180°). These smaller/larger angles reflect significant deviation from ideal pentagonal bipyramidal geometry.

The metallocyclic intraligand $O-Cu-O$ ring angle opens with increasing size of the ring: 51° (four-membered) $< 72^\circ$ (five-membered). For five-membered metallocyclic rings, $N-Cu-N$ angles are 69° (for unsaturated ligand) and 80° (for saturated). One would expect that at the greater size of the chelate ring, the $M-L$ bond lengths would be somewhat shorter. The data presented in Table II, show this trend. For example, the mean $Cu-O$ bond distance of 2.37 \AA for four-membered chelate rings is about 0.11 \AA longer than the mean $Cu-O$ bond distance (2.26 \AA) for a five-membered chelate ring. The mean values of $L-Cu-L$ bond angles of the metallocyclic rings found for seven-coordinate $Cu(II)$ derivatives are smaller than those of six-, five- and four-coordinated (Table IB).

It is well known, that coordination number seven is unusual for $Cu(II)$, but there is an example, $[Cu(C_{10}H_{20}O_5)(H_2O)](NO_3)_2$,⁴⁸⁷ which contains ten crystallographically independent molecules differing by degree of distortion in the same crystal, a rarity in the complex chemistry of $Cu(II)$.

There are two examples, which contain two non-equivalent forms, the $[Cu(15\text{-crown-5})(MeCN)_2][Cu_3Cl_8]^{490}$ a monomeric cation with the CuO_5N_2 chromophore and a trimeric complex $[Cu_3Cl_8]^{2-}$ anion in which each

Cu(II) is a tetrahedral coordinated by four chlorine atoms and [Cu(15-crown-5)(MeCN)Cl]·[CuCl₃]^{490,491} consists of complex cation, where Cu(II) is surrounded by seven donor atoms (CuO₅NCl) and a dimeric complex [CuCl₃]⁻ anion, with a *pseudo*-tetrahedrally coordinated Cu(II) atom. Hepta-coordinate Cu(II) compounds are mostly red (orange) and by far “prefer” a monoclinic crystal class (Table II).

OCTACOORDINATE Cu(II) COMPOUNDS

Coordination number eight is rare for transition elements and especially for Cu(II). There are four examples^{496–499} for which crystallographic and structural data are given in Table II. There are two principal ways in which a cube may become distorted, square antiprism (symmetry D_{4d}) and dodecahedron (symmetry D_{2h}). The second distortion (all four examples belong to this class) can be best understood by recognizing that the cube is composed of two interpenetrating tetrahedra. The distortion occurs when the vertices of one of these tetrahedra are displaced so as to decrease the two vertical angles, that is, to elongate the tetrahedron, while the vertices of the other one are displaced to produce a flattened tetrahedron. It is important to note that its vertices (symmetry D_{2h}), are not all equivalent but are divided into two bisphenoidal sets, those within each set being equivalent; Table II shows this for Cu(II) examples. All examples^{496–498} contain bidentate ligands in which the two coordinated atoms are very close together. In these, the close pairs of ligand atoms lie on the *m* edges of the dodecahedron; these edges are the very short. The mean values of the two sets of Cu–O bond distances are 1.96 and 2.75 Å, respectively.

CONCLUSIONS

This review has classified almost six hundred six-, seven- and eight-coordinate Cu(II) coordination compounds which have been characterized by X-ray (neutron) diffraction. There are five hundred and fifty examples which are six-coordinate, and from the three principal forms of distortion of the octahedron, all occur in the chemistry of Cu(II) in varying degrees. The tetragonal distortion most commonly involves an elongation of one C_4 axis. In the series of heptacoordinate Cu(II) compounds only a pentagonal bipyramidal geometry around Cu(II) is observed. For Cu(II) octacoordinate complexes only a dodecahedral configuration is found.

The ligands in these complexes are homo- and hetero-donors. The former are uni-, bi-, tri-, tetra-, penta-, hexa- and even heptadentate; the latter are bi-, tri-, tetra-, penta- and hexadentate. The O- and N-donor ligands are most prevalent along with chloride. In general, the mean Cu–L_{eq} bond distance is shorter than that for Cu–L_{ax} in the series of six-coordinate complexes, but for seven-coordinate complexes and for some six-coordinate complexes the opposite is observed. In both series, there are some exceptions, which were discussed in the respective sections, and will not be repeated here.

We compared six-coordinate with five-coordinate Cu(II) compounds² and found that the mean Cu–L_{eq} bond distance increases as follows with configuration around Cu(II): square-pyramidal < tetragonal-bipyramidal < trigonal-bipyramidal; and Cu–L_{ax} bond distance: trigonal-bipyramidal < square-pyramidal < tetragonal-bipyramidal. In general, distortion around Cu(II) increases in the order: trigonal-bipyramidal < pentagonal-bipyramidal; and square-pyramidal < tetragonal-bipyramidal.

The effect of both electronic and steric hindrance of the ligands can be seen in the opening of the L–Cu–L angles of the respective metallocyclic rings. There is an impressive correlation between the Cu–L bond length and bite angle, at greater Cu–L bond lengths the angle is smaller and vice versa. The four- and five-membered rings are relatively rigid; however six- and higher-membered rings show more flexibility. In general, the mean values of the L–Cu–L bond angles of the respective metallocyclic rings, open with a decrease of coordination number about Cu(II) and with an increase of covalent radius of the donor atom/ligand.

There are several examples^{20,21,51,52,90,156,216,217,224,225,227,228,300,317} which exist in two isomeric forms, differing mostly by degree of distortion. Two crystallographically independent molecules differing by degree of distortion and coexisting in the same crystal have been found in several species.^{15,16,43,50,67,72,112,141,187,201,208,216,220b,236,240,280,359,376,391,408,445,485} In [Cu(en)₃]Cl₂·0.75en⁸⁶ three such molecules are present and in [Cu(C₁₀H₂₀O₅)(H₂O)₂](NO₃)₂⁴⁸⁷ ten exist. The coexistence of two or even more species differing only by degree of distortion within the same crystal is typical of distortion isomerism.⁵⁰⁰

The influence of temperature on degree of distortion of structures has been reported for some compounds.^{5,12,71,85,94,109} There are examples^{18,19,30,111,137,150,422} which contain two non-equivalent monomeric species. In [Cu(15-crown-5)(MeCN)₂][Cu₃Cl₈]⁴⁹⁰ the monomeric complex cation contains about the Cu(II) a pentagonal bipyramid (CuO₅N₂) and the trimeric complex anion has a tetrahedral (CuCl₄) arrangement. In

[Cu(15-crown-5)(MeCN)Cl][Cu₂Cl₆]^{490,491} besides the monomeric complex cation (CuO₅N₂), there is also the dimeric complex anion (CuCl₄).

Cu(II) compounds are mostly blue or green; there are also examples of red and brown. The majority have a monoclinic crystal class, but triclinic and orthorhombic are also observed.

Relationships between the various structural parameters have been discussed within each section.

This review, together with its precursor for four- and five-coordinate mononuclear Cu(II) compounds,² represents the first overview of structural data for over one thousand eight hundred mononuclear Cu(II) derivatives. In collecting and organizing the data it has become clear that despite the increasing availability of data retrieval systems, the tracing of relevant material is not always a straightforward task. One of the problems appears to be associated with the choice of keywords for indices, resulting in the effective invisibility of some material from a particular point of view. Some of the data is only available as supplementary material, and this can lead to overlooking of relevant structural features for comparative purposes. In view of the limitations inherent in information retrieval, we believe that such reviews will continue to serve a useful function by crystallizing available material, and delineating areas of both interest and weakness. Related reviews on the dimeric, as well as oligomeric and polymeric derivatives of Cu(II) are currently in progress.

Acknowledgement

The authors wish to thank the Ministry of Education of the Slovak Republic for financial support.

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